

**U.S. Department of Energy**  
**DUF<sub>6</sub> MATERIALS USE ROADMAP**

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## ACRONYMS

AHF	anhydrous hydrogen fluoride
ANSI	American National Standards Institute, Inc.
AVLIS	atomic vapor laser isotope separation
CFR	Code of Federal Regulations
DoD	Department of Defense
DOE	Department of Energy
DOE-EM	DOE Office of Environmental Management
DOE-MD	DOE Office of Fissile Materials Disposition
DOE-NE	DOE Office of Nuclear Energy, Science, and Technology
DOE-NNSA	DOE National Nuclear Security Agency
DOE-RW	DOE Office of Civilian Radioactive Waste Management
DOT	Department of Transportation
DU	depleted uranium
EIS	environmental impact statement
FEMP	Fernald Environmental Management Project
HEU(F)	highly enriched uranium (fuel)
HF	hydrogen fluoride (hydrofluoric acid)
HLW	high-level waste
INEEL	Idaho National Engineering and Environmental Laboratory
LEU	low-enriched uranium
LLW	low-level waste
LWR	light-water reactor
MOU	memorandum of understanding
MOX	mixed oxide
MT	metric tons
MTU	metric tons of elemental uranium
NEPA	National Environmental Policy Act of 1969
NRC	U.S. Nuclear Regulatory Commission
RCRA	Resource Conservation and Recovery Act of 1976

**ACRONYMS**  
(continued)

RFP	request for proposal
SNF	spent nuclear fuel
SRS	Savannah River Site
USEC	United States Enrichment Corporation
WAC	waste acceptance criteria
WP	waste package

## EXECUTIVE SUMMARY

The U.S. government has - 500,000 metric tons (MT) of surplus depleted uranium (DU) in various chemical forms stored at U.S. Department of Energy (DOE) sites across the United States. This DU, most of which is DU hexafluoride ( $\text{DUF}_6$ ) resulting from uranium enrichment operations, is the largest amount of nuclear material in DOE's inventory. On July 6, 1999, DOE issued the *Final Plan for the Conversion of Depleted Uranium Hexafluoride as required by Public Law 105-204*, in which DOE committed to develop a *Depleted Uranium Hexafluoride Materials Use Roadmap* in order to establish a strategy for the products resulting from conversion of  $\text{DUF}_6$  to a stable form. This report meets the commitment in the Final Plan by providing a comprehensive roadmap that DOE will use to guide any future research and development activities for the materials associated with its  $\text{DUF}_6$  inventory. The Roadmap supports the decision presented in the *Record of Decision for Long-Term Management and Use of Depleted Uranium Hexafluoride*, namely to begin conversion of the  $\text{DUF}_6$  inventory as soon as possible, either to uranium oxide, uranium metal, or a combination of both, while allowing for future uses of as much of this inventory as possible. In particular, the Roadmap is intended to explore potential uses for the  $\text{DUF}_6$  conversion products and to identify areas where further development work is needed. It focuses on potential governmental uses of  $\text{DUF}_6$  conversion products but also incorporates limited analysis of using the products in the private sector. The Roadmap builds on the analyses summarized in the recent *Programmatic Environmental Impact Statement for Alternative Strategies for the Long-Term Management and Use of Depleted Uranium Hexafluoride*. It also addresses other surplus DU, primarily in the form of DU trioxide and DU tetrafluoride.

The DU-related inventory considered here includes the following:

- Components directly associated with the  $\text{DUF}_6$  presently being stored at gaseous diffusion plant sites in Paducah, Kentucky; Portsmouth, Ohio; and Oak Ridge, Tennessee
  - 470,500 MT of DU
  - 225,000 MT of fluorine chemically combined with the DU
  - 74,000 MT of carbon steel comprising the storage cylinders<sup>1</sup>
- Approximately 27,860 MT of DU in the form of uranium trioxide, tetrafluoride, and various other forms containing varying amounts of radioactive and chemical impurities, presently stored primarily at DOE's Savannah River Site.

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<sup>1</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.

This Roadmap characterizes and analyzes alternative paths for eventual disposition<sup>2</sup> of these materials, identifies the barriers that exist to implementing the paths, and makes recommendations concerning the activities that should be undertaken to overcome the barriers. The disposition paths considered in this roadmap and shown in Fig. ES.1 are (a) implementation of cost-effective and institutionally feasible beneficial uses of DU using the products of DUF<sub>6</sub> conversion and other forms of DU in DOE's inventory, (b) processing the fluorine product resulting from DUF<sub>6</sub> conversion to yield an optimal mix of valuable fluorine compounds [e.g., hydrogen fluoride (hydrofluoric acid), boron trifluoride] for industrial use, and (c) processing emptied cylinders to yield intact cylinders that are suitable for reuse, while maintaining an assured and cost-effective direct disposal path for all of the DU-related materials.<sup>3</sup> Most paths consider the potential beneficial use of the DU and other DUF<sub>6</sub> conversion products for the purpose of achieving overall benefits, including cost savings to the federal government, compared with simply disposing of the materials. However, the paths provide for assured direct disposal of these products if cost-effective and institutionally feasible beneficial uses are not found.

Many of the paths included in this Roadmap face technical or institutional barriers, and significant uncertainties surround projections of cost-reduction benefits and operational improvements. DOE's approach to DU disposition activities should be focused into five areas:

*First*, DOE will support a broad spectrum of investments to reduce the barriers to paths related to nuclear material storage and/or disposal that have relatively low technical risk and use large quantities of DU in regulated areas.

*Second*, DOE will support targeted investments to reduce barriers for a number of paths where there is potential to use substantial amounts of DUF<sub>6</sub> conversion products or other forms of DU, but where the uses are more speculative or simply require a small investment before the path could be followed.

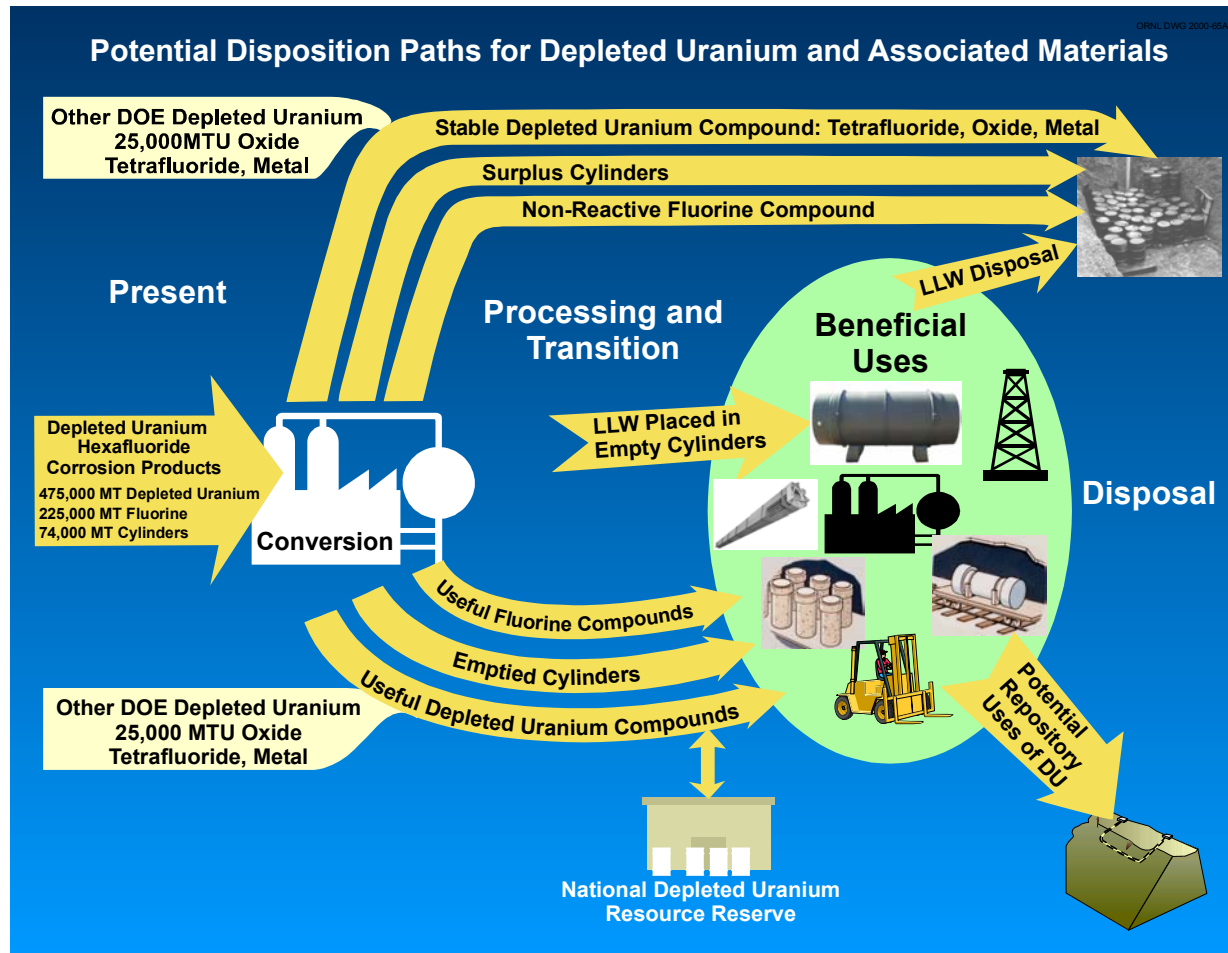
*Third*, DOE will make appropriate investments to ensure that there are no barriers to following an optimal path for long-term storage or direct disposal of the DU conversion products that are not beneficially used, or to disposal of DU-bearing products at the end of their useful lives.

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<sup>2</sup>Disposition constitutes some combination of long-term storage, beneficial use, and eventual disposal.

<sup>3</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.





**Fig. ES.1. Paths for disposition of surplus DU.** Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.

*Fourth*, DOE will invest in basic and mission-directed research that is related to beneficial use of DUF<sub>6</sub> conversion products. These investigations are necessary to provide a basis for evaluating the feasibility, impacts, and economics of potential DU disposition paths and to identify new beneficial uses of the DU conversion products, and other forms of DU.

*Fifth*, DOE will invest in system analysis and support activities that cross-cut multiple DUF<sub>6</sub> conversion products and other forms of DU. These activities include establishing roles and responsibilities for disposition of these products, efforts to foster acceptance of useful DU-bearing products and materials, and system baseline and optimization.

## 1. INTRODUCTION

The U.S. government has - 500,000 metric tons (MT) of surplus depleted uranium (DU) in various chemical forms stored at Department of Energy (DOE) sites across the United States. This DU, most of which is DU hexafluoride ( $\text{DUF}_6$ ) resulting from uranium enrichment operations, is the largest amount of nuclear material in DOE's inventory. On July 6, 1999, DOE issued the *Final Plan for the Conversion of Depleted Uranium Hexafluoride as required by Public Law 105-204* [DOE 1999b], in which DOE committed to develop a *Depleted Uranium Hexafluoride Materials Use Roadmap* in order to establish a strategy for the products resulting from conversion of  $\text{DUF}_6$  to a stable form. This report meets the commitment in the Final Plan by providing a comprehensive roadmap that DOE will use to guide any future research and development (R&D) activities for the materials associated with its  $\text{DUF}_6$  inventory. This Roadmap supports the decision presented in the *Record of Decision for Long-Term Management and Use of Depleted Uranium Hexafluoride* [FR 1999], namely to begin conversion of the  $\text{DUF}_6$  inventory as soon as possible, either to uranium oxide, uranium metal, or a combination of both, while allowing for future uses of as much of this inventory as possible. In particular, this roadmap is intended to explore potential uses for the  $\text{DUF}_6$  conversion products and to identify areas where further development work is needed. The roadmap focuses on potential governmental uses of  $\text{DUF}_6$  conversion products but also incorporates limited analysis of using the products in the private sector.<sup>1</sup> This Roadmap builds on the analyses summarized in the recent *Programmatic Environmental Impact Statement for Alternative Strategies for the Long-Term Management and Use of Depleted Uranium Hexafluoride* [DOE 1999a] and documented in engineering [Dubrin 1997] and cost [Elayat 1997] analysis reports. This Roadmap also addresses other surplus DU, primarily in the form of DU trioxide and DU tetrafluoride.

### 1.1 ROADMAPPING PROCESS

The process of defining and characterizing DOE's surplus DU inventory, specifying the alternative paths that could result in disposition of the DU, evaluating the paths, and recommending the preferred paths and activities required to overcome barriers along the paths is called *roadmapping*. The process steps involved in roadmapping DU and documented in this report are summarized as follows:

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<sup>1</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.

- Define the DU materials to be considered: The scope of the roadmap in terms of the range of DU materials considered is discussed in Sect. 1.2. Within this scope, the general disposition paths for DOE's surplus DU and the materials associated with it are described in Sect. 1.3.
- Characterize the present state of the DU inventory: Section 2 summarizes the characteristics of the relevant DU materials, including the following:
  - The inventory of DU, fluorine, and cylinders
  - Important characteristics of the inventory such as chemical form, contaminant concentrations, and enrichments
  - The regulatory context of the inventory and potential use of DU
- Specify and analyze alternative disposition paths: Section 3 summarizes the paths that are considered and characterizes each of them with respect to the following factors:
  - Existence of barriers to be overcome along the path
  - Amount of the inventory that could be used
  - Technical maturity and barriers
  - Institutional (including regulatory, legal, and policy) status and barriers
  - Economic and market aspects
  - Other impacts (e.g., changes in risk or environmental effects)
- Evaluate and categorize the paths: Section 4 summarizes the approach to and results of evaluating the DU disposition paths against criteria based on the characterization factors listed immediately above. The paths were evaluated by the following process:
  - Convening a workshop to review the information available concerning the paths
  - Having a group of knowledgeable scientists, engineers, and DOE staff members individually rank the paths
  - Summarizing and discussing the rankings to the point of consensus
  - Assigning each path to one of four categories:
    - @ Further barrier-reduction activities recommended
    - @ Further barrier-reduction activities should be considered
    - @ Further barrier-reduction activities should not be considered
    - @ No additional federal barrier-reduction activities required to support DU disposition
- Specify DU disposition barrier-reduction activities for each path: Section 5 summarizes the barrier-reduction activities for each path in the following categories:
  - Recommended path-specific activities
  - Activities that should be considered
  - Recommended cross-cutting and systems activities
  - Recommended topics for DOE research programs

- Define DOE's path forward regarding DU disposition: Using the results of Sect. 5 as a basis, Sect. 6 summarizes a five-point plan that constitutes DOE's preferred approach for overcoming the barriers to DU disposition. It is this plan that DOE intends to follow, subject to budget limitations.

## 1.2 ROADMAP SCOPE

The materials considered in this Roadmap are all DOE surplus DU or related materials, including fluorine and emptied storage cylinders associated with  $\text{DUF}_6$ . Consideration of establishing a national resource reserve is also included, but DU required for ongoing programs such as those in DOE's National Nuclear Security Agency (DOE-NNSA) or Department of Defense (DoD) is excluded.

The range of alternative DU disposition paths considered is as follows:

- Begin with these materials:
  - Products from conversion of  $\text{DUF}_6$
  - Non- $\text{DUF}_6$  forms of surplus DU in DOE's inventory
- Explore beneficial uses of DU, fluorine by-products, and emptied cylinders<sup>2</sup> while maintaining the option to directly dispose of some or all of these conversion products. The beneficial uses considered will emphasize potential federal applications, with some consideration being given to potential nonfederal applications.
- End with disposal of the DU-bearing products as an integral part of use or at the end of their useful life.

## 1.3 GENERAL DU DISPOSITION PATHS

The general disposition paths that DOE envisions for DU and DU-related materials are:

(a) implementation of cost-effective and institutionally feasible beneficial uses of DU using the products of  $\text{DUF}_6$  conversion and other forms of DU in DOE's inventory, (b) processing the fluorine product resulting from  $\text{DUF}_6$  conversion to yield an optimal mix of valuable fluorine compounds [e.g., hydrogen fluoride (hydrofluoric acid) (HF), boron trifluoride] for industrial use, and (c) processing emptied cylinders to yield intact cylinders that are suitable for reuse,<sup>2</sup> while maintaining an assured and cost-effective direct disposal path for all of the DU and DU-related materials.

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<sup>2</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.

## **2. CURRENT INVENTORY AND CHARACTERISTICS OF DU AND ASSOCIATED FLUORINE AND CYLINDERS**

This section provides foundational information for the remainder of the Roadmap. First, Sect. 2.1 defines the inventory and characteristics of DU, fluorine, and cylinders that are within the scope of this effort, including the form and location of the inventory. Sections 2.2–2.4 provide general background information on regulations relevant to the disposition of this inventory.

### **2.1 INVENTORY OF DU AND ASSOCIATED MATERIALS**

#### **2.1.1 DU**

Depleted uranium—uranium with  $^{235}\text{U}$  content less than the naturally occurring concentration of 0.711 wt %—has been generated in the United States as a by-product (called tails) from uranium enrichment operations. This study addresses the technology associated with management and disposition of DU. It focuses on the  $\text{DUF}_6$  that constitutes the majority of the inventory but also includes the relatively small quantities of DU that resulted from purification of products from fuel reprocessing operations at both the Hanford Site and the Savannah River Site (SRS).

The estimated U.S. inventory of purified DU that has been generated as tails from uranium enrichment and spent fuel reprocessing totals - 500,000 MTU of elemental uranium (MTU) (see Table 2.1). About 470,000 MTU of DU, managed by DOE's Office of Environmental Management (DOE-EM), is stored as  $\text{UF}_6$  in cylinders at three sites—Paducah, Kentucky; Portsmouth, Ohio; and Oak Ridge, Tennessee. Also, - 25,360 MTU exists at various sites in a number of chemical forms, including oxides, tetrafluoride, metal, alloys, and process residues. Another - 2,500 MTU that is managed by DOE-NNSA, DoD, and U.S Nuclear Regulatory Commission (NRC) licensees is excluded from consideration because it is intended for use by these organizations.

##### **2.1.1.1 Uranium Hexafluoride**

The U.S. uranium enrichment plants have been operated since the mid-1940s, resulting in the continuous production of  $\text{DUF}_6$ . However, they have been operated under varying policy and economic conditions and with different feed materials. As a result, the stored  $\text{DUF}_6$  has a range of characteristics important to DU disposition. In particular, the following characteristics are noted:

**Table 2.1. Estimated U.S. inventory of DU**

Form	Owner	Location	MTU		
			Inventory by site	Subtotal	Total
UF <sub>6</sub>	DOE	Paducah	230,000	375,000	
		Portsmouth	108,000		
		Oak Ridge	37,000		
	USEC	Paducah	73,300	95,500	470,500
	Portsmouth	22,200			
UO <sub>3</sub>	DOE	FEMP <sup>a</sup>	34	- 19,500	- 19,500
		SRS	19,440		
		Other	- 1		
U metal	DOE	FEMP <sup>a</sup>	1,860	- 3,000	- 3,000
		RFETS	22		
		Other	>1,000		
UF <sub>4</sub> , other <sup>b</sup>	DOE	FEMP <sup>a</sup>	1,120	- 3,000	- 2,860
		Paducah	1,140		
		SRS	- 500		
		Other	>100		
Various <sup>c</sup>	Various	Various		<2,500	<2,500
					- 498,360
<b>Total (round-off)</b>					<b>- 500,000</b>

<sup>a</sup>This material is in the process of being transferred to Portsmouth.

<sup>b</sup>These forms, at various DOE sites, include UF<sub>4</sub>, oxides other than UO<sub>3</sub>, process residues, and solutions.

<sup>c</sup>These components, which are estimated to account for <0.5% of the total U.S. inventory, are those of (1) DOE-NNSA, (2) DoD, (3) licensed commercial users, and (4) returns expected from foreign licensees.

- The  $^{235}\text{U}$  content of the  $\text{DUF}_6$  (i.e., the extent to which the uranium was depleted in  $^{235}\text{U}$  during enrichment) varies from <0.2% to nearly the levels in natural uranium (0.711%). However, the average  $^{235}\text{U}$  content is about 0.27%, and 91% of the tails have a  $^{235}\text{U}$  content of <0.4%.
- At times the uranium enrichment plants were fed with recycled uranium that contained trace amounts of radionuclides such as  $^{237}\text{Np}$  and  $^{99}\text{Tc}$  that form volatile fluoride compounds. As a result, there are traces of these radionuclides in some  $\text{DUF}_6$  cylinders [Hightower 2000].
- The  $\text{DUF}_6$  placed in the storage cylinders is a radioactive material. As a result, radioactive decay products have built up in the storage cylinders over the years. The consequence is that the DU decay products in the emptied cylinders will be about 20 times more radioactive than the DU removed from the cylinders, and the radioactivity of products made from DU will also increase by - 2.5 within - 20 years after manufacture. The dose rate 30 cm from a large unshielded mass of DU is 1.5–2.0 mrem/h.

#### **2.1.1.2 Uranium Trioxide ( $\text{UO}_3$ )**

DOE-EM currently manages about 23,400 MT  $\text{UO}_3$  containing about 19,500 MTU as DU trioxide ( $\text{DUO}_3$ ), resulting from historical weapons production programs at the U.S. defense complexes. Depleted  $\text{UO}_3$ , recovered either through chemical separation procedures used in  $^{239}\text{Pu}$  production programs or as a by-product from target and weapons component fabrication, is now stored at the Fernald Environmental Management Project (FEMP) and SRS. The  $\text{UO}_3$  at FEMP is in process of being transferred to DOE-NE at Portsmouth.

#### **2.1.1.3 Uranium Metal**

Most of the - 3,000 MT DU metal covered in this Roadmap are located primarily at FEMP, is in the form of ingots or “derbies,” typically 12 in. diam. by 8 in. long, with a density of 19.05 kg/L. The derbies are the raw material for producing finished products, including coated metal and alloys. This material is in the process of being transferred to DOE-NE management at Portsmouth.

#### **2.1.1.4 Uranium Tetrafluoride ( $\text{UF}_4$ )**

The  $\text{UF}_4$  was produced, primarily at FEMP, as a step in the production of uranium metal. This material is in the process of being transferred to DOE-NE management at Portsmouth. Discussions are currently in progress to document possible DoD interest in this  $\text{UF}_4$ .

#### **2.1.1.5 Other Forms**

Other materials, including miscellaneous oxides, solutions, and process residues, may need to be converted and packaged for disposition.



### 2.1.2 Fluorine

Approximately 225,000 MT of elemental fluorine could be derived from the - 700,000 MT of  $\text{DUF}_6$  stored at DOE enrichment sites. This fluorine is potentially recoverable as elemental fluorine, HF (aqueous HF), anhydrous HF (AHF), or other fluorine-bearing compounds—all of which could be recycled to conserve natural resources and partially defray costs associated with the conversion of  $\text{DUF}_6$  to forms that are more acceptable for storage. The HF products could be used in many commercial industrial activities, particularly in the nuclear industry to fluorinate natural uranium.

### 2.1.3 $\text{UF}_6$ Cylinders

The number and types of  $\text{DUF}_6$  cylinders at the three enrichment sites were determined by the Bechtel Jacobs Company in May 1999 [Manuel 1999], and detailed properties of the various models of  $\text{DUF}_6$  cylinders were obtained from the U.S. Enrichment Corporation (USEC) [USEC 1999]. Greater than 99.92% of the total cylinder mass of - 74,000 MT is composed of formed and welded carbon steel plates per American Society for Testing and Materials A516 or A285 (DNFSB 1995). Cylinders considered acceptable for  $\text{UF}_6$  handling and shipment must be inspected, tested, and maintained within the intent of the standard American National Standards Institute, Inc. (ANSI) N14.1. Old or corroded cylinders not meeting the ANSI standard require special handling—features such as special overpacks, transfer of contents to approved cylinders, and approval for exception by regulatory agencies such as the U.S. Department of Transportation (DOT).

In addition to the steel cylinders, there are nickel and Monel cylinders. A total of 49 MT of nickel and 9 MT of Monel comprising 779 small cylinders will become available as a result of  $\text{DUF}_6$  disposition.

Cylinders of uranium enrichment tailings normally contain  $\text{DUF}_6$  with a purity exceeding 99.9%. Aged cylinders have small quantities of other uranium fluorides (e.g.,  $\text{DUF}_5$ ), fluorides of uranium decay products, and fluorides of container-corrosion products. Breached empty cylinders may also contain hydrous oxides, including uranium oxyfluoride generated by reaction of the  $\text{DUF}_6$  with moist air. Tailings may also contain small quantities of the radioisotope  $^{237}\text{Np}$ , a residual from the use of “reactor recycle” cascade feed. All of these impurities generally remain as a solid residue, or “heel,” in the cylinder when the  $\text{DUF}_6$  is removed by liquefaction or vaporization [Michelhaugh 1999]. Cylinders previously containing  $\text{DUF}_6$  generally exceed radiation and chemical hazard standards for unrestricted use unless they have been decontaminated and inspected. Most management strategies will require that emptied cylinders be cleaned to remove the solid residues.

## **2.2 CURRENT REGULATORY STATUS OF DU AND ASSOCIATED MATERIALS**

All paths to disposition of DU discussed in this report are composed of one or more of the following activities involving DU: storage; transfer; processing (e.g., conversion); product use; and disposal. The regulations applicable to an activity will depend largely on whether, within a particular path, the activity is controlled by DOE or is regulated instead by the NRC or an NRC Agreement State. This section summarizes the current regulatory status of DU, and for each of the five activities, summarizes existing regulatory requirements and issues relevant to the roadmapping effort.

### **2.2.1 Regulatory Status of DU**

Under the Atomic Energy Act of 1954 (PL 1954), as amended, the purified chemical forms of DU contained in DOE's inventory are classified as "source material." As such, DOE's DU is excluded from the definitions of solid and hazardous waste in the Resource Conservation and Recovery Act (RCRA) [RCRA 1976]. This means that RCRA should not apply to DU activities within any path, unless the purified DU becomes mixed with some other material to which hazardous waste provisions of RCRA do apply.

Under the definitions of radioactive wastes contained in the Nuclear Waste Policy Act of 1982, the Low-Level Radioactive Waste Policy Amendments Act of 1985, and the Waste Isolation Pilot Project Land Withdrawal Act of 1992, DU is low-level waste (LLW) when disposed of. In general, materials contaminated with significant residual DU (e.g., empty  $\text{DUF}_6$  cylinders and discarded  $\text{CaF}_2$  from the neutralization of gas waste streams) are also low-level radioactive waste when disposed of.

### **2.2.2 Requirements and Issues Related to Storage**

DOE has entered into consent orders with the responsible regulatory agencies in Ohio and Tennessee regarding ongoing storage of  $\text{DUF}_6$  which should not affect future regulation of DU storage. DOE will conduct appropriate site-specific reviews, as required by the National Environmental Policy Act (NEPA) of 1969, for  $\text{DUF}_6$  conversion product storage at any new or existing DOE-controlled site or non-DOE facility. If commercial storage facilities are used, DOE must ensure that such facilities are authorized by valid NRC or NRC Agreement State licenses (DOE M 474.1-2).

## **2.2.3 Requirements and Issues Related to Transfers**

### **2.2.3.1 DU**

A DOE-controlled  $\text{DUF}_6$  facility is generally allowed to transfer DU, regardless of form, only to authorized DOE-controlled facilities or non-DOE facilities with appropriate NRC or NRC Agreement State licenses (DOE M 474.1). In some cases, the impurities in the DU may not be acceptable under existing licenses for some non-DOE facilities, and changes would be required before the DU could be accepted.

### **2.2.3.2 Residual Radioactive Material in Fluorine Products**

DOE has established no generic release limits for the transfer from DOE control of fluorine products containing volumetrically distributed residual radioactivity. Therefore, case-specific release (authorization) limits developed and approved by DOE according to the process explained in DOE Order 5400.5 would be developed before releasing such material from a DOE-controlled DU conversion facility. Among other things, these release limits must ensure against releasing a licensable quantity of uranium to any person not licensed to receive it. Uranium becomes licensable by the NRC in a mixture or solution (such as a fluorine by-product) at a concentration  $\geq 0.05$  wt % (500 ppm by weight) [10 CFR 40.13(a)]. It is anticipated that the uranium concentration in fluorine products will be much less than 500 ppm.

### **2.2.3.3 Residual Radioactive Material in Empty $\text{DUF}_6$ Cylinders**

If a DOE-controlled  $\text{DUF}_6$  conversion facility releases empty  $\text{DUF}_6$  cylinders to a non-DOE metal-recycling facility, the recycling facility must hold an NRC or Agreement State license. In January 2000, in response to concerns about the release of volumetrically contaminated nickel from the East Tennessee Technology Park, the Secretary of Energy established a moratorium prohibiting the release of all volumetrically contaminated metals from DOE facilities to give the NRC time to develop national standards for such materials, allow the public to weigh in on the development of a national policy, and permit DOE to establish its policy, directives, and guidance in this regard. In addition, on July 13, 2000, DOE suspended the unrestricted release for recycling of scrap metals from radiation areas within DOE facilities. This suspension remains in effect until DOE implements improved release criteria and information management requirements relating to these materials.

## **2.2.4 Requirements and Issues Related to Processing**

Commercial facilities that fabricate DU products must hold an NRC or NRC Agreement State specific source material license issued under 10 CFR Part 40, “Domestic Licensing of Source Material.” If such a facility fabricates and distributes products or devices that will be subject to the general source material license (see Sect. 2.2.5.2), the facility’s specific license application must demonstrate with reasonable assurance that possession, use, or transfer of the product or device is not likely to cause any individual user to receive a radiation dose in excess of 10% of the annual limits delineated in 10 CFR 20.1201(a), “Occupational Dose Limits for Adults.”

## **2.2.5 Requirements and Issues Related to DU Product Use**

Use of DU products must either be exempt from the requirement to obtain an NRC/Agreement State license or be covered by a general or specific source material license.

### **2.2.5.1 Exempt Uses**

Use of DU products by DOE is exempt from the requirement to obtain an NRC/Agreement State source material license (10 CFR 40.11). Also exempt is use of the following DU products by persons who comply with specified conditions:

- Counterweights installed in aircraft, rockets, projectiles, and missiles [10 CFR 40.13(c)(5)]
- Metal shielding components of any shipping container [10 CFR 40.13(c)(6)]
- Detector heads in fire detection devices [10 CFR 40.13(d)]

### **2.2.5.2 General Source Material License**

Existing NRC and NRC Agreement State regulations grant a general source material license. Under the general license, DU can be used in industrial products or devices for the purpose of providing a concentrated mass in a small volume, as long as the products or devices are manufactured or initially transferred in accordance with a specific source material license and certain other conditions are met (10 CFR 40.25).

### **2.2.5.3 Specific Source Material License**

A user of a DU product must apply to the NRC or an NRC Agreement State for a specific source material license, unless the product is either exempt or covered by the existing general license described in 10 CFR 40.25.

## 2.2.6 Requirements and Issues Related to Disposal

In the February 25, 2000, *Record of Decision for the Department of Energy's Waste Management Program: Treatment of Low-Level Waste and Mixed Low-Level Waste; Amendment of the Record of Decision for the Nevada Test Site*, DOE decided to perform minimum treatment of LLW at all sites and continue, to the extent practical, disposal of on-site LLW at Idaho National Engineering and Environmental Laboratory (INEEL), Los Alamos National Laboratory, Oak Ridge Reservation, and SRS. In addition, DOE decided to make the Hanford Site and the Nevada Test Site (NTS) available to all DOE sites for LLW disposal [FR 2000].

DOE Order O 435.1, "Radioactive Waste Management," and its implementing manual, DOE M 435.1-1, govern disposal of DU and materials containing residual DU (e.g., empty DUF<sub>6</sub> cylinders and calcium fluoride from the neutralization of gas waste streams) in DOE-controlled LLW disposal facilities. The manual explains that DOE-controlled LLW disposal facilities must have a radioactive waste management basis consisting of a performance assessment, composite analysis, disposal authorization statement, closure plan, waste acceptance requirements, and monitoring plan. The waste acceptance requirements contain minimum criteria that could preclude disposal of some chemical forms and some physical forms of DU without special packaging and/or stabilization.

DOE M 435.1-1 prohibits disposal of DOE-generated LLW, including DU or materials containing residual DU (e.g., calcium fluoride and empty DUF<sub>6</sub> cylinders), in non-DOE LLW disposal facilities unless the responsible DOE Field Element Manager approves an exemption for use of non-DOE facilities based, in part, on a determination that DOE-controlled disposal capabilities are not practical or cost-effective. If disposal in an NRC- or NRC Agreement State-licensed LLW disposal facility is approved, such facilities must be subject to 10 CFR Part 61, "Licensing Requirements for Land Disposal of Radioactive Wastes," or compatible state regulations. Title 10 CFR Part 61 requires a demonstration of compliance with specified performance objectives and technical standards. Title 10 CFR Part 61 also requires facilities to establish waste characteristic limitations that could preclude disposal of some chemical forms and some physical forms of DU without special packaging and/or stabilization.

In 1995, during the scoping process for the programmatic environmental impact statement concerning long-term management of DUF<sub>6</sub>, the NRC staff expressed its opinion that DU<sub>3</sub>O<sub>8</sub> is a likely chemical form for DU disposal. They also advised DOE that, although DU<sub>3</sub>O<sub>8</sub> could be disposed of in limited quantities in conventional near-surface disposal facilities, large quantities (such as would be derived from the nation's enrichment tailings inventory) suggest the possible need for a unique disposal facility, such as a mined cavity or an exhausted uranium mine [NRC, Letter from NRC (R. Bernero) to DOE (C. Bradley, Jr.), January 3, 1995].

## 2.3 INSTITUTIONAL INFLUENCES ON ROADMAP DEVELOPMENT

In addition to the regulatory requirements and issues discussed in Sect. 2.2, several institutional influences have affected the development of this Roadmap. These influences included two decision documents recently issued by DOE, two memoranda of understanding (MOUs) signed jointly by DOE and USEC, the availability of beneficial uses for materials resulting from the conversion of  $\text{DUF}_6$  to DU oxide or DU metal, and the acceptance by the public and industry of the products made from such materials. Each of these is briefly described below.

The “*Record of Decision for Long-Term Management and Use of Depleted Uranium Hexafluoride*” was issued in July 1999 [FR 1999]. The decision is, in part, to convert DOE’s  $\text{DUF}_6$  inventory to DU oxide, DU metal, or both. The *Final Plan for the Conversion of Depleted Uranium Hexafluoride*, submitted to Congress in July 1999, as required by Public Law 105-204 [DOE 1999b], presents a timetable for beginning the conversion process by the fourth quarter of 2004. Therefore, this Roadmap assumes that the conversion will take place as planned.

Two MOUs signed between DOE and USEC in May and June of 1998 [DOE/USEC 1998a, 1998b], have transferred or will transfer - 98,000 MT of DU in the form of  $\text{DUF}_6$  in - 11,200 cylinders from USEC to DOE. As discussed in Sect. 2, this brings the total number of  $\text{DUF}_6$  cylinders that fall under DOE’s responsibility for managing to - 57,700. Although it is conceivable that DOE’s inventory of  $\text{DUF}_6$  may change in the future due to transfers between DOE and USEC, for the purposes of this exercise, it is assumed that the inventory will stay constant.

It is recognized that important elements in development and realization of potential uses of DU in products are public acceptance and industry interest. One concern is the risk associated with radiation dose from DU in industrial products during normal usage and following postulated accidents. Another concern is financial liability of companies that manufacture and use these products. Therefore, some effort to estimate the risks from potential products and to communicate such risks to the public and industry could facilitate use of DU conversion products.

## 2.4 TRANSPORT REGULATIONS FOR $\text{DUF}_6$ -DERIVED MATERIALS

In accomplishing the  $\text{DUF}_6$  disposition mission, multiple types of materials will require transport in the public domain. In addition to transporting  $\text{DUF}_6$  from current storage locations to processing facilities, the materials arising from the processing of  $\text{DUF}_6$  [including the primary fluorine product (e.g., AHF, calcium fluoride), cut and crushed cylinders, and secondary wastes] may also require transport to user or disposal sites. In all cases, the packaging and transport of these materials will be governed by DOT and NRC regulations. The packaging and transport of the  $\text{DUF}_6$  may pose some unique problems as described in Sect. 2.3.

### 3. DISPOSITION PATHS FOR DU AND ASSOCIATED MATERIALS

This section describes the overall approach to DU disposition and a specific set of paths for such disposition as a basis for evaluation of the paths and identification of the preferred paths. The evaluation process and results are discussed in Sect. 4.

The options that are available for DU disposition are shown in Fig. 3.1. Disposition begins with material from two sources. The first material source is the  $\text{DUF}_6$  inventory containing - 470,000 MTU of DU; 225,000 MT of fluorine; 74,000 MT of steel; and 58 MT of nickel and Monel. The second material source is the 25,500 MTU of surplus DU in DOE's inventory that is not in the form of  $\text{DUF}_6$ .

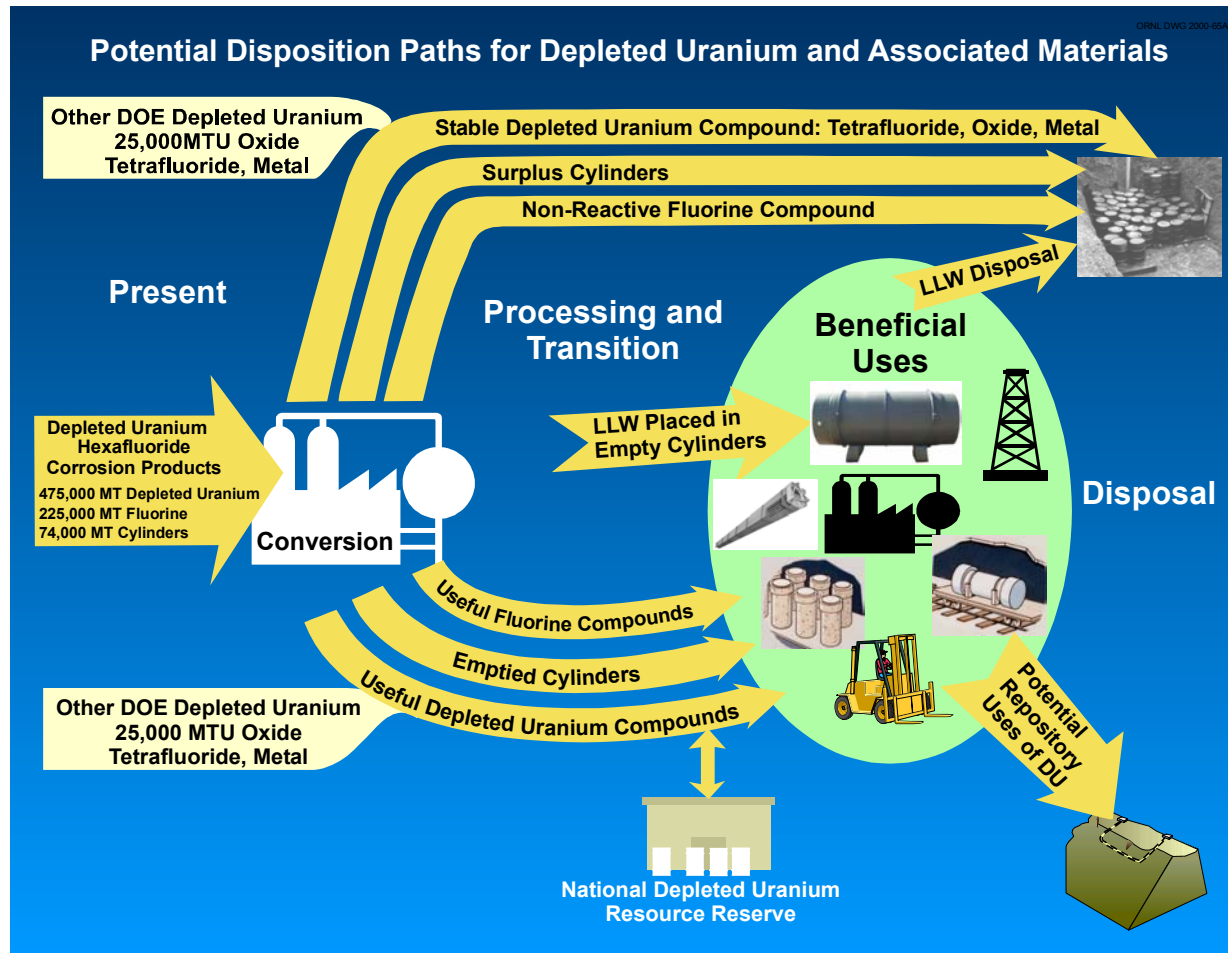
The first step in disposition is to convert the  $\text{DUF}_6$  to a stable form in a conversion plant. Anticipated products of this plant are:

- DU in the form of tetrafluoride, oxide, metal, or a combination of these
- Emptied  $\text{DUF}_6$  storage cylinders
- Fluorine in a form such as liquid HF, gaseous AHF, other fluorine compounds having a higher unit value (e.g.,  $\text{BF}_3$ ), or solid calcium difluoride

Other surplus DU is expected to require characterization and possibly treatment of some portions of the inventory to meet subsequent disposition requirements.

Disposition of  $\text{DUF}_6$  conversion products plus the other surplus DU then proceeds according to one or a combination of two scenarios: direct disposal or beneficial uses. Direct disposal can be accomplished by following paths such as near-surface disposal as LLW of various chemical forms of DU, near-surface disposal as LLW of cylinders after cutting them into segments, and near-surface disposal as LLW of a stable fluorine compound (e.g.,  $\text{CaF}_2$ ,  $\text{MgF}_2$ ) produced from the fluorine by-products of  $\text{DUF}_6$  conversion, which contains residual radioactivity. Direct disposal is the reference path for all except the fluorine-bearing product, where industrial use of HF or AHF is established practice.

The second scenario involves beneficial use of  $\text{DUF}_6$  conversion products plus other surplus DU to reduce the overall cost to the government for DU disposition or to achieve other worthwhile goals. This scenario can be accomplished by following paths such as use of appropriate chemical forms of DU in various products (e.g., spent fuel storage and shipping casks), reuse of cylinders or their components, or industrial use of fluorine-bearing products. Ultimately, most of the DU-bearing products that are beneficially used will require disposal. For most DU products, disposal will involve burial in near-



**Fig. 3.1. Paths for disposition of surplus DU.** Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.



surface facilities for LLW. Other DU products might be used in the potential spent fuel repository. It is possible that relatively small amounts of unique DU forms will be retained in long-term storage as a national resource reserve to meet unspecified future demand. Designation as a national resource material would generally be accorded to those materials that would be very difficult to replicate or where there are multiple users of a particular DU form to the point that a single custodian is not practical.

There are barriers to implementing many of the candidate paths, especially those involving beneficial use of  $\text{DUF}_6$  conversion products or other surplus DU. These barriers may be technical, economic, or institutional (i.e., policy, regulatory, legal). It is necessary to elucidate these barriers as a foundation for decisions on which path(s) should be followed and which should be abandoned. The first step in elucidating the barriers is to define candidate disposition paths for  $\text{DUF}_6$  conversion products and other surplus DU. Based on an analysis of the existing literature and input from a diverse group of experts, a list of candidate disposition paths was developed for  $\text{DUF}_6$ , forms of DU other than  $\text{DUF}_6$ , the fluorine from  $\text{DUF}_6$  conversion, and  $\text{DUF}_6$  storage cylinders. Paths that would implement the disposal scenario are described in Table 3.1, and paths that would implement the beneficial use scenario are described in Table 3.2. The reference disposition paths for  $\text{DUF}_6$  conversion products are conversion to a stable form and disposal of most of the DU, retention of relatively small amounts of DU in various chemical forms as a national resource reserve, disposal of steel cylinder segments, and industrial use of AHF or HF. The reference disposition paths are also DOE's preferred alternatives based on numerous studies to date and the technical, economic, and institutional factors that have been considered. Other paths constitute alternatives that might offer cost savings or other benefits to DOE. These lists of candidate paths are believed to comprehensively represent presently conceived uses of the materials associated with DU disposition irrespective of the development status, feasibility, or potential of each use. It is expected that most paths will not be pursued further because of one or more technical, economic, or institutional factors that are considered in the evaluation process described in Sect. 4. The remainder of this report will evaluate disposition pathways and recommend barrier reduction for each.

**Table 3.1. Description of candidate disposition paths for direct disposal of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate path	Candidate path description
<b>DUF<sub>6</sub></b>	
LLW disposal <u>Reference path for most DU</u>	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to a stable form such as tetrafluoride, oxide, or metal</li> <li>• Package DU form and include DU products at the end of their useful lives</li> <li>• Dispose of the DU packages and products via burial in an LLW disposal facility</li> </ul>
Mined cavity disposal	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to a stable form such as tetrafluoride, oxide, or metal</li> <li>• Package DU form, possibly with a matrix such as grout, and include DU products at the end of their useful lives</li> <li>• Dispose of the DU packages and products in a new or existing deep geologic facility</li> </ul>
Salt mine disposal as DUF <sub>6</sub>	<ul style="list-style-type: none"> <li>• Dispose of DUF<sub>6</sub> cylinders in a deep salt mine, possible with overpacks</li> </ul>
Subsurface engineered vault disposal	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to a stable form such as tetrafluoride, oxide, or metal</li> <li>• Package DU form, possibly with a matrix such as grout, and include DU products at the end of their useful lives</li> <li>• Dispose of the DU packages and products in subsurface concrete vaults</li> </ul>
<b>DU other than DUF<sub>6</sub></b>	
LLW disposal of other DU <u>Reference path</u>	<ul style="list-style-type: none"> <li>• Characterize, convert, and package the DU other than DUF<sub>6</sub> to the minimum extent possible to meet WAC</li> <li>• Dispose of the DU as LLW</li> </ul>
<b>Fluorine products</b>	
Disposal of fluorine	<ul style="list-style-type: none"> <li>• Convert fluorine to CaF<sub>2</sub> or MgF<sub>2</sub></li> <li>• Dispose of fluorine compounds</li> </ul>
<b>DUF<sub>6</sub> storage cylinders</b>	
Dispose of metals <u>Reference path</u>	<ul style="list-style-type: none"> <li>• Remove UF<sub>6</sub></li> <li>• Wash cylinder internally</li> <li>• Reduce volume by sectioning and flattening</li> <li>• Send metal to an LLW disposal facility</li> </ul>

**Table 3.2. Description of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate path	Candidate path description
<b>DUF<sub>6</sub></b>	
<b>DU matrix and shielding products</b>	
Cement-Lock™	<ul style="list-style-type: none"> <li>• Directly convert DUF<sub>6</sub> and other wastes to slag in a reactive melter</li> <li>• Quench molten material, grind with additives, and mix with cement to form high-density concrete</li> <li>• Form concrete into useful products</li> <li>• Eventually dispose of the products</li> </ul>
DU metal shielding	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to metal</li> <li>• Manufacture large metal shapes for use in shielding, primarily in spent fuel storage and transportation casks</li> <li>• Eventually dispose of the DU components</li> </ul>
DUCRETE™	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Convert oxide to DU aggregate (DUAGG™)</li> <li>• Mix DUAGG™ with cement to make high-density concrete</li> <li>• Form high-density concrete into useful products such as spent fuel storage silos</li> <li>• Eventually dispose of products; possibly use as LLW packages</li> </ul>
DUPoly	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Mix uranium oxide with molten polyethylene to form high-density polyethylene (DUPoly)</li> <li>• Form DUPoly into useful products such as shielding</li> <li>• Eventually dispose of products; possibly use as LLW packages</li> </ul>
PYRUC	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Process DU oxide into small, sintered particles</li> <li>• Coat particles with a layer of pyrolytic carbon</li> <li>• Mix coated particles with binder</li> <li>• Form bound particles into monoliths for useful products such as spent fuel storage silos</li> <li>• Eventually dispose of products; possibly use as LLW packages</li> </ul>
Uranium silicide	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> directly into USi<sub>x</sub> particles by reaction in molten silicon</li> <li>• Form particles into USi<sub>x</sub> aggregate</li> <li>• Mix aggregate with cement to make high-density concrete</li> <li>• Form concrete into useful products such as spent fuel storage silos</li> <li>• Eventually dispose of products; possibly use as LLW packages</li> </ul>
<b>Proposed applications in the potential repository</b>	
Backfill component	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Mix DU oxide with rock and other additives at the potential repository site</li> <li>• Use the mixture to fill drifts containing spent fuel and WPs at the time of closure</li> <li>• Backfill in emplacement drifts is not presently part of the reference design of the potential repository and is planned to be installed only in nonemplacement drifts. This may be reevaluated in the future, but in any case, backfill would not be installed any earlier than the 22nd century and maybe later.</li> </ul>
Invert	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Make the DU oxide into particles or noncementitious DU aggregate</li> <li>• Insert DU form into cells formed by steel plates used to level the bottom of cylindrical tunnels in the potential repository</li> <li>• Inverts containing DU are not part of the reference design of the potential repository</li> </ul>
Package fill	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to dioxide</li> <li>• Load spent fuel into WP</li> <li>• Insert DU dioxide particles in all gas spaces inside the WP but outside the fuel rods</li> <li>• Store and dispose of the packages in a repository</li> <li>• Package fill is not part of the reference potential repository design, nor is it an alternative design option</li> </ul>

**Table 3.2. Description of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate path	Candidate path description
<b>Fissile material disposition applications</b>	
Ceramics for Pu disposition	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to dioxide</li> <li>• Mix DU dioxide with Pu dioxide and form a ceramic by sintering</li> <li>• Dispose of the ceramic in small cans within a larger can of HLW</li> </ul>
Dilution of HEU	<ul style="list-style-type: none"> <li>• Mix DUF<sub>6</sub> with HEUF<sub>6</sub> to yield LEU</li> <li>• Make LWR fuel</li> <li>• Dispose of spent fuel in the potential repository</li> </ul>
MOX fuel for Pu disposition	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Mix DU oxide with Pu oxide and make LWR fuel</li> <li>• Use the fuel in a reactor</li> <li>• Dispose of spent fuel in the potential repository</li> </ul>
<b>Commercial applications<sup>3</sup></b>	
Aluminum-refining electrodes	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Mix DU oxide with other compounds and form electrodes</li> <li>• Use electrodes to refine aluminum from ores</li> <li>• Dispose of DU from electrode degradation with slag as industrial waste</li> <li>• Some DU is released as trace contamination in aluminum products</li> </ul>
Catalyst for fluid cracking and to promote oxidation	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Combine with other materials to manufacture catalyst</li> <li>• Use catalyst to refine petroleum and process other chemicals</li> <li>• Dispose of spent catalyst as LLW or release as trace contaminants in products</li> </ul>
Catalyst for automotive exhaust	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Combine with other materials to manufacture catalyst</li> <li>• Install catalyst in automobile catalytic converters</li> <li>• Recycle catalyst where possible; the excess will be disposed as LLW</li> </ul>
Catalyst for fuel cells and steam reforming	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to oxide</li> <li>• Combine with other materials to manufacture catalyst</li> <li>• Use catalyst to promote fuel cell reactions and steam decomposition to produce hydrogen</li> <li>• Dispose of spent catalyst as LLW</li> </ul>
Heavy-lifting-vehicle counterweight and high-traction devices	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to metal</li> <li>• Form DU metal into large shapes</li> <li>• Use shapes as counterweights located under heavy-lifting equipment or locomotive wheels</li> <li>• Eventually dispose of the DU components as LLW</li> </ul>
Oil well penetrators and drilling collars	<ul style="list-style-type: none"> <li>• Convert DUF<sub>6</sub> to metal</li> <li>• Manufacture penetrators and drilling collars</li> <li>• Use charges deep underground to open strata and collars deep underground to stabilize drill bit</li> <li>• Use of penetrators constitutes disposal; collars would require disposal as LLW when not lost in the subsurface</li> </ul>
<b>National resource reserve</b>	
Long-term storage  <u>Reference path for a portion of the DU</u>	<ul style="list-style-type: none"> <li>• Decide how much of which DU forms should be part of the reserve</li> <li>• Convert DUF<sub>6</sub> to the desired form(s) and include other existing forms of DU as appropriate</li> <li>• Package DU form(s)</li> <li>• Store DU in a retrievable storage facility until it is used or a new decision declares it to not be needed</li> </ul>

<sup>3</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas. Limits for release of uranium in commercial products need to be established.

**Table 3.2. Description of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate path	Candidate path description
<b>DUF<sub>6</sub></b>	
<b>Fuel Cycle Applications</b>	
AVLIS reenrichment	<ul style="list-style-type: none"> <li>Convert DUF<sub>6</sub> to metal</li> <li>Enrich DU metal to yield LEU</li> <li>Make LWR fuel with product</li> <li>Dispose of remaining DU tails using one of the direct disposal or other beneficial use paths</li> <li>Dispose of spent fuel in the potential repository</li> </ul>
Fast reactor fuel	<ul style="list-style-type: none"> <li>Convert DUF<sub>6</sub> to dioxide</li> <li>Mix DU dioxide with Pu dioxide to make fast reactor fuel and make U dioxide directly into blanket fuel</li> <li>Recycle DU recovered from fuel reprocessing until it is consumed by transmutation and fission</li> <li>Dispose of fission products in the potential repository as part of the high-level waste</li> </ul>
SILEX reenrichment	<ul style="list-style-type: none"> <li>Enrich DUF<sub>6</sub> without conversion to yield LEU</li> <li>Make LWR fuel with product</li> <li>Dispose of remaining DU tails using one of the direct disposal or other beneficial use paths</li> <li>Dispose of spent fuel in the potential repository</li> </ul>
<b>DU other than DUF<sub>6</sub></b>	
Reuse as is	<ul style="list-style-type: none"> <li>Sell materials to NRC licensees "as is" for less than the cost of disposal</li> </ul>
Reuse with further processing	<ul style="list-style-type: none"> <li>Non-DUF<sub>6</sub> is processed by, or on behalf of, the government to desirable forms</li> <li>DU materials are used by the government or sold to industry for commercial use subject to DOE policy on release of scrap metal</li> </ul>
<b>Fluorine products</b>	
Anhydrous and aqueous HF for industrial use <u>Reference path</u>	<ul style="list-style-type: none"> <li>Sell very slightly contaminated anhydrous and aqueous HF to industry for commercial use</li> </ul>
Calcium difluoride for industrial use	<ul style="list-style-type: none"> <li>Convert fluorine to CaF<sub>2</sub></li> <li>Sell very slightly contaminated CaF<sub>2</sub> to industry for commercial use</li> </ul>
Elemental fluorine for industrial use	<ul style="list-style-type: none"> <li>Sell very slightly contaminated elemental fluorine to industry for commercial use</li> </ul>
High-value compounds for industrial use	<ul style="list-style-type: none"> <li>Convert fluorine to high-value compounds such as BF<sub>3</sub> or fluoropolymers</li> <li>Sell very slightly contaminated high-value fluorine compounds to industry for commercial use</li> </ul>
<b>DUF<sub>6</sub> storage cylinders</b>	
Decontaminate and recycle metals	<ul style="list-style-type: none"> <li>Remove UF<sub>6</sub></li> <li>Wash cylinder internally</li> <li>Possibly perform more extensive decontamination, including surface cleaning or smelting</li> <li>Sell slightly contaminated metals to industry for commercial use subject to DOE policy on release of scrap metal</li> </ul>
Intact cylinders as LLW disposal packages	<ul style="list-style-type: none"> <li>Remove UF<sub>6</sub> and convert</li> <li>Refill cylinders with DU conversion product or some other LLW through opening cut in cylinder</li> <li>Weld patch over fill opening</li> <li>Store and dispose of refilled cylinders as LLW</li> </ul>
Refabricate metal for use in regulated areas	<ul style="list-style-type: none"> <li>Remove UF<sub>6</sub></li> <li>Smelt steel and form shielding blocks or waste containers</li> <li>Use slightly contaminated shielding blocks in radiologically regulated applications and eventually dispose of blocks</li> <li>Fill waste containers and dispose of as LLW</li> </ul>

## **4. EVALUATION OF CANDIDATE DISPOSITION PATHS**

The purpose of this section is to provide a comparative analysis and evaluation of the candidate DUF<sub>6</sub> conversion product disposition paths identified in Sect. 3 for the purpose of determining which of the paths require further barrier-reduction activities, whether such barrier-reduction activities are justified, and the relative priority of the activities. The approach used to accomplish this first involved establishing a set of criteria against which the paths would be evaluated, which is described in Sect. 4.1. Then, using the process described in Sect. 4.2, information related to the current status of each path was developed for each criterion and analyzed. The results of the analysis formed the basis for an evaluation of each path and assignment to one of four categories defined in Sect. 4.3. The category definition is based on whether the path should be pursued and, if so, the relative priority of the path.

### **4.1 DISPOSITION DECISION CRITERIA**

This section defines the criteria against which the candidate disposition paths for DUF<sub>6</sub> conversion products are evaluated. These criteria are used to analyze whether further barrier-reduction activities are justified for a particular path and, if so, the relative priority of such activities.

#### **4.1.1 Barrier Existence**

This criterion relates to whether any barriers were identified for a particular path. If there are no barriers, the path could be pursued without technical or institutional impediment, and it was assigned to Category D. DOE's interest is then reduced to ensuring that adequate supplies of the proper form of DU are available to allow the path to be implemented. Potential responses to this criterion were that barrier reduction is required or barrier reduction is not required. If barriers do exist, the relative priority of additional barrier-reduction activities is analyzed by considering the other criteria.

#### **4.1.2 Utilization of DU**

This criterion addresses the extent to which a particular DU disposition path could result in net consumption of the DU inventory over 20 years, which, in turn, provides justification for investing in barrier-reduction activities related to that path. The basis for this criterion is that the cost of barrier-reduction activities must be allocated to each unit of product, and a small number of product units would likely result in an unacceptably high product cost even if the product were otherwise economic.

### **4.1.3 Economics**

This criterion reflects the potential for a particular disposition path to result in net cost savings as compared with a reference path. The reference path is taken to be conversion of the DU to a stable form followed by disposal at a site where large amounts of DU would be acceptable in the near surface without need for a waste form matrix such as grout. The reference path for fluorine is industrial use as lower-value compounds (e.g., HF, CaF<sub>2</sub>). The reference path for cylinder disposition involves volume reduction and disposal as LLW. Consideration of *net* cost savings is intended to recognize the fact that some paths may involve increased cost to one part of DOE while reducing costs in another part of DOE.

### **4.1.4 Other Impacts**

This criterion encompasses the extent to which beneficial use of DU might improve or degrade some aspect of programs that are relevant to DOE, but where the impact is not reflected in cost. Examples might be a change in occupational or public health risk, or better reliability, performance, or predictability of some activity.

### **4.1.5 Technical Maturity**

This criterion reflects the likelihood that an investment in the technical aspects of development for a particular path would eventually lead to a deployable technology. It includes consideration of the current status, feasibility of the projected technical requirements, and likelihood of success.

### **4.1.6 Institutional Challenges**

This criterion is similar to that for technical maturity but addresses the likelihood that an investment to modify policy, regulatory, and legal barriers to a particular path is likely to be successful and allow a particular disposition path to be deployed.

## **4.2 ANALYSIS AND EVALUATION PROCESS FOR DISPOSITION PATHS**

The process that was used to analyze and evaluate the potential disposition paths for DU is summarized as follows:

- Background information on specific topics was collected and organized by researchers at five national laboratories (Argonne National Laboratory, INEEL, Lawrence Livermore National Laboratory, Oak Ridge National Laboratory, and Pacific Northwest National Laboratory) that had extensive experience in specific aspects of DU disposition as a result of prior programmatic involvements.

- The contributions of individual researchers were consolidated into a draft report and circulated to all involved researchers plus multiple parts of DOE [NE, EM, Office of Civilian Radioactive Waste Management (RW)] for round-robin review. The resulting background information is summarized in Sects. 2 and 3.
- The researchers plus representatives of all relevant DOE organizations convened in a workshop. The workshop involved two major activities: (a) presentation and analysis of the background information developed by each researcher and (b) assignment of the potential DU disposition paths to one of the four categories based on the criteria described in Sect. 4.1. Each path was assigned to a category using the following methodology:
  - Each attendee independently assigned each path to one of the four following categories:
    - @ Further barrier-reduction activities recommended
    - @ Further barrier-reduction activities should be considered
    - @ Further barrier-reduction not recommended
    - @ No additional federal barrier-reduction activities needed
  - The assignments were then tallied, the paths were provisionally assigned to one of the four groups, and these results were shared with the entire group.
  - The resulting recommendations were discussed, modified slightly, and adopted by consensus.
- The results of the workshop are documented as the remainder of this report, which was reviewed by the workshop attendees as well as other elements of DOE.

The results of analyzing the background information is summarized in Appendix A for the disposal (Table A.1) and beneficial-use (Table A.2) paths. The recommended category assignments of each path are described in the following section and the tables associated with it.

### **4.3 RECOMMENDED CATEGORIZATION OF DISPOSITION PATHS**

The information summarized in Appendix A was used by the workshop attendees to evaluate the disposition paths for DU-related materials and to assign them to one of the four categories using the process summarized in Sect. 4.2. The results of this evaluation and the associated explanation are summarized in the following sections, which correspond to the four categories. Within each category, the disposition paths are presented in alphabetical order.

#### **4.3.1 Category A: Barrier Reduction Recommended**

The disposition paths assigned to Category A are the most promising of all the paths considered or constitute a reference approach that could be reliably implemented. In general, the beneficial-use disposition paths in this category could use the majority of the DU, have good potential to yield net system-wide cost savings relative to the reference case or other benefits that justify their cost, and are judged to have barriers that are likely to be overcome in a straightforward manner.



The workshop concluded that barrier-reduction activities in Category A should be immediately funded at a level sufficient to bring them to the point where they can be reliably deployed. The paths in this category and an explanation for their inclusion are summarized in Table 4.1.

**Table 4.1. Category A: Disposition paths for which barrier-reduction activities are recommended**

Path	Explanation
<b>DUF<sub>6</sub></b>	
LLW Disposal	<ul style="list-style-type: none"> <li>• <u>Reference</u> disposition path for all DU not beneficially used</li> <li>• Substantial post-conversion cost for disposal of unconsolidated packaged DU at a DOE site</li> <li>• Establishing DU-specific requirements to meet the WAC, negotiating terms and conditions, and possible integration with long-term storage are barriers to be reduced</li> </ul>
Long-term storage	<ul style="list-style-type: none"> <li>• <u>Reference</u> path needed to maintain limited amounts of unique forms of DU as a national resource</li> <li>• Desirable contingency in case other disposition options are delayed</li> <li>• Net cost that grows with the length of storage</li> <li>• Barrier-reduction activities should focus on ensuring long-term package integrity and operating efficiencies such as relying on recovery from a disposal site in the case of an urgent national need</li> </ul>
Heavy concrete	<ul style="list-style-type: none"> <li>• Focus on uses such as radiation shielding and spent nuclear fuel (SNF)/HLW transportation and storage</li> <li>• Use of DU-based heavy concrete is prohibited in the potential repository because cementitious matrices might adversely affect water chemistry, WP corrosion, and radionuclide migration</li> <li>• Significant previous development; barrier reduction appears straightforward</li> <li>• Potential for net system cost savings by deferring DU disposal and end-of-life use as an LLW package</li> </ul>
<b>DU other than DUF<sub>6</sub></b>	
LLW disposal	<ul style="list-style-type: none"> <li>• <u>Reference</u> path for all DU other than DUF<sub>6</sub> that is not used for beneficial purposes</li> <li>• Can accommodate all of the non-DUF<sub>6</sub> inventory</li> <li>• This should be pursued in case the private sector cannot, or will not, absorb all of this inventory</li> <li>• Barrier-reduction activities should be limited and focused on meeting WAC at DOE disposal sites</li> </ul>
<b>DUF<sub>6</sub> storage cylinders</b>	
Intact cylinders as LLW disposal packages	<ul style="list-style-type: none"> <li>• <u>Reference</u> path for disposition of all cylinders except those having sufficiently impaired integrity</li> <li>• Has been studied, and barriers are minimal</li> <li>• Significant net savings as compared with cylinder disposal and other cylinder disposition options</li> </ul>

#### **4.3.2 Category B: Further Barrier Reduction Should Be Considered**

The disposition paths assigned to Category B have some promise of being justifiable based on cost or other improvements. Most of these can also use a significant portion of the DU inventory. However, compared with Category A paths, these paths suffer from some combination of a lower probability of yielding net system-wide cost savings or other benefits, and being able to overcome their respective barriers successfully. In particular, many of these paths would involve use of significant amounts of DU outside of radiologically regulated areas. Such paths face regulatory uncertainties and issues of risk perception that can present significant institutional barriers. Previous attempts to overcome similar barriers have been demonstrably unsuccessful (e.g., NRC's attempt to establish "Below Regulatory Concern" levels for releasing materials containing minuscule amounts of radioactivity to unregulated disposal facilities). Earlier this year, in response to concerns about the release of volumetrically contaminated nickel from the East Tennessee Technology Park, the Secretary of Energy established a moratorium prohibiting the release of all volumetrically contaminated metals from DOE facilities to give the NRC time to develop national standards for volumetrically contaminated materials, allow the public to weigh in on the development of a national policy, and permit DOE to establish its moratorium policy, directives, and guidance in this regard. In addition, on July 13, 2000, DOE suspended the unrestricted release for recycling of scrap metals from radiation areas within DOE facilities. This suspension is to remain in effect until DOE implements improved release criteria and information management requirements relating to these materials. The impact of these activities and decisions on potential use of fluorine products years hence is unknown.

DOE should selectively consider investing in the DU disposition paths in Category B, shown in Table 4.2, based on judgments concerning the relative merits of specific proposals and the availability of funds.

**Table 4.2. Category B: Disposition paths for which barrier-reduction activities should be considered**

Path	Explanation
<b>DUF<sub>6</sub></b>	
Aluminum-refining electrodes	<ul style="list-style-type: none"> <li>• Can use up to 100% of the DU inventory as oxide in an industrial environment, but with potential for trace amounts of DU in aluminum products and slag waste</li> <li>• Significant technical issues need to be addressed, especially the rate at which DU oxide electrodes dissolve in the aluminum product and slag</li> <li>• Limited generic barrier-reduction activities to define the specific regulatory framework applicable to use of this product, evaluate degradation rates and performance, and provide key data on electrode solubility and performance</li> </ul>
Catalyst for fuel cells and steam reforming	<ul style="list-style-type: none"> <li>• Can use up to 50% of DU inventory as the oxide in an industrial environment or, conceivably, in consumer products (e.g., small fuel cells for vehicles or homes)</li> <li>• Limited generic barrier-reduction activities to define the specific regulatory framework applicable to use of this product and limited investigation of catalytic performance</li> </ul>
Catalyst for automotive exhaust	<ul style="list-style-type: none"> <li>• Can use up to 50% of the DU inventory as the oxide in consumer products</li> <li>• Recovery and recycle of used converters are possible, but efficiency of recovery is uncertain</li> <li>• Limited generic barrier-reduction activities to define the specific regulatory framework applicable to use of this product and limited research on catalytic performance</li> </ul>
Heavy-lifting-vehicle counterweights and high-traction devices	<ul style="list-style-type: none"> <li>• Can use up to 100% of the DU inventory as metal in an unregulated industrial environment</li> <li>• Higher cost of counterweights may be offset by warehouse cost reductions in the case of forklifts, which are a major potential application. Should also reduce forklift fatality frequency</li> <li>• Limited generic barrier-reduction activities to define the specific regulatory framework applicable to use of this product in unregulated areas</li> </ul>
Invert	<ul style="list-style-type: none"> <li>• Use of DU-based heavy concrete is prohibited in the potential repository because cementitious matrices may adversely affect water chemistry, WP corrosion, and radionuclide migration</li> <li>• Consider limited investigation of inserting DU oxides in sealed cells formed by invert made of steel plate that could provide ballast and might enhance performance of the potential repository</li> </ul>
Mined cavity disposal	<ul style="list-style-type: none"> <li>• Could use up to 100% of the DU inventory</li> <li>• Significantly more expensive than near-surface disposal</li> <li>• Limited activities recommended to ascertain the terms and conditions for mined cavity DU disposal</li> </ul>
Oil well penetrators and drilling collars	<ul style="list-style-type: none"> <li>• Can use up to 50% of DU inventory as the metal</li> <li>• Some historical use in this application</li> <li>• Limited generic barrier-reduction activities to define the specific regulatory framework applicable to use of this product and limited evaluation of market conditions</li> </ul>
Package fill	<ul style="list-style-type: none"> <li>• Net cost increase may be justified if the performance of the potential repository were to be improved</li> <li>• Further study would be needed before it can be determined if package fill would be beneficial or detrimental to the performance of the potential repository</li> <li>• Timing is an issue relative to an environmental impact statement (EIS) and license application for the potential repository</li> </ul>
Uranium silicide	<ul style="list-style-type: none"> <li>• Could use up to 100% of the DU inventory</li> <li>• One-step defluorination and conversion to a form potentially suitable as aggregate in heavy concrete or for disposal offers the possibility of a less costly, second-generation DUF<sub>6</sub> conversion process</li> <li>• Concept is presently theoretical, and significant R&amp;D would be required for several years</li> <li>• Consider limited investment to elucidate chemistry</li> </ul>

**Table 4.2. Category B: Disposition paths for which barrier-reduction activities should be considered**

Path	Explanation
<b>DU other than DUF<sub>6</sub></b>	
Reuse as is	<ul style="list-style-type: none"> <li>• No need for material within the DOE system</li> <li>• Amount of DU is relatively small (about 25,500 MTU), composed of multiple lots of different forms of DU with variable impurities that are not well known</li> <li>• DOE does not have the capability to process this material without an investment that would not be cost effective</li> <li>• Limited activities needed to (a) characterize materials so they can be beneficially reused over a period of years at a price that results in a net cost reduction and (b) examine liabilities of high-profile impurities such as Pu</li> </ul>
<b>Fluorine products</b>	
High-value fluorine compounds for industrial use	<ul style="list-style-type: none"> <li>• Can use up to 100% of fluorine</li> <li>• Use of very slightly uranium-contaminated fluorine is regulated for users under present NRC regulations</li> <li>• Potential to yield larger revenues from fluorine by-product</li> <li>• Only at the concept stage; potential exists to flood small markets</li> <li>• Barrier-reduction activities need to focus on synergistic market analysis, flowsheet development, and enabling policy changes</li> </ul>

### 4.3.3 Category C: Further Barrier Reduction Not Recommended

The disposition paths assigned to Category C (Table 4.3) have limited promise of being justifiable compared with paths in Categories A and B. Most of these can use a significant portion of the DU inventory. However, these paths either have high fundamental barriers (e.g., substantial technical impediments, conflict with U.S. laws or policies) or they perform the same function as other paths that are much more promising. There is little chance that additional work would make Category C disposal paths viable.

DOE should not invest in the paths in Category C for the purposes of DU material disposition.

**Table 4.3. Category C: Disposition paths for which barrier-reduction activities are not recommended**

<b>Path</b>	<b>Explanation</b>
<b>DUF<sub>6</sub></b>	
AVLIS reenrichment	<ul style="list-style-type: none"> <li>• Net utilization is &lt;5% of inventory</li> <li>• Requires significant increase in the cost of natural uranium for economic break-even</li> <li>• U.S. AVLIS Program has been terminated</li> <li>• Other laser-based processes are more promising because they can accept DUF<sub>6</sub> as feed</li> </ul>
Cement-Lock™	<ul style="list-style-type: none"> <li>• Process is primarily for treatment of hazardous organic chemical waste to yield a construction material</li> <li>• Flowsheet that can handle DUF<sub>6</sub> with the difficult fluorine by-product stream needs considerable development</li> <li>• The specific regulatory framework applicable to use of this product is not defined, but the general regulatory framework is not conducive to such use</li> </ul>
DUPoly	<ul style="list-style-type: none"> <li>• Potential for combustion and radiolytic hydrogen production</li> <li>• Organic chemicals are prohibited in the potential repository</li> <li>• Costs may be more than cement-based heavy concretes for the same function</li> </ul>
Catalyst for fluid cracking and to promote oxidation	<ul style="list-style-type: none"> <li>• Would use &lt;1% of the inventory in an industrial environment with potential for traces of DU in products</li> <li>• Use is in an unregulated area</li> </ul>
PYRUC	<ul style="list-style-type: none"> <li>• Uses a complicated process to coat UO<sub>2</sub> microspheres mixed with binder matrix to achieve the same result as less costly heavy concretes based on simple DU oxide aggregates</li> </ul>
Salt mine disposal as DUF <sub>6</sub>	<ul style="list-style-type: none"> <li>• Has the potential to accommodate entire inventory without need for conversion</li> <li>• Chloride-based salt has potential to be relatively compatible with DUF<sub>6</sub></li> <li>• Development of a new facility for this purpose could require new statutory authority and regulations, and is likely to be costly and contentious</li> <li>• Potential for reaction of DUF<sub>6</sub> with brine in salt beds</li> </ul>
Subsurface engineered vault disposal	<ul style="list-style-type: none"> <li>• This type of facility could accommodate the entire inventory of DU</li> <li>• Such a facility is unique and more costly than near-surface disposal but offers few additional benefits</li> </ul>
<b>DU other than DUF<sub>6</sub></b>	
Reuse with further processing	<ul style="list-style-type: none"> <li>• Would require refurbishment and restart of closed DOE facilities</li> <li>• Small amount of material would result in a unit cost likely to be much larger than the cost of disposal</li> </ul>
<b>DUF<sub>6</sub> storage cylinders</b>	
Decontaminate and recycle metals	<ul style="list-style-type: none"> <li>• Much more costly than other alternatives</li> <li>• Exception is the small amount of Monel in cylinder valves, which is presently being recovered using existing processes</li> </ul>
Reuse intact cylinders for newly generated DUF <sub>6</sub>	<ul style="list-style-type: none"> <li>• Only a small volume of existing cylinders utilized</li> </ul>

#### **4.3.4 Category D: No Barrier Reduction Requiring Federal Action Needed**

The disposition paths assigned to Category D, shown in Table 4.4, have merit but also have attributes that make barrier-reduction activities currently undesirable. This situation occurs for a number of reasons:

- The potential use is laudable but is so far in the future that near-term investments cannot be justified.
- The potential use represents existing practice, and no further federal investment for the purpose of supporting DU disposition is needed.
- The potential use is already being adequately supported.

DOE should monitor programs related to disposition paths that involve distant future demands and be prepared to consider investing in barrier-reduction activities if the demand is imminent. For uses involving existing practice or meeting the needs of ongoing programs, DOE should monitor these programs and be prepared to supply appropriate DU feed material from its inventory as required.

#### **4.4 SUMMARY**

The evaluation results discussed in Sect. 4.3 are summarized in Table 4.5 by category and material type.

**Table 4.4. Category D: Disposition paths for which further barrier-reduction activities are not needed**

Path	Explanation
<b>DUF<sub>6</sub></b>	
Backfill	<ul style="list-style-type: none"> <li>• Could use up to 100% of the DU inventory</li> <li>• Decision on whether to use any backfill will not be made for decades, and the need is yet further in the future</li> <li>• Such use may be worthwhile and should be considered when the backfill situation is clarified</li> <li>• Backfill in emplacement drifts is not presently part of the reference design of the potential repository and is planned to be installed only in nonemplacement drifts. This may be reevaluated in the future, but in any case, backfill would not be installed any earlier than the 22nd century and maybe later</li> </ul>
Ceramics for Pu disposition	<ul style="list-style-type: none"> <li>• DU will be used for this purpose, but the amount that might be used is &lt;0.1% of the inventory</li> <li>• This use of DU is already being supported by DOE's Office of Fissile Material Disposition (DOE-MD)</li> <li>• DOE should be prepared to supply the required DU once the requirements are known</li> </ul>
Dilution of HEU(F)	<ul style="list-style-type: none"> <li>• The amount that might be used is &lt;5% of the inventory</li> <li>• The use of DU in this application is already being supported by DOE-MD</li> <li>• DOE should be prepared to supply the required DUF<sub>6</sub> once the precise specifications are known</li> </ul>
DU metal shielding	<ul style="list-style-type: none"> <li>• Use of DU metal in spent fuel casks and other applications is existing practice, and a number of such casks and other applications presently exist</li> <li>• DOE should be prepared to supply DU feed to the private sector as demand requires, but the amount of such demand is expected to be small</li> </ul>
Fast reactor fuel	<ul style="list-style-type: none"> <li>• This could consume the entire inventory of DU over many years</li> <li>• As a matter of policy, the United States is not supporting development of fast reactors, and such a program does not appear likely because of the low cost of natural uranium and concerns about the recycle of plutonium</li> <li>• DOE should reevaluate DU needs for fast reactors if such a program is considered in the future</li> </ul>
MOX fuel for Pu disposition	<ul style="list-style-type: none"> <li>• The amount that might be used is &lt;1% of the inventory</li> <li>• Use of DU in this application is already being supported by DOE-MD</li> <li>• DOE should be prepared to supply the required DU once the requirements are known</li> </ul>
SILEX reenrichment	<ul style="list-style-type: none"> <li>• Net utilization is &lt;5% of inventory if all DU were reenriched</li> <li>• Claimed to have more potential than AVLIS or other atomic processes</li> <li>• Process is not well developed for uranium (demonstrated only at the laboratory scale) but is being supported by USEC: has not been commercially demonstrated</li> <li>• Requires increase in the cost of natural uranium for economic break-even ranging from slight for DU &gt;0.4% <sup>235</sup>U to substantial for the bulk of the DU in the 0.2–0.3% range</li> <li>• DOE should monitor development and be prepared to supply higher-value tails if this process is deployed by others</li> </ul>
<b>Fluorine products</b>	
Anhydrous and aqueous HF for industrial use	<ul style="list-style-type: none"> <li>• Uranium concentrations are sufficiently low that users of released material are not required to obtain an NRC license, but there may be some reluctance by the commercial sector in using these materials in non-nuclear applications until DOE's policy relating to the release of scrap metal is finalized</li> <li>• HF is frequently recycled, although calcium difluoride is sometimes the preferred product when it must be transported off-site</li> </ul>
Calcium difluoride for industrial use	<ul style="list-style-type: none"> <li>• Uranium concentrations are sufficiently low so that users of released materials are not required to obtain an NRC license, but present DOE policy prohibits release of such material and this could become permanent</li> <li>• Calcium difluoride is frequently recycled and is often the preferred product when it must be stored or transported off-site</li> </ul>
Dispose of fluorine	<ul style="list-style-type: none"> <li>• This is established practice</li> <li>• Disposal of fluorine products from commercial defluorination of DUF<sub>6</sub> is uncommon because these products have value</li> </ul>
Elemental fluorine for industrial use	<ul style="list-style-type: none"> <li>• Uranium concentrations are sufficiently low that users of released material are not required to obtain an NRC license</li> <li>• Fluorine is not an item in intersite commerce because it is effectively untransportable in significant amounts</li> </ul>

**Table 4.4. Category D: Disposition paths for which further barrier-reduction activities are not needed**

Path	Explanation
<b>DUF<sub>6</sub> storage cylinders</b>	
Dispose of metals	<ul style="list-style-type: none"> <li>• This is established practice because the cleaned cylinders can easily meet WAC at LLW disposal sites</li> <li>• This path is more costly than use of the cylinders as waste containers, but it might be justified under some circumstances (e.g., for cylinders that do not have integrity adequate for use as a WP)</li> </ul>
Refabricate metal for use in regulated areas	<ul style="list-style-type: none"> <li>• This is established practice because industry has made waste containers from recycled contaminated steel</li> <li>• The cost of smelting and refabrication is estimated to be greater than the value of the steel</li> </ul>



Table 4.5. Summary of disposition path evaluation for products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>

Category A: Barrier-reduction activity recommended	Category B: Barrier-reduction activity should be considered	Category C: Barrier-reduction activity is not recommended	Category D: Barrier-reduction activity is not needed
<b>DUF<sub>6</sub></b>			
<ul style="list-style-type: none"> <li>• LLW disposal</li> <li>• Long-term storage</li> <li>• Heavy concrete</li> </ul>	<ul style="list-style-type: none"> <li>• Aluminum-refining electrodes</li> <li>• Catalyst for fuel cells and steam reforming</li> <li>• Catalyst for automotive exhaust</li> <li>• Heavy-lifting-vehicle counterweights and high-traction devices</li> <li>• Invert</li> <li>• Mined cavity disposal</li> <li>• Oil well penetrators and drilling collars</li> <li>• Package fill</li> <li>• Uranium silicide</li> </ul>	<ul style="list-style-type: none"> <li>• AVLIS reenrichment</li> <li>• Cement-Lock™</li> <li>• DUPoly</li> <li>• Catalyst for fluid cracking and to promote oxidation</li> <li>• PYRUC</li> <li>• Salt mine disposal as DUF<sub>6</sub></li> <li>• Subsurface engineered vault disposal</li> </ul>	<ul style="list-style-type: none"> <li>• Backfill</li> <li>• Ceramics for Pu disposition</li> <li>• Dilution of HEU(F)</li> <li>• DU metal shielding</li> <li>• Fast reactor fuel</li> <li>• MOX fuel for Pu disposition</li> <li>• SILEX reenrichment</li> </ul>
<b>DU other than DUF<sub>6</sub></b>			
<ul style="list-style-type: none"> <li>• Reuse as is</li> </ul>	<ul style="list-style-type: none"> <li>• LLW disposal</li> </ul>	<ul style="list-style-type: none"> <li>• Reuse with further processing</li> </ul>	
<b>Fluorine products</b>			
	<ul style="list-style-type: none"> <li>• High-value fluorine compounds for industrial use</li> </ul>		<ul style="list-style-type: none"> <li>• Anhydrous and aqueous HF for industrial use</li> <li>• Calcium difluoride for industrial use</li> <li>• Dispose of fluorine product</li> <li>• Elemental fluorine for industrial use</li> </ul>
<b>DUF<sub>6</sub> storage cylinders</b>			
<ul style="list-style-type: none"> <li>• Intact cylinders as LLW disposal packages</li> </ul>		<ul style="list-style-type: none"> <li>• Decontaminate and recycle metals</li> </ul>	<ul style="list-style-type: none"> <li>• Dispose of metals</li> <li>• Refabricate metal for use in regulated areas</li> </ul>

## 5. RECOMMENDED DISPOSITION BARRIER-REDUCTION ACTIVITIES

Section 4 analyzed and evaluated a number of DU disposition paths leading to identification of paths that were recommended or should be considered for further development. The purpose of this section is to (a) present a consolidated list of barrier-reduction activities for these two groups, (b) indicate where specific activities would benefit multiple paths, and (c) identify major cross-cutting or systems issues that should be addressed. This section does not attempt to prioritize the activities; nor does it constitute a plan for implementing a DU disposition program.

### 5.1 BARRIER-REDUCTION ACTIVITIES TO SUPPORT RECOMMENDED PATHS

Barrier-reduction activities that are required to support the recommended (Category A) paths are summarized in Table 5.1. The major component of these activities would be a broad spectrum of activities to bring DU disposition paths involving heavy concrete and fill in packages destined for the potential repository to the point where these technologies could be deployed, if justified, by the increased knowledge obtained from the activities. Other paths for which barrier-reduction activities are recommended should involve targeted investments to address specific barriers.

**Table 5.1. Category A: Barrier-reduction activities to support recommended paths**

Path	Barrier-reduction activities
<b>DUF<sub>6</sub></b>	
LLW disposal	<ul style="list-style-type: none"> <li>Technical studies to support establishment of DU-specific requirements that meet WAC for disposal of DU oxide and tetrafluoride and result in minimal DU disposition costs</li> </ul>
Long-term storage	<ul style="list-style-type: none"> <li>Establish national resource reserve requirements for various forms of DU</li> <li>Develop specifications for a long-term WP for DU oxide, tetrafluoride, and metal</li> <li>Limited systems studies to determine optimal long-term storage options (e.g., disposal from which DU could be recovered if an urgent national need arose)</li> </ul>
Heavy concrete	<ul style="list-style-type: none"> <li>Measurement of thermal and mechanical properties of heavy concrete (e.g., thermal conductivity, strength, and chemical stability) to be combined with specific cask designs in order to meet overall functional requirements necessary to obtain NRC approval of containers</li> <li>Optimization and measurement of nuclear shielding properties (e.g., direct measurements of shielding attenuation utilizing neutron and gamma sources, and use of materials containing boron and hydroxide)</li> <li>Development of high-performance heavy concrete (e.g., increase flexural strengths impact resistance, energy absorption, and fracture toughness; primarily to be achieved through addition of metal fibers)</li> <li>Fabrication of prototype structures and samples, including development of preplaced aggregate and pumped grout</li> <li>Further examination and modeling of oxidation processes of DU aggregate under conditions of elevated temperature and humidity when surrounded (and not surrounded) by the aluminosilicate grain boundary phase in the concrete matrix in order to predict stability over long periods of time</li> <li>Optimization of the process for producing the DU aggregate and formulation of the heavy concrete</li> <li>Facilitation of manufacturer-purchaser relationships to establish a market for heavy concrete products</li> </ul>
<b>DU other than DUF<sub>6</sub></b>	
Reuse as is	<ul style="list-style-type: none"> <li>Additional characterization of impurities to allow various lots of DU other than DUF<sub>6</sub> to be matched with potential users</li> </ul>
<b>DUF<sub>6</sub> storage cylinders</b>	
Intact cylinders as LLW disposal packages	<ul style="list-style-type: none"> <li>Procedures for detecting substandard cylinders, filling cylinders, and sealing penetrations</li> <li>Demonstration of use of cylinders as an LLW package</li> </ul>

## **5.2 BARRIER REDUCTION NEEDED TO SUPPORT PATHS THAT SHOULD BE CONSIDERED**

Barrier-reduction activities that are required to support paths that should be considered (Category B) are summarized in Table 5.2. Investigating use of DU oxide in steel invert in the potential civilian repository would require a broad spectrum of work. Such an investigation can benefit from results that would be produced by barrier-reduction activities concerning package fill for the potential repository and heavy concrete. All other paths in this category should be considered for targeted investment to pursue specific issues, after which additional decisions on their worth would be required. In both cases, many activities benefit multiple projects, and these are discussed in the next section.

## **5.3 CROSS-CUTTING BARRIER-REDUCTION ACTIVITIES**

Many of the DU disposition paths for which barrier-reduction activities are recommended or should be considered have common barrier-reduction needs in two areas. The first area includes activities that benefit multiple paths because the paths involve the same set of materials: DU, fluorine, and cylinders. The second area includes activities, typically called “systems studies,” that are needed to design and optimize any program involving multiple components. Recommended barrier-reduction activities in these two areas were developed in the DU disposition workshop and are summarized below.

### **5.3.1 Barrier-Reduction Activities Supporting Multiple Paths**

Barrier-reduction activities that could benefit multiple disposition paths for DU-related materials are as follows:

- Establishing the policy and regulatory framework for the extent and conditions under which DU-bearing products could be used in various nongovernmental applications. This regulatory framework needs to be pursued in the context of concern over the trace amounts of some fission-product and transuranic elements potentially present in  $\text{DUF}_6$  and consideration of rule making concerning release of contaminated solids by the NRC. Earlier this year, in response to concerns about the release of volumetrically contaminated nickel from the East Tennessee Technology Park, the Secretary of Energy established a moratorium prohibiting the release of all volumetrically contaminated metals from DOE facilities to give the NRC time to develop national standards for volumetrically contaminated materials, allow the public to weigh in on the development of a national policy, and permit DOE to establish its moratorium policy, directives, and guidance in this regard. In addition, on July 13, 2000, DOE's Secretary Richardson suspended the unrestricted release for recycling of scrap metals from radiation areas within DOE facilities. This suspension is to remain in effect until DOE implements improved release criteria and information management requirements relating to these materials.

**Table 5.2. Category B: Barrier-reduction activities to support paths that should be considered**

Path	Barrier-reduction activities
<b>DUF<sub>6</sub></b>	
Aluminum-refining electrodes	<ul style="list-style-type: none"> <li>• A generic effort to define the framework applicable to DOE release and private-sector use of DU (see Sect. 5.3)</li> <li>• Limited initial investment in the following to allow this option to be evaluated:               <ul style="list-style-type: none"> <li>– Determining the solubility and corrosion rate of UO<sub>2</sub> in the cryolite melt at 950EC</li> <li>– Establishing the electrical and mechanical properties of UO<sub>2</sub>-Cu composites</li> </ul> </li> </ul>
Catalyst for automotive exhaust Catalyst for fuel cells and steam reforming	<ul style="list-style-type: none"> <li>• A generic effort to define the framework applicable to DOE release and private-sector use of DU (see Sect. 5.3)</li> <li>• Limited initial investment in the following to allow this option to be evaluated:               <ul style="list-style-type: none"> <li>– Synthesis techniques for mesostructured uranium oxide catalysts</li> <li>– Measurement of catalyst activity</li> <li>– Thermal and mechanical stability of promising catalysts</li> </ul> </li> <li>• Evaluation of facility contamination and decontamination issues</li> </ul>
Heavy-lifting-vehicle counterweights and high-traction devices	<ul style="list-style-type: none"> <li>• A generic effort to define the framework applicable to DOE release and private-sector use of DU (Sect. 5.3)</li> </ul>
Invert	<ul style="list-style-type: none"> <li>• <i>Uranium form.</i> The DU could be added as an oxide, silicate, or other chemical form. The preferred form to maximize invert performance per dollar invested must be determined</li> <li>• <i>Material compatibility.</i> The compatibility of the DU with the engineered barrier system must be demonstrated</li> <li>• <i>Ion-exchange capacity.</i> The WP-fill ion-exchange studies described earlier are also needed for invert applications. In addition, studies would be required to determine how much groundwater from the WP could realistically be expected to flow through the degraded invert with subsequent removal of the radionuclides</li> <li>• <i>Criticality.</i> Criticality studies are required to determine the degree of isotopic exchange between the invert and SNF uranium as groundwater flows through the degraded WP and invert</li> <li>• <i>Performance assessment.</i> An integrated model of WP performance with DU is required to demonstrate the impact of DU on system performance of the potential repository</li> <li>• <i>Economic analysis.</i> Cost-benefit analysis is required</li> <li>• <i>Legal and institutional analysis.</i> If DU were used in this application, it would presumably be a useful material—similar to the metal in the WP and thus legally may be treated like the WP materials of construction. However, it might also be considered a waste. An analysis of the issues associated with this possible duality is required</li> </ul>
Mined cavity disposal	<ul style="list-style-type: none"> <li>• Limited investment to pursue potential disposal in a mined cavity</li> </ul>
Oil well penetrators and drilling collars	<ul style="list-style-type: none"> <li>• A generic effort to define the framework applicable to DOE release and private-sector use of DU (see Sect. 5.3)</li> <li>• Limited investment to achieve better understanding of the needs and barriers regarding this use of DU</li> </ul>

**Table 5.2. Category B: Barrier-reduction activities to support paths that should be considered**

Path	Barrier-reduction activities
<b>DUF<sub>6</sub></b>	
Package fill	<ul style="list-style-type: none"> <li>• <i>Fill permeability.</i> Initial studies indicate that DU fill should lower the permeability of the WP to water and gas flow. Experiments and supporting models are required to (1) quantify this effect in terms of (a) maintaining chemical reducing conditions within the WP to prevent degradation of SNF and (b) minimizing water flow and subsequent transport of radionuclides from the WP and (2) analyze the effect of fill swelling on the SNF</li> <li>• <i>Ion-exchange capacity.</i> DU oxides may act as inorganic ion-exchange material that reduces release of radionuclides from the WP. This effect must be quantified—particularly for long-lived radionuclides that are important to performance of the potential repository</li> <li>• <i>Analogue Behavior.</i> Some natural UO<sub>2</sub> has remained intact under geological conditions similar to Yucca Mountain, Nevada, for several million years. A better understanding of the mechanisms (chemical reducing conditions, protective layers, etc.) is needed to provide licensing support that such a WP will minimize releases for very long time periods</li> <li>• <i>Criticality control.</i> DU lowers the average uranium enrichment of the WP below that required for nuclear criticality. Additional studies are required to confirm criticality control as the WP degrades and materials are transported from the WP</li> <li>• <i>WP and fill design.</i> If DUO<sub>2</sub> fill is used, the optimum WP and fill design to maximize performance and reduce costs may change. The incentives to change these components with a DU fill system must be evaluated</li> <li>• <i>Thermal properties and heat transfer.</i> The replacement of the baseline fill gas within a WP with DU oxide particles will have an effect on heat transfer. Limited analytical studies based on very uncertain thermal data indicate that this is a small effect [Forsberg 1995]. The thermal properties of candidate fill materials need to be measured and used as input to sophisticated heat transfer modeling techniques that have been validated by benchmark experiments</li> <li>• <i>Radiation shielding.</i> The reduction in radiation emitted by the WP and its effects on operation and post-closure performance of the potential repository have not been investigated</li> <li>• <i>Emplacement technique.</i> The Canadians investigated many fill materials, and their SNF has smaller clearances between fuel pins than does LWR SNF. However, DUO<sub>2</sub> particulate properties and LWR design features are somewhat different from their counterparts in the Canadian work, and, thus, confirmatory studies are required</li> <li>• <i>Optimization.</i> The preferred oxide and mix of particle sizes have not been investigated</li> <li>• <i>Performance assessment.</i> An integrated model of WP performance with DU is required to demonstrate the impact of DU on system performance of the potential repository</li> <li>• <i>Economic analysis.</i> A thorough analysis of the cost of using fill material is required</li> <li>• <i>Legal and institutional analysis.</i> If DU were used in this application, it would presumably be a useful material—similar to the metal in the WP and thus legally may be treated like the WP materials of construction. However, it might also be considered a waste. An analysis of the issues associated with this possible duality is required</li> </ul>
Uranium silicide	<ul style="list-style-type: none"> <li>• Fundamental research on the chemistry of uranium silicide production, leading to proof-of-concept experiments involving the production of small amounts of aggregate</li> <li>• If successful, the entire suite of activities listed under heavy concrete in Table 5.1 must be undertaken</li> </ul>
<b>DU other than DUF<sub>6</sub></b>	
Disposal	<ul style="list-style-type: none"> <li>• Additional characterization of impurities to establish acceptability for disposal</li> </ul>
<b>Fluorine products</b>	
Recycle high-value fluorine compounds	<ul style="list-style-type: none"> <li>• A generic effort to define the framework applicable to DOE release and private-sector use of DU-contaminated material (see Sect. 5.3)</li> <li>• Studies of the chemistry of fluorine as it relates to producing potential high-value fluorine compounds such as BF<sub>3</sub>, SF<sub>6</sub>, fluoropolymers</li> <li>• Market studies to elucidate the preferred mix of higher-value fluorine products, potential impacts on the fluorine industry, and mechanisms for ameliorating the impacts</li> <li>• Assuming successful outcomes of the above, engineering development and demonstration of an integrated process for producing higher-value fluorine projects would be required</li> </ul>

- Establishing the framework of roles and responsibilities for DU use, including:
  - Responsibility for products made from DU-related materials
  - Regulatory responsibilities
  - Budget responsibilities
  - Market development and establishment of incentive structures
  - Interfaces with other programs that might use DU
- Fostering public awareness of issues and benefits related to use of DU-bearing products<sup>4</sup>

### 5.3.2 Systemic Barrier-Reduction Activities

Barrier-reduction activities that are needed to establish the overall architecture of DU-related material disposition are as follows:

- Characterizing a reference disposition scenario for DU-related material disposition against which alternative disposition scenarios can be compared. These should cover everything between conversion and disposal and all surplus DOE DU and related materials.
- Conducting systems analysis and trade studies to identify preferred approaches for disposition of DU-related materials, and as a basis for allocation of costs and benefits.

## 5.4 RESEARCH NEEDED TO SUPPORT DU DISPOSITION

It is desirable to continue generating knowledge leading to new uses of DU conversion products and to provide the scientific underpinning for known uses of these products. Topics relevant to the disposition of DU-related materials that could constitute topics for existing or supplemental mission-relevant research programs were developed in the DU disposition workshop and are as follows:

- Long-term interaction of DU oxides, metal, and tetrafluoride with container materials and the environment to support paths concerning package fill and invert for the potential repository, disposal, and long-term storage
- Alteration of uranium oxides and tetrafluoride in aqueous or cement media over the long term to support paths concerning high-density concrete, fill and invert for the potential repository, and disposal
- Processes for producing higher-value fluorine compounds that might reduce the cost of DUF<sub>6</sub> disposition

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<sup>4</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.

- Direct conversion of  $\text{DUF}_6$  to useful products (e.g., DU conversion producing  $\text{USi}_x$ ) that might reduce the total cost of conversion and disposition
- Catalytic chemistry of DU oxides
- DU alloy science

A number of these topics could serve as vehicles for investigating DU-related material disposition paths where limited investment to establish feasibility is indicated.

## 6. DOE'S APPROACH TO DU MATERIAL DISPOSITION

Previous sections of this Roadmap contain consensus recommendations concerning the DU disposition paths that might be worth pursuing and the barrier-reduction activities that would be required for deployment to be possible. On this basis, the goal of DOE's DU material disposition activities will be to pursue a prudent contingency approach to further research concerning disposition of DU materials. Specifically, DOE plans on making investments in barrier-reduction activities supporting the most promising paths involving beneficial use of DU materials while also making appropriate investments to ensure that economical disposal alternatives are reliably available and compatible with the potential uses. Those beneficial uses that are determined to be feasible, worthwhile, and acceptable will be implemented using the products of the DUF<sub>6</sub> conversion plants plus any acceptable DOE surplus inventory of other forms of DU. The DU-related materials that do not have clear beneficial uses will be destined for disposal. DOE's approach is further described as five objectives in the following paragraphs.

*First*, support a broad spectrum of investments to reduce barriers along paths related to nuclear material storage and/or disposal that have relatively low technical risk and use large quantities of DU in radiologically regulated areas. These paths have technical or institutional barriers that must be overcome before they can be fully evaluated or deployed. Example areas of investment are:

- Heavy Concrete. To support economical manufacture of radiation shielding and spent fuel/nuclear waste storage casks (silos) from high-density concrete containing DU, DOE's barrier-reduction activities will concentrate on characterizing and improving the potential for use of such products followed by disposal at the end of their useful life as LLW packages.
- Package Fill. DOE will support activities focusing on characterizing the impacts of using DU oxide fill particles inside spent fuel disposal packages on the design and performance of the engineered barrier system to provide the basis for a subsequent decision as to whether such use is justified and, if so, how to license such use.

*Second*, make targeted investments to reduce barriers for a number of paths where there is potential to use substantial amounts of DU-related materials but where the uses are more speculative or simply require a small investment before the path could be followed. For example:

- Use of DU in Non-Governmental Applications. This includes potential use of DU in forklift counterweights, catalysts, aluminum-refining electrodes, metal alloys, oil well penetrators, and drilling collars. These paths share a common barrier because they involve use of DU in industrial or consumer settings.



- Invert Containing DU. This path involves putting DU oxide into the cells of steel invert used to level the rounded bottom of tunnels in the potential repository to provide ballast and the possibility that the DU might improve the performance of the potential repository.
- Characterization of DOE's Non-DUF<sub>6</sub> Inventory. Potential paths for disposition of DOE's non-DUF<sub>6</sub> inventory are to sell the inventory for less than the cost of disposal, or, if this is not possible or desirable, to dispose of this material as LLW.
- Facilitating Use of Intact Cylinders as LLW Packages. The preferred path for disposition of DUF<sub>6</sub> storage cylinders is to use them intact as LLW packages by cutting an opening, loading them with LLW, welding a plug into the opening, and disposing of the package at a LLW disposal facility.

*Third*, make appropriate investments to ensure that there are no barriers to following an optimal path for long-term storage or direct disposal of the DU conversion products that are not beneficially used or to disposal of DU-bearing products at the end of their useful lives. For example:

- DU Disposal. To ensure availability of a reliable and economic disposal path for all DU-associated materials, DOE will undertake targeted technical and institutional activities.
- Long-Term Storage. Long-term storage of some DU may be desired for the purpose of maintaining a national resource reserve or necessitated by impediments to other disposition paths.

*Fourth*, invest in basic and mission-directed research that is related to beneficial use of DU-related materials. These investigations are necessary to expand our knowledge of the basic properties of uranium that are necessary to provide a basis for evaluating the feasibility, impacts, and economics of potential DU disposition paths and to identify new beneficial uses of the DU conversion products. These research areas include the following:

- Long-term interaction of DU oxides, metal, and tetrafluoride with container materials and the environment to support paths concerning package fill and invert for the potential repository, disposal, and long-term storage
- Alteration of uranium oxides and tetrafluoride in aqueous or cement media over the long term to support paths concerning high-density concrete, package fill, and invert for the potential repository, and disposal
- Processes for producing higher-value fluorine compounds that might reduce the cost of DUF<sub>6</sub> disposition
- Direct conversion of DUF<sub>6</sub> to useful products (e.g., DU conversion producing USi<sub>x</sub>) that might reduce the total cost of conversion and disposition
- Catalytic chemistry of DU oxides
- DU alloy science

Proposal solicitations should be structured to encourage new concepts that hold promise to economically use significant amounts of DU.

*Fifth*, invest in system analysis and support activities that benefit multiple aspects of DU material disposition.

- Establishing Institutional Roles and Responsibilities. DOE will facilitate establishment of the roles and responsibilities for funding, regulation, market development, incentive structures, and DU-related products and the interfaces between the elements having these responsibilities. This framework is necessary to effectively coordinate DU disposition activities that involve multiple DOE programs, regulators, and the private sector. Efforts to foster public acceptance of DU-bearing products will also be supported.
- System Optimization. DOE will characterize a reference system for DU-related material disposition against which alternative paths can be compared. This will then provide the basis for analyses to optimize the system.

## 7. REFERENCES

- DNFSB 1995 Defense Nuclear Facilities Safety Board, Recommendation 95-1 to the Secretary of Energy (May 5, 1995).
- DOE 1997 DOE, *Draft Handbook for Controlling Release for Reuse or Recycle of Property Containing Residual Radioactive Material*, Interim Guide—For Interim Use and Comment (June 1997).
- DOE 1999a U.S. Department of Energy, *Final Programmatic Environmental Impact Statement for Alternative Strategies for Long-Term Management and Use of Depleted Uranium Hexafluoride*, DOE/EIS-0269 (April 1999).
- DOE 1999b U.S. Department of Energy, *Final Plan for the Conversion of Depleted Uranium Hexafluoride, as Required by Public Law 105-204* (July 1999).
- DOE 1999c U.S. Department of Energy, *DOE Research and Development Portfolio—Environmental Quality*, Vol. 3 of 5 (September 1999).
- DOE/USEC 1998a U.S. Department of Energy and United States Enrichment Corporation, “Memorandum of Agreement Relating to Depleted Uranium Generated Prior to Privatization Date” (May 18, 1998).
- DOE/USEC 1998b U.S. Department of Energy and United States Enrichment Corporation, “Memorandum of Agreement Relating to Depleted Uranium” (June 30, 1998).
- Dubrin 1997 J. W. Dubrin et al., *The Engineering Analysis Report for the Long-Term Management of Depleted Uranium Hexafluoride*, UCRL-AR-124080, Vols. 1 and 2 (May 1997).
- Elayat 1997 H. Elayat, J. Zoller, and L. Szytel, *Cost Analysis Report for the Long-Term Management of Depleted Uranium Hexafluoride*, UCRL-AR-127650, Lawrence Livermore National Laboratory, Livermore, California (May 1997).
- Forsberg 1995 C. W. Forsberg, R. B. Pope, R. C. Ashline, M.D. Dehart, K. W. Childs, and J. S. Tang, *DUSCOBS—A Depleted-Uranium Silicate Backfill for Transport, Storage, and Disposal of Spent Nuclear Fuel*, ORNL/TM-13045, Oak Ridge National Laboratory (November 30, 1995).
- FR 1999 “Record of Decision for Long-Term Management and Use of Depleted Uranium Hexafluoride,” *Fed. Reg.* 43358, 64(153), August 10, 1999.
- FR 2000 “Record of Decision for the Department of Energy's Waste Management Program: Treatment and Disposal of Low-Level Waste and Mixed Low-Level Waste; Amendment of the Record of Decision for the Nevada Test Site,” *Fed. Reg.* 10061, 65(38), February 25, 2000.

Hightower 2000 J. R. Hightower, et al., *Strategy for Characterizing Transuranics and Technetium Contamination in Depleted UF<sub>6</sub> Cylinders*, ORNL/TM-2000/242, Oak Ridge National Laboratory (October 2000).

Manuel 1999 Janet Manuel, Bechtel Jacobs Corporation, personal communication (July 1999).

Michelhaugh 1999 Rick Michelhaugh, Lockheed Martin Energy Research, personal communication (June 1999).

NRC 1994 U.S. Nuclear Regulatory Commission, *Final Environmental Impact Statement for the Construction and Operation of Claiborne Enrichment Center, Homer, Louisiana*, Docket No. 70-3070, NUREG-1484, Volumes 1 and 2, Office of Nuclear Material Safety and Safeguards (August 1994).

NRC 1999 “Release of Solid Materials at Licensed Facilities: Issues Paper, Scoping Process for Environmental Issues, and Notice of Public Meetings,” *Fed. Reg.* 35090, 64(125), June 30, 1999.

PL 1954 Atomic Energy Act of 1954, as amended, Public Law 83-703, 68 Stat. 919 (August 30, 1954).

RCRA 1976 Resource Conservation and Recovery Act, Public Law 89-272 (October 20, 1965), 79 Stat. 997, as amended by Public Law 94-580, 90 Stat. 2795 (October 21, 1976).

USEC 1999 *The UF<sub>6</sub> Manual, Good Handling Practices for Uranium Hexafluoride*, USEC-651, Rev. 8 (January 1999).

**APPENDIX A:**  
**ANALYSIS OF CANDIDATE DISPOSITION PATHS**

**Table A.1. Analysis of candidate disposition paths for direct disposal of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate disposal path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>DUF<sub>6</sub></b>						
LLW disposal <u>Reference path for most DU</u>	Yes; meeting specific WAC and lowering cost	Up to 100%	Relatively low cost	None unless coupled with long-term storage	Requires discussion with disposal site to establish optimal approach and costs	Potential environmental impacts and EIS
Mined cavity disposal	Yes; siting, legislation, and potential licensing	Up to 100%	Relatively high cost in a new facility, depending on the form of the DU; moderate cost in an existing facility	None	Requires siting, legislation, and potential licensing a new disposal facility	Potential environmental impacts and EIS
Salt mine disposal as DUF <sub>6</sub>	Yes; interaction with salt, siting, legislation, and potential licensing	Up to 100%	Same as for a mined cavity, but the conversion cost would be avoided	Interaction of DUF <sub>6</sub> with halite and brine	Requires siting, legislation, and potential licensing for a new disposal facility	Potential environmental impacts and EIS
Subsurface engineered vault disposal	Yes; siting, legislation, and potential licensing	Up to 100%	Moderately low cost depending on the form of the DU	None	Requires siting, legislation, and potential licensing for a new disposal facility	Potential environmental impacts and EIS
<b>DU other than DUF<sub>6</sub></b>						
Beneficial use without further processing	Yes; some additional characterization of impurities	Up to 100%	Limited amount likely to have value to the private sector, and the avoided cost of disposal can be used as an incentive	Need to characterize potentially troublesome chemical and radioactive impurities	Impact of potential trace transuranic and fission-product impurities on use and liability	Potential environmental impacts and EIS
LLW disposal <u>Reference path</u>	Yes; additional characterization of impurities	Up to 100%	Selling material is likely to be more economical to avoid disposal cost, but disposal may be most economical path for impure material	Need to characterize potentially troublesome chemical and radioactive impurities	None	Potential environmental impacts and EIS
<b>Fluorine products</b>						
Dispose of fluorine	No; disposal of contaminated Ca or Mg difluoride is existing commercial practice	Up to 100%	Very small cost as long as DU concentration permits disposal in sanitary landfill	None	May be contrary to policy concerning release of volumetrically contaminated material to be established by DOE or NRC as applicable	Potential environmental impacts and EIS
<b>DUF<sub>6</sub> storage cylinders</b>						
Dispose of metals <u>Reference path</u>	No	Up to 100%	Moderate net cost; this is reference case	None	None	Potential environmental impacts and EIS

Table A.2. Analysis of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>

Candidate beneficial use path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>DUF<sub>6</sub></b>						
<b>DU Matrix and Shielding Products</b>						
Cement-Lock™	Yes; no experience with DU materials	Up to 100%	Good potential for conversion and disposal cost to be less than cost of this complex, high-temperature process	Need proof-of-concept experiments, characterization of products, and measurement of properties	Use of a substantial amount of this product requires overcoming barriers to use in unregulated areas	Potential environmental impacts and EIS
DU metal shielding	No; existing practice	Up to 50%	DU metal is not presently cost-effective; unclear whether deferred disposal cost is enough to compensate	Mature technology, although there may be some benefits to establishing an American Society for Mechanical Engineers code section on DU metal so it could be used as a structural component	None	None
DUCRETET™	Yes; technical data, manufacturing techniques, user acceptance, and licensing	To be determined	Approximately equal to standard concrete storage silos on a system-wide basis, but with the possibility of further reductions after more development	Need to characterize properties, validate manufacturing techniques and long-term performance, produce demonstration casks, and optimize designs	Identify specific products and clarify NRC licensing requirements; gain acceptance from purchasers and manufacturers	Environmental impacts. Can reduce dose during loading in situations that have previously required on-site transfer casks. Timing relative to license application for the potential repository
DUPoly	Yes; made at laboratory scale. Prototype container has been designed.	Up to 100%	Costs uncertain; high product loading results in lower overall costs	Need to clarify polyethylene stability issues; need economic modeling and costs estimates for shielding	Organic materials are not acceptable in the potential repository	Environmental impacts. Can reduce dose during loading in situations that have previously required on-site transfer casks
PYRUC	Yes; technology is in proposal stage	Up to 100%	Low potential due to high costs from complicated and expensive processing technologies	Need proof-of-concept experiments, optimization of sol-gel methods, fabrication of composites and product characterization	Identify specific products and clarify NRC licensing requirements; gain acceptance from purchasers and manufacturers	Environmental impacts. Can reduce dose during loading in situations that have previously required on-site transfer casks
Uranium silicide	Yes; technology is conceptual	Up to 100%	Some potential for eliminated conversion cost and deferred disposal cost to be less than cost of high-temperature process	Need proof-of-concept experiments, economic modeling, oxidation experiments, product characterization	Identify specific products and clarify NRC licensing requirements; gain acceptance from purchasers and manufacturers	Environmental impacts. Can reduce dose during loading in situations that have previously required on-site transfer casks

**Table A.2. Analysis of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate beneficial use path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>DUF<sub>6</sub></b>						
<b>Proposed Applications in the Potential Repository</b>						
Backfill component	Yes; manufacture and use of DU oxide as a backfill component in the potential repository are not existing practice	To be determined	Trade-off of cost reduction from use constituting disposal against cost of storage until use occurs unlikely to be favorable	Need to understand how DU oxide interacts with groundwater and WP	Obtain approval by DOE-RW and regulators for this application. Decision on whether to use backfill may not be made for decades	Potential environmental impacts and EIS. May lower long-term risk at the cost of some increased risk during emplacement
Invert	Yes; manufacture and use of DU oxide for invert in the potential repository are not existing practice	To be determined	Constitutes disposal; net cost of emplacement in the potential repository likely to be higher than disposal as LLW	Need to understand interactions with steel plate used in invert, groundwater, and WP	Obtain approval by DOE-RW and regulators for this application. May need additional NEPA documentation and licensing actions because it is out of sequence with design and licensing process	Timing an issue relative to the license application for the potential repository. May lower long-term risk at the cost of some increased risk during emplacement and operations
Package fill	Yes; manufacture and use of DU oxide to fill packages in the potential repository are not existing practice	To be determined	Near-term cost addition for potential improvement; more speculative potential to eliminate other engineered barriers in the late 22nd century	Large-scale prototype work on WP fill in Canada; need to characterize fill, develop insertion technology, determine impacts on package and the performance of the potential repository	Obtain approval by DOE-RW and regulators for this application. May need additional NEPA documentation and licensing actions because it is out of sequence with design and licensing process	Timing an issue relative to the license application for the potential repository. May lower long-term risk at the cost of some increased risk during filling
<b>Fissile Material Disposition Applications</b>						
Ceramics for plutonium disposition	No; development under way by DOE-MD	<0.1%	Could use available DU dioxide or trioxide at low cost while eliminating disposal cost	Only bench-top demonstration so far	Unclear as to whether small cans of Pu/DU oxide ceramic in a large canister of HLW will provide adequate protection	None
Dilution of HEU(F)	No; needed studies are being supported by DOE-MD	1–5%	Eliminates conversion and disposal cost, but for a small amount	None	None	None
MOX fuel for plutonium disposition	No; development under way by DOE-MD	<1%	Eliminates disposal cost, but for small amount	Some development on pit alloying constituents	About 130 cylinders of DUF <sub>6</sub> have been set aside at Portsmouth for this purpose	None



**Table A.2. Analysis of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate beneficial use path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>DUF<sub>6</sub></b>						
<b>Fuel Cycle Applications</b>						
AVLIS reenrichment	Yes; AVLIS has only been demonstrated at laboratory scale	1–5%	Not competitive until natural uranium price increases slightly for 0.4% DU; much more for bulk of DU that has lower enrichments	Complete demonstration of AVLIS technology	AVLIS program has been discontinued	Potential environmental impacts and EIS
Fast reactor fuel	No; existing technology	Up to 100%	Fast reactors not economic without a major increase in uranium costs	None	Presidential Decision Directive 13 prohibits reprocessing; public acceptance of plutonium recycle	Potential environmental impacts and EIS. Fast reactors constitute a long-term, secure energy supply
SILEX reenrichment	No; technology not proven but being supported by USEC	1–5%	Not competitive until natural uranium price increases slightly for 0.4% DU; much more for bulk of DU that has lower enrichments	Can use DUF <sub>6</sub> without conversion, but still in research stage	SILEX is being supported using industry funding. Not a government program, and licensing is expected to be straightforward	Potential environmental impacts and EIS
<b>Commercial Applications</b>						
Aluminum-refining electrodes	Yes; use of DU in this application is not existing practice	Up to 100%	Estimates indicate that DU oxide electrodes could increase operating efficiency equivalent to DU having worth much greater than its cost	Virtually no work on this application; need solubility and degradation rate of electrodes and information on efficiency	Regulatory framework for use of electrodes and products has not been defined and may not be conducive to use	Potential environmental impacts and EIS
Catalyst for fluid cracking and to promote oxidation	Yes; use of DU as a commercial catalyst is not established practice	<1%	Unknown	Performance and degradation rates	Regulatory framework for use of electrodes and products has not been defined and may not be conducive to use	Potential environmental impacts and EIS
Catalyst for automotive exhaust	Yes; use of DU in automobiles is not established practice	Up to 50%	Unknown	Performance and degradation rates	Regulatory framework for use of electrodes and products has not been defined and may not be conducive to use	Potential environmental impacts and EIS
Catalyst for fuel cells and steam reforming	Yes; use of DU as a commercial catalyst is not established practice	Up to 50%	Unknown	Performance and degradation rates	Regulatory framework for use of electrodes and products has not been defined and may not be conducive to use	Potential environmental impacts and EIS

**Table A.2. Analysis of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate beneficial use path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>DUF<sub>6</sub></b>						
<b>Commercial Applications</b>						
Heavy-lifting-vehicle counterweights and high-traction devices	Yes; manufacture of equipment with DU components is not established practice	Up to 100%	DU products are more costly; however, warehouses can be less costly and locomotives with DU wheels can haul more payload	Prototyping and demonstration needed	Regulatory worker exposure, ability to maintain control of counterweights	Potential environmental impacts and EIS. Can reduce forklift accidents, of which 90,000 occur each year with 85 fatalities
Oil well penetrators and drilling collars	Yes; DU has been used down-well in the petroleum industry, but market impediments are apparent	Up to 50%	Improved performance would have to compensate for the additional cost to produce DU metal	None	Regulatory framework for use of electrodes and products has not been defined and may not be conducive to use	Potential environmental impacts and EIS. Density of DU metal may improve drilling efficiency; increased dose to oil well workers
<b>National Resource Reserve</b>						
Long-term storage <u>Reference path for a portion of the DU</u>	Yes; no previous experience with low-maintenance storage of relatively large quantities for decades	<10% of the inventory in a variety of forms having unique characteristics	A significant cost that may be minimized if potential exhumation of DU LLW is acceptable for this purpose	Need to develop concept and package specifications that minimize cost; establishing detailed strategic reserve requirements	Public, local governments, and some regulators object to long-term storage without definable use in sight	Potential environmental impacts and EIS
<b>DU other than DUF<sub>6</sub></b>						
Reuse as is	Yes; some additional characterization of impurities	Up to 100%	Limited amount likely to have value to the private sector, and the avoided cost of disposal can be used as an incentive	Need to characterize potentially troublesome chemical and radioactive impurities	Potential impact of trace transuranic and fission-product impurities on use and liability	Potential environmental impacts and EIS
Reuse with further processing	Yes; technology exists but operational DOE facilities to further process the DU into useful forms do not	Up to 100%	Selling unprocessed DU to private sector likely to be more economical than processing this small amount of material	Would need to reactivate or establish facilities for processing; need to characterize potentially troublesome chemical and radioactive impurities	None	Potential environmental impacts and EIS
Disposal	Yes; additional characterization of impurities	Up to 100%	Selling material is likely to be more economical to avoid disposal cost, but this may be most economical path for impure material	Need to characterize potentially troublesome chemical and radioactive impurities	None	Potential environmental impacts and EIS

Table A.2. Analysis of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>

Candidate beneficial use path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>Fluorine products</b>						
Anhydrous and aqueous HF for industrial use  <u>Reference path</u>	No; these materials are produced and used commercially	Up to 100%	HF from uranium defluorination being sold or reused now in industry—could supply about 5% of U.S. demand	None	May be contrary to policy concerning release of volumetrically contaminated material to be established by DOE or NRC as applicable	Potential environmental impacts and EIS
Calcium difluoride for industrial use	No; these materials are produced and used commercially	Up to 100%	HF from uranium conversion being sold or reused now in industry—could supply about 5% of U.S. demand	None; easier to store and transport than HF	May be contrary to policy concerning release of volumetrically contaminated material to be established by DOE or NRC as applicable	Potential environmental impacts and EIS
Elemental fluorine for industrial use	No; production of elemental fluorine from HF is established technology	Up to 50%; use is limited because F <sub>2</sub> is not transportable and only used when absolutely required	Could command higher price than AHF; this source (about 13,000 MT/y) is two-thirds of the worldwide demand	None	Regulations limit the amount that can be transported; significant quantities must be used at production site. May be contrary to policy concerning release of volumetrically contaminated material to be established by DOE or NRC as applicable	Potential environmental impacts and EIS
High-value fluorine compounds for industrial use	Yes; production of F-containing compounds other than HF and CaF <sub>2</sub> is not existing practice	Up to 100%	High-value fluorine compounds sell for much more than HF but cost more to make; overproduction could lower prices and impact the private sector	Develop technology to produce a suite of high-value compounds		Potential environmental impacts and EIS

**Table A.2. Analysis of candidate disposition paths for beneficial use of products from DUF<sub>6</sub> conversion and DU other than DUF<sub>6</sub>**

Candidate beneficial use path	Is additional barrier reduction needed?	Inventory utilization	Net cost savings	Technical maturity and barriers	Institutional, legal, regulatory, and policy challenges and barriers	Other impacts
<b>DUF<sub>6</sub> storage cylinders</b>						
Decontaminate and recycle metals <sup>a</sup>	Yes; related to potential use of slightly contaminated metals in nongovernmental applications	Up to 100%	About the same as direct disposal cost. Monel and nickel are being recycled, but cost of smelting steel outweighs value plus avoided disposal cost	None	Release is currently prohibited pending further NRC and DOE decisions	Potential environmental impacts and EIS
Intact cylinders as LLW disposal packages	Yes; procedures for cylinder reuse and loading	Approaching 100%. Some cylinders are in poor shape and would require disposal	Moderate net cost savings from avoided purchase of LLW packages plus avoided cylinders disposal cost	None	Need procedures to qualify, load, and seal cylinders	Potential environmental impacts and EIS
Reuse intact cylinders for newly generated DUF <sub>6</sub>	Yes; procedures for cylinder reuse and loading needed	Small volume (<1%) of existing cylinder inventory utilized	Relatively low net cost savings because of small volume	None	Need procedures to qualify, load, and seal cylinders	None
Refabricate metal for use in regulated areas <sup>a</sup>	No; refabrication of slightly contaminated metal is established practice	Up to 100% as LLW packages	Relatively low net cost savings compared with direct disposal from avoided purchases plus avoided cylinder disposal, less smelting and refabrication costs	None	None	Potential environmental impacts and EIS

<sup>a</sup>Reuse of storage cylinders may be subject to the Secretary of Energy's memorandum of July 13, 2000, suspending unrestricted release of contaminated metals from radiation areas.

**APPENDIX B:**

**DU MATERIALS USE ROADMAP: RESPONSE TO PUBLIC COMMENTS**

## Appendix B

### DU MATERIALS USE ROADMAP: RESPONSE TO PUBLIC COMMENTS

The following individuals and organizations provided comments concerning the September 1, 2000, draft DUF<sub>6</sub> Materials Use Roadmap:

1. U.S. Nuclear Regulatory Commission (Eric J. Leeds)
2. Alex Murray, engineer and private citizen
3. University of Kentucky—Paducah (William Murphy)
4. Brookhaven National Laboratory (Paul Kalp)
5. Bechtel, Inc. (Kenneth M. Cooke)
6. Oak Ridge National Laboratory (Charles Forsberg)
7. MACTEC, Inc. (Larry Harmon)
8. Duke Engineering Services (R. G. Morgan)
9. Foster Wheeler Environmental Corp. (Ron Izatt)
10. CAMECO Corp. (A. J. Oliver)
11. Jacobs Engineering Group (Ken Cruits Shank)
12. BNFL, Inc. (A. Joiner)
13. Ohio EPA (D. Goodman)
14. Envirocare of Utah (C. A. Judd)
15. Tennessee Department of Environment and Conservation (D. Rector)
16. Lawrence Livermore National Laboratory (S. Hargrove)

It is difficult to categorize and summarize the 16 sets of comments. Commentators clearly read the draft Roadmap carefully and put considerable effort into their review. Their comments improved the final document. DOE has made numerous changes to expand and clarify the specific text of the report; however, the overall conclusions of the report did not change. The following are pervasive or significant public comments and the DOE responses:

#### Comment 1

Several of the private sector companies were confused about the relationship between the DUF<sub>6</sub> Materials Use Roadmap and the DOE request for proposals (RFP) for design, construction, and operation of DUF<sub>6</sub> conversion facilities. The Roadmap was released for public comment prior to issuance of the final RFP, and some prospective RFP respondents believed that the Roadmap was intended to provide direction and criteria for preparing responses to the RFP.

## **Response**

The RFP itself answered many of the questions, once it was released. Specifically, the RFP states that R&D is not part of the RFP solicitation. Accordingly, to avoid confusion, the final version of the Roadmap, including the response to public comments, was not issued until after DOE had received responses to the RFP for design, construction, and operation of DUF<sub>6</sub> conversion facilities.

## **Comment 2**

Several organizations who filed comments seemed to misunderstand the purpose and scope of the Roadmap. Many comments suggested a belief by the commentator that the Roadmap was intended to provide a detailed assessment of DU disposition path alternatives.

## **Response**

The Roadmap provides guidance for R&D. The Roadmap is not being used by DOE to decide between alternatives and proposals for DU conversion, potential applications, and disposal. Rather, the Roadmap summarizes DOE's surplus DU inventory, specifies alternative paths that could result in the disposition of DU, evaluates these paths, and recommends a portfolio of activities required to overcome barriers along the paths. Accordingly, Sects. 3 through 5 in the Roadmap contain consensus recommendations concerning the DU disposition paths that might be worth pursuing and the barrier-reduction activities that would be required for deployment to be possible.

## **Comment 3**

Three commentators suggested that the Roadmap be modified to provide a more-detailed description of a baseline or reference disposition path to provide objective criteria against which other potential paths could be compared regarding such factors as determination of utilization, economics, technical maturity, and health and safety impacts.

## **Response**

As is stated in Sect. 3 of the Roadmap, the reference path for disposition of DU (i.e., baseline against which other candidate paths are compared) was assumed to be as follows: direct disposal of non-DUF<sub>6</sub> forms of DU as low-level radioactive waste; direct disposal of DUF<sub>6</sub> conversion products as low-level radioactive waste (except fluorine-bearing products such as hydrofluoric acid or AHF); and direct

disposal of emptied  $\text{DUF}_6$  storage cylinders as low-level radioactive waste. The purpose of the Roadmap, which is a record of the roadmapping process, is to explore potential uses for  $\text{DUF}_6$  conversion products and to identify areas (e.g., institutional, technical) where further development work is needed. It is not the function of the Roadmap to provide a detailed assessment of DU disposition path alternatives. Notwithstanding, Sect. 5.3.2 of the Roadmap acknowledges the need for improved characterization of a reference disposition scenario for DU-related material against which alternative disposition scenarios can be compared during future DOE decision-making efforts.

#### **Comment 4**

Two commentators suggested that the Roadmap be modified to indicate that  $\text{DUF}_4$  is not a likely candidate for disposal in large quantities. Several others suggested that the Roadmap explain more fully why  $\text{DUF}_4$  is listed as a potential form for disposal since the PEIS and its corresponding Record of Decision indicate that the  $\text{DUF}_6$  inventory will be converted to either DU oxide or DU metal, or a combination of both.

#### **Response**

Based on preliminary investigations, DOE believes that site-specific processes for evaluating and demonstrating acceptability of LLW streams at the NTS and the Envirocare of Utah site could confirm  $\text{DUF}_4$  as an acceptable waste form for near-surface disposal at these particular arid sites under certain conditions. For this reason, the roadmapping process did not exclude consideration of  $\text{DUF}_4$  as a possible form for disposal as LLW.

#### **Comment 5**

A few commentators expressed disappointment in the roadmapping process, primarily because their organizations were not invited to participate. One organization suggested that the Roadmap provide a better description and more information about the process.



## **Response**

As Sect. 4.2 of the Roadmap explains, after substantial preparation, a diverse group of experts from four national laboratories, consultants, and three DOE Program Offices were convened in a workshop. The roadmapping workshop analyzed and evaluated potential DU disposition paths and made recommendations about funding of barrier-reduction activities. At the workshop, the status of technologies supporting various DU disposition paths were discussed, barriers to implementing each path were identified, and recommendations about barrier-reduction activities were formulated. The Roadmap documents the results of the workshop, as well as DOE's expected approach to making investments in barrier-reduction activities. DOE believes that Sect. 4.2 of the Roadmap adequately explains the roadmapping process, which is unlike peer review in that roadmapping does not involve review of a completed work prepared by a different entity. The commercial sector was excluded because of potential procurement issues. Universities were not included because they were not familiar with the context of DU disposition and had limited resources to conduct the workshop. They also had the foreknowledge that future basic science activities would be geared to universities and would be part of follow-on R&D (which has since occurred). Further, this opportunity for comment provides the means to consider universities and private-sector views and data.

## **Comment 6**

Some comments suggested that the Roadmap focuses too much on costs and does not adequately discuss safety and environmental compliance aspects of the alternative DU disposition paths.

## **Response**

DOE disagrees that the Roadmap inappropriately emphasizes costs over safety and regulatory compliance. The Roadmap assumes that full compliance with existing applicable safety and environmental regulations would be mandatory for all DU disposition paths. If a path (described in Sect. 3 of the Roadmap) involves activities for which existing safety or environmental regulations are not fully appropriate, or for which the ability to comply with existing regulations is questionable due to the technological characteristics of such activities, then for the purpose of the evaluation reported in Sect. 4 of the Roadmap, that path was identified as having regulatory barriers, which could reduce the likelihood for it to be useful in providing a final disposition for DU. Section 5 then identifies barrier-reduction activities that could be required to make the path viable. Since all paths are assumed to achieve at least the level of safety and environmental protection mandated by applicable regulations, the Roadmap appropriately concentrates on identifying paths that are most likely to do so in a manner that reduces overall cost to the government.

### **Comment 7**

Three commentators believed that uses of DU in shielding for stored SNF and in applications related to the HLW repository should be emphasized in the Roadmap. For example, one commentator pointed out that DU can be converted and fabricated into shielding materials suitable for SNF applications, including storage, transportation, and disposal, and that DU may also have backfill, shielding, or other applications in the repository (e.g., macroscopic criticality poison). The repository could even be considered as a potential disposal site for DU.

### **Response**

As the tables in Sect. 5 of the Roadmap illustrate, a number of barrier-reduction activities are expected to be necessary before DU could be fully evaluated and subsequently deployed as SNF shielding material or in repository applications. Section 6 of the Roadmap provides an approach that DOE plans to use in making investments in such barrier-reduction activities. Specifically, Sect. 6 describes a five-point plan whereby DOE would make investments in barrier-reduction activities supporting the most promising paths involving beneficial use of DU materials while also making appropriate investments to ensure that economical disposal alternatives are reliably available and compatible with the potential uses. Several of the areas of investment under the five-point plan are intended to reduce technical and institutional barriers surrounding the use of DU in SNF shielding materials and in repository applications.

### **Comment 8**

One commentator suggested that the Roadmap acknowledge that disposition of significant quantities of DU by disposal will likely entail additional regulator review.

### **Response**

Section 2.2 in the Roadmap summarizes existing regulations because such regulations provide the basis used during the roadmapping process for determining whether regulatory barriers exist along each path. The requirements of future regulations, unless imminent, were not themselves considered during the roadmapping process. DOE understands that NRC and other agencies may independently initiate rulemaking proceedings as a result of DOE's efforts to implement paths to DU disposition. Notwithstanding, DOE believes it would not be appropriate to modify the Roadmap at this time, as suggested by this comment, because the requirements imposed by such future independent rulemaking proceedings by non-DOE federal agencies, over which DOE has no control, were not themselves considered as barriers during the roadmapping process. Instead, the Roadmap identifies the existence of regulatory issues as a potential barrier for certain paths.

## Comment 9

Among other comments, the NRC commented as follows:

On page 8, Sect. 2.2.6 makes reference to statements made in NUREG-1484 related to disposal of DU at LLW facilities. Specifically, the report states that “NRC has determined that near-surface disposal facilities in wet locations are extremely unlikely to successfully make such a demonstration if they accept  $\text{DU}_3\text{O}_8$ .” This statement appears to indicate a generic conclusion that is out of context. NUREG-1484 is the EIS for the proposed Claiborne Enrichment Center project. As part of the assessment of potential environmental impacts of the project, the disposal of DU was evaluated assuming a generic LLW disposal facility in the humid southeast. The EIS concluded that it was likely that deep disposal would be required to dispose of DU wastes. The analysis was not done using characteristics of a particular site. The Roadmap discussion may wish to highlight the finer points from the EIS analysis that may have wider applicability; for example, the use of an oxide DU form, a unique disposal facility with better confinement, etc. In addition, an arid site will change the performance assessment and dose results. However, the magnitude of the dose from the generic assessment exceeded the regulatory limits by a significant margin. It may be appropriate to state that disposal of all or most of DU at a single LLW disposal facility may not comply with a site's WAC. As noted in Table A.1, additional discussion with specific disposal facilities may be required to establish an optimal disposal approach.

## Response

The last paragraph of Sect. 2.2.6 (page 8) will be modified to read as follows:

DOE M 435.1-1 prohibits disposal of DOE-generated LLW, including DU or materials containing residual DU (e.g., calcium fluoride and empty  $\text{DUF}_6$  cylinders), in non-DOE LLW disposal facilities, unless the responsible DOE Field Element Manager approves an exemption for use of non-DOE facilities based, in part, on a determination that DOE-controlled disposal capabilities are not practical or cost-effective. If disposal in an NRC- or NRC Agreement State-licensed LLW disposal facility is approved, such facilities are subject to 10 CFR Part 61, “Licensing Requirements for Land Disposal of Radioactive Wastes,” or compatible state regulations. Title 10 CFR Part 61 requires a demonstration of compliance with specified performance objectives and technical standards. Title 10 CFR Part 61 also requires facilities to establish waste characteristic limitations that could preclude disposal of some chemical forms (e.g.,  $\text{DUF}_6$ ) and some physical forms (e.g., finely divided or powdered metal) of DU without special packaging and/or stabilization.

In addition, the following paragraph will be added to the end of Sect. 2.2.6:

In 1995, during the scoping process for the PEIS concerning long-term management of  $\text{DUF}_6$ , the NRC staff expressed its opinion that  $\text{DU}_3\text{O}_8$  is a likely chemical form for DU disposal. They also advised DOE that, although  $\text{DU}_3\text{O}_8$  could be disposed of in limited quantities in conventional near-surface disposal facilities, large quantities (such as would be derived from the nation's enrichment tailings inventory) suggest the possible need for a unique disposal facility, such as a mined cavity or an exhausted uranium mine [NRC, Letter from NRC (R. Bernero) to DOE (C. Bradley, Jr.), January 3, 1995].

**Comment 10**

One commentator recommended adding DU dioxide (DUO<sub>2</sub>)-steel cermets to the list of possible uses for DU and as a possible waste form. Cermets are ceramic metallic composites that contain a ceramic (DUO<sub>2</sub>) embedded in a metal (steel). Typically, the cermet is produced as a sandwich structure with clean steel on both sides of the cermet. DU cermets could be used for (1) general radiation shielding applications and (2) as a replacement for selected steel components in a waste repository package.

**Response**

The Roadmap is a report of the results of the workshop on potential DU uses. The use of DU dioxide-steel cermets is a new idea developed after that meeting. DU dioxide-steel cermets will be viewed as research on a new concept in the follow-on R&D program; other concepts will also be expected from investments in science related to DU disposition.