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#### OAK RIDGE NATIONAL LABORATORY

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## Characterization of the Radioactive Sludge from the ORNL MVST Waste Tanks

J. M. Keller J. M. Giaquinto



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Chemical and Analytical Sciences Division

# Characterization of the Radioactive Sludge from the ORNL MVST Waste Tanks

J. M. Keller J. M. Giaquinto

September 2001

Prepared by the
OAK RIDGE NATIONAL LABORATORY
Oak Ridge, Tennessee 3783 1
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#### ABBREVIATIONS AND ACRONYMS

ALARA As Low As Reasonably Achievable

CAO Carlsbad Area Office

CASD Chemical and Analytical Sciences Division

CVAA Cold Vapor Atomic Absorption

DOO Data Quality Objective

EPA Environmental Protection Agency

GC/MS Gas Chromatography/Mass Spectrometry

GC Gas Chromatography

GFAA Graphite Furnace Atomic Absorption

IC Ion Chromatography

ICP Inductively Coupled Plasma

ICP-AES Inductively Coupled Plasma - Atomic Emission Spectroscopy

ICP-MS Inductively Coupled Plasma - Mass Spectrometry

IDL Instrument Detection Limit
LCS Laboratory Control Sample
LLLW Liquid Low-Level Waste
MDL Method Detection Limit

MS Matrix Spike

MSD Matrix Spike Duplicate

MVST Melton Valley Storage Tanks

NTS Nevada Test Site

ORNL Oak Ridge National Laboratory

QA Quality Assurance

QAPjP Quality Assurance Project Plan QAPP Quality Assurance Program Plan

QC Quality Control

RCRA Resource Conservation and Recovery Act
RH-TRU Remote Handled Transuranic Waste

RMAL Radioactive Materials Analytical Laboratory (Building 2026)

TC Total Carbon

TCLP Toxicity Characteristic Leaching Procedure

TDS Total Dissolved Solids

TIC Total Inorganic Carbon or Tentatively Identified Compounds

TIMS Thermal Ionization Mass Spectrometry

TOC Total Organic Carbon

TRU Transuranic

TWCP Transuranic Waste Characterization Program

WAC Waste Acceptance Criteria WIPP Waste Isolation Pilot Plant



#### **EXECUTIVE SUMMARY**

Over the last several years most of the sludge and liquid from the Liquid Low-Level Waste (LLLW) tanks at ORNL has been transferred and consolidated in the Melton Valley Storage Tanks (MVST). The contents of the MVST tanks at the time the sludge samples were collected for this report included the original inventory in the MVSTs along with the sludge and liquid from the Bethel Valley Evaporator Service Tanks (BVEST), Old Hydrofi-acture (OHF) tanks, and the Gunite and Associated Tanks (GAAT). During the summer of 2001 full core samples of sludge were collected from the MVST tanks. The purpose of this sampling campaign was to characterize and validate that the current radiochemical and chemical contents of the MVST sludge, which was needed to meet the contract agreements prior to the transfer of the waste to another DOE contractor for processing. This report only discusses the analytical characterization of the sludge from the MVST waste tanks.

The isotopic data presented in this report supports the position that **fissile** isotopes of uranium (<sup>233</sup>U and <sup>235</sup>U) and plutonium (<sup>239</sup>Pu and <sup>241</sup>Pu) were "denatured" as required by the administrative controls stated in the ORNL LLLW waste acceptance criteria (WAC). In general, the MVST sludge was found to be hazardous by RCRA characteristics based on total analysis of chromium, mercury, and lead. Also, the alpha activity due to transuranic isotopes was well above the 100 nCi/g limit for TRU waste. The characteristics of the MVST sludge relative to the WIPP WAC limits for **fissile** gram equivalent, plutonium equivalent activity, and thermal power from decay heat, were estimated from the data in previous reports and found to be far below the upper boundary for any of the **remote**-handled transuranic waste (RH-TRU) requirements for disposal of the waste in WIPP. Therefore, the WIPP WAC limits were not evaluated for this set of samples.

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### Characterization of the Radioactive Sludge from the ORNL MVST Waste Tanks

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#### 1.0 Introduction

The active ORNL Liquid Low Level Waste (LLLW) system consists of the set of waste tanks summarized in Table 1. As indicated in Table 1, this report only discusses the analytical characterization data for the Melton Valley Storage Tanks (MVST). The characterization data summarized in this report was needed to verify the current composition of the sludge present after the transfer of sludge from the Bethel Valley Evaporator Service Tanks (BVEST), Gunite and Associated Tanks (GAAT), and the Old Hydrofracture (OHF) tanks. The data in this report can also be used to address waste processing options, support the performance assessment (PA) requirements for the Waste Isolation Pilot Plant (WIPP), evaluate the waste characteristics with respect to the waste acceptance criteria (WAC) for WIPP and Nevada Test Site (NT'S), address criticality concerns, and to meet DOT requirements for transporting the waste.

The samples and data collected for this project were performed during May and June of 2001. The level of quality assurance approximates what is required for regulatory measurements with the understanding that, when needed, sample size requirements were reduced, and steps were taken to reduce sample handling to ensure radiation exposures were as-low-as-reasonably-achievable (ALARA). Some procedure modifications were required to handle chemical matrix problems due to the high levels of sodium nitrate, uranium, and thorium present. Any deviations from procedures or problems observed with the tank samples were documented in the data tiles maintained by the laboratory. The regulatory holding time requirements for mercury and the organic measurements were complied with unless noted differently in the data tables. The Quality Control (QC) Acceptance Criteria for measurements used on this project are summarized in Appendix B.

Table 1 Summary of MVST Tanks in the ORNL LLLW System

	Data Presented in this report			
Tanks	Liquid	Sludge		
MVST TANKS				
W-24	none	✓		
W-25	none	<b>√</b>		
W-26	none	✓		
W-27	none	✓		
W-28	none	✓		
w-29	none	✓		
w-30	none	✓		
w-3 1	none	<b>√</b>		

#### 2.0 Sample Collection Activities

A detailed description on the background, operation of the LLLW system, and the sample collection techniques has been presented in previous reports and will not be discussed here (see Sections 2 and 3 of Reference 1). The staff from the Liquid and Gaseous Waste Operations (LGWO) provided all sample collection support and delivered the samples to the analytical laboratory. The tank location for sludge samples collected in 2000 are illustrated in Appendix A and there were only minor changes in the tank contents for this sampling campaign. The documentation for chain-of-custody was prepared, maintained for each sample collected, and stored with the data files by the analytical laboratory.

#### 3.0 Analytical Methodology

The information and data collected from these studies are used to support various activities. The activities include demonstration of regulatory compliance, measurements to support processing options, and to meet data needs for risk assessments and other safety related assessments such as

criticality. Standardized analytical procedures are used to the extent possible to ensure broad acceptance of the data generated. Unless stated otherwise, the U. S. Environmental Protection Agency (EPA) methods are used for the analyses of constituents listed as, hazardous under the Resource Conservation and Recovery Act (RCRA), which includes all the inorganic and organic measurements presented in this report. In general the EPA Guidance Manual, *Test Methods for Evaluating Solid Waste*<sup>2</sup> (SW-846), is used for inorganic and organic methods. Some modifications of the standard procedures are necessary to handle the high radiation levels and the high salt/solids content. Some procedure modifications are required to generate valid data, these changes were usually needed to correct for chemical or other matrix related interferences All deviations from the standard procedures are documented in the raw data files and can be provided upon request to data users.

#### 3.1 Sample Preparation

The interstitial liquid samples collected from the sludge were obtained by centrifuging a well mixed suspended sludge. The clarified liquids were then digested by the SW-846 Method 3015, *Microwave Assist&d Acid Digestion of Aqueous Samples and Extracts*, This sample preparation for aqueous samples was then used for all mercury analyses by CVAA.. Based upon results from a collaborative study<sup>3</sup> with Argonne National Laboratory - East (ANL-E), Method 3015/3051 demonstrated excellent recovery for mercury and was used to prepare tank samples for mercury determination.

The primary method for digesting the sludge samples was SW-846 Method 3051, *Microwave Assisted Acid Digestion of Sediments, Sludges, Soils, and Oils.* This sample preparation is considered to be a total digestion for metals and radionuclides by regulatory agencies and yields good results for most metals and radionuclides of interest. This digestion gave poor performance on two of the metals of interest, silver and silicon. Although nitric acid is excellent for dissolving' silver compounds, there is usually enough chloride present in waste samples to form an insoluble silver chloride (AgCl) precipitate. If the chloride concentration is increased sufficiently, a silver chloride complex (AgCl<sub>3</sub>-<sup>2</sup>) forms which is soluble in the aqueous environment. Improved matrix spike recovery and defensible data for silver were obtained in earlier sampling campaigns using a separate sample digestion discussed in the earlier reports.

If the total silicon content in the sludge must be known to develop waste treatment options such as vitrification, another sample digestion is required. A simple nitric acid treatment will not dissolve most siliceous materials. The SW-846 Method 3052, Microwave Assisted Acid Digestion of Siliceous and Organically Based Matrices, provides the necessary digestion chemistry to yield good silicon data. Sludge samples were prepared for measurement of total silicon, by taking approximately 0.5 g of sludge and mixing with 7 mL of concentrated nitric acid and 3 mL of hydrofluoric acid in a fluorocarbon microwave vessel. The samples were digested for 10 minutes at full power (1200 watts) holding the digestion temperature at 190°C and then cooled to room temperature. The acid solution was then treated with excess boric acid and heated to 80°C for ten minutes to complex any free fluoride. This digestion mixture is cooled, filtered into a 50 mL volumetric flask, and diluted to volume with ASTM Type II water. Care'must be exercised to ensure the digestion solution is cooled to room temperature prior to opening the sealed microwave vessel or there may be a significant loss of the volatile  $SiF_4$ . The free fluoride is complexed with the boron to protect the sample introduction system to the ICP-AES and to prevent a high silicon background from the instrument glassware. This sample digestion with hydrofluoric acid should not be used for radiochemical measurements, especially for measurement of lanthanides or actinides.

Most of the metal and radionuclide data presented in this report are based'upon a Method 3051 digestion with approximately a 0.5 gram sludge sample and  $10\,\mathrm{mL}$  of concentrated nitric acid. After the microwave digestion is completed and the solution cooled to room temperature, the **sample** is filtered into a volumetric flask and diluted to  $50\,\mathrm{mL}$  with ASTM Type II water or better. Any residue remaining after the nitric acid digestion consisted of mostly  $\mathrm{SiO}_2$  and was discarded.

#### 3.2 Metal Analysis

Three analytical measurement methods were used to determine all of the metals included in this report. Most of the metals are first determined by SW-846 Method 6010A, *Inductively Coupled Plasma -Atomic Emission Spectroscopy (ICP-AES)*. There are several elements of interest for which the ICP-AES has insufficient detection limits, and these elements must be determined by Method 7000A, *Atomic Absorption Methods*. The Radioactive Materials Analytical Laboratory (RMAL) uses a Graphite Furnace Atomic Absorption (GFAA) Spectrometer or Inductively Coupled Plasma-Mass Spectrometry (ICP-MS) for elements that require better sensitivity. The elements lead (Pb), thallium (Tl) and antimony (Sb) were analyzed by ICP-MS (Method 6020). The GFAA methods were used for arsenic (Method 7060A) and selenium (Method 7740). All the mercury measurements are done by either Method 7470A, *Mercury in Liquid Waste (Manual Cold-Vapor Technique)*, or Method 7471A, *Mercury in Solid or Semisolid Waste (Manual Cold-Vapor Technique)*. The samples discussed in this report were prepared for mercury analysis by the microwave technique discussed in section 3.1, the sample preparation specified in the mercury methods (7470A and 7471A) were not used.

The level of radioactivity in most LLLW tank samples required that the analytical systems used for metal measurements be modified for operation in a radiochemical hood or glove box. Custom instrument configurations are necessary to ensure contamination control and worker safety. All work was performed in radiochemical laboratories which are operated under strict radiation protection programs, with the use of protective clothing and routine contamination monitoring. Both an ICP-AES system and a GFAA system can generate dry, dusty particles which are difficult to contain and are highly hazardous when radioactive. A detailed description of the RMAL setup for these instruments are given in Appendix B of Reference 1.

The instrument detection limits (IDL) for various metals with undiluted aqueous samples are listed in data tables along with the results. For sludge samples, these detection limits must be increased by a factor that represents the dilution that results from the sample preparation. 'For all the MVST sludge samples approximately 0.5 g of sample was digested and then diluted to 50 mL which results in about a 100 fold dilution for the sample, and thus a 100 fold increase in the detection limits.

The analytical error for the metal measurements depends upon the analytical method, the concentration level, and the chemical matrix. Inductively coupled plasma-atomic emission spectroscopy (ICP-AES) and inductively coupled plasma-mass spectrometry (ICP-MS) are both multi-element measurement techniques that are designed for the best average performance for all elements analyzed. In general, these measurement techniques are not optimized for any single element, The sample introduction system for ICP instruments adds additional variability due to changes in sample density, viscosity, and solids content between samples and/or calibration standards. Overall, the expected analytical error for ICP measurements range from  $\pm 4-6\%$  at concentrations above 10 times the detection limit to  $\pm 20-50\%$  near the detection limit. These error estimates are typical for both ICP-AES and ICP-MS measurements.

Graphite Furnace AA instruments are generally optimized for a specific element and usually provide lower detection limits and better precision. The expected analytical error for GFAA measurements range from 3-5% for concentrations greater than 10 times the detection limit to 20-40% near the detection limit. One advantage of GFAA analysis is that the measurements are normally well above the method's detection limits. The mercury measurements were done by Cold Vapor Atomic Absorption (CVAA), which is very selective and sensitive for mercury. The analytical errors for CVAA measurements are similar to GFAA work.

#### 3.3 Anion Analysis

The determination of the inorganic anions was needed for the development of process treatment options, to provide information to explain the distribution and chemical behaviors observed in the waste tanks, and to ensure the major chemical constituents were identified in the waste for which data was used to calculate the mass and charge balance for each sample. The common inorganic anions; including fluoride, chloride, bromide, phosphate, nitrate, nitrite, and sulfate; were measured by ion chromatography (IC) with a Dionex Model DX600 system. In addition, several water soluble organic acids were measured along with the inorganic anions. These organic acids were measured in their ionized form and included formate, acetate, citrate, and oxalate. Both the citrate and the oxalate can form strong complexes with many metals and change the solution chemistry of these metals in the waste. The ion chromatography system used for measurements on these radioactive

samples was configured such that the components that come into contact with radioactivity were isolated in a radiochemical hood for contamination control,

From past observations, the nitrate content dominates both the mass and charge balance calculations with sludge samples taken from the active LLLW tanks. There are many other anions present in the waste, some of which are measured directly by ion chromatography and others which can be estimated from the metal data such as chromate, dichromate, permanganate, and others. The carbonate is estimated from the total inorganic carbon measurement.

The anion measurement technique was ion chromatography. For simple water samples, without complex chemical matrix problems, the empirical analytical error for ion chromatography measurements ranges from 4-6% for concentrations above 10 times the detection limits to 20-40% near the detection limit. The measurement of anions present at concentration much lower (<1/25) than other anionic species present may increase the overall error of the measurement.

#### 3.4 Radiochemical Analysis

The only standard radiochemical methods useful for radioactive waste characterization are EPA Method 600/900.0, *Gross Alpha and Beta Radioactivity in Drinking Water*, and EPA Method 600/901.1: *Gamma Emitting Radionuclides in Drinking Water*. The EPA Method 6001905.0, *Radioactive Strontium in Drinking Water*, gave poor performance with the chemical matrix found in ORNL LLLW supematant and sludge samples. The EPA method for gross alpha/beta measurements uses gas-flow proportional counting. In general, this counting technique requires drying a sample at elevated temperatures onto a metal (usually stainless steel) plate, which resulted in the loss of cesium chloride from the MVST samples and yielded poor gross beta results. To avoid this problem, all gross beta measurements reported are based on measurements by liquid scintillation counting. Other than the gamma spectroscopy measurements, all of the radionuclide measurements were done with in-house procedures. The method detection limits for radiochemical measurements are dependent on both sample matrix and count time and are not listed here. In general, the radiochemical measurements used count times to yield at least 1% (10,000 counts) counting statistics. The expected errors for the radiochemical data range from ±5-10 % for gross alpha/beta

and gamma emitter measurements to  $\pm 1$  O-20 % for radionuclides that require chemical separations before counting (i.e.  $^{99}$ Tc,  $^{90}$ Sr,  $^{129}$ I, and  $^{237}$ Np).

#### 3.5 Criticality Controls

The current ORNL waste acceptance criteria (WAC) for liquid-low level waste requires that the fissile isotopes of uranium and plutonium be isotopically diluted with  $^{238}$ U and  $^{232}$ Th, respectively. These administrative controls require that the ratio of the  $^{238}$ U mass divided by the fissile equivalent mass (FEM) for uranium be greater than 100. The  $^{235}$ U FEM is a useful scale for criticality calculations that normalizes the fission probability for each fissile isotope to  $^{235}$ U. These FEM factors, designated as  $f_{35}$  for  $^{235}$ U mass factors, are discussed and listed in the Appendix A, Table 1 of the ORNL Procedure NCS-1 .0, *Nuclear Criticality Safety Program*.

The major fissile isotopes of concern in the ORNL waste tanks are <sup>233</sup>U, <sup>235</sup>U, and <sup>239</sup>Pu. The fissile isotope <sup>241</sup>Pu is also present in the waste but the mass is usually several orders of magnitude lower and below a level that would influence the isotopic dilution ratio for plutonium. Other fissile isotopes present in the ORNL waste include isotopes of neptunium, americium, and curium, but the actual mass present in the waste has been too low for major concern, and the low concentration would make it difficult and expensive to measure by mass spectrometry.

The data presented in this report for isotopic dilution ratios (also referred to as denature ratios) reflect both the current ORNL standard practices for disposal of fissile isotopes of uranium and plutonium. The administrative controls which were in effect when the waste was generated were different than current practices.

All calculations dealing with isotopic dilution for criticality safety are based on isotope mass ratios and must not be confused with activity ratios. For any data discussed in this report that uses <sup>232</sup>Th relative to isotopic mass ratios, the total thorium concentration and the <sup>232</sup>Th concentration are the same value.

The current administrative requirements for criticality control are more conservative than past practices and require that the following conditions be satisfied for uranium,

$$\frac{\binom{238}{U} - 200\binom{233}{U}}{\binom{235}{U}} \ge 110 \tag{1}$$

$$\frac{\binom{238}{U} - 110\binom{235}{U}}{\binom{233}{U}} \ge 200 \tag{2}$$

The administrative controls for plutonium require a dilution ratio of 200 parts thorium to one part plutonium (past practices only required a ratio of 100).

#### 4.0 Quality Assurance

Quality assurance during the sampling activities was primarily addressed by the use of approved procedures for sampling the sludge phase found in each waste tank. These procedures provide detailed instructions for the collection, labeling, and shipping of each sample. Chain-of-custody forms were used to track individual samples from their collection point to the analytical laboratory.

The QA for the sludge characterization was based upon the RMAL Radioactive 'Waste Characterization QA Plan<sup>4</sup> which defines the basis for quality assurance and quality control used for the analysis of the waste tank samples. The QA plans discuss staff qualification requirements, laboratory participation in performance demonstration programs, quality control acceptance criteria for analytical methods, sample management, and most other laboratory operations. The QA plans implemented at the RMAL for waste characterization meet both the WIPP and the Nevada Test Site (NTS) QA requirements for inorganic, organic, and radiochemical measurements.

#### 5.0 Summary of Inorganic and Radiochemical Analytical Results

#### 5.1 <u>Description of Data Tables</u>

A summary of the inorganic, physical, and radiochemical analytical results for MVST sludge samples are presented in Table 2 through Table 7. Also, MVST data<sup>5</sup> collected in 1996 was included in these data tables for comparison. These tables are arranged in a similar format to facilitate comparing data from different tanks and to group information into useful units. The analytical data presented in these tables are the consolidation of data from a single project which had a fixed set of analytical requirements. Any parameter reported with a dash ("-") indicates that the data was not measured for that sample.

The first section, "Physical properties and miscellaneous data", includes unrelated information that does not fit well into other table groups. The first parameters entered in a column include the RMAL request and sample numbers, which are laboratory filing codes used to track sample information. The next set of data includes information on the moisture or water content and the solids content of the sample. The group is completed with data on the inorganic and organic carbon content. For MVST waste tank samples the inorganic carbon can be assumed to be all carbonate and bicarbonate. The Total Organic Carbon (TOC) provides an upper limit on the organic content in the tank waste but does not include volatile organic compounds. Most of the liquid waste in the active system has been through an evaporator which removes the highly volatile organic compounds from the waste.

The next two sections include groups of metals; the "RCRA metals" are separated out for quick reference. The regulatory limit for the concentrations are listed in parentheses next to each RCRA metal. For the liquid samples the RCRA regulatory limits are used directly, since the supernatant would be defined as the TCLP leachate in the determination of waste characteristics for hazardous waste. The RCRA metal sludge data represents total metal measurements, as defined by EPA. Exceeding the RCRA regulatory limits listed for the sludge samples only indicates that the waste has the potential to be classified as hazardous. The sludge waste should only be classified as RCRA waste if the final waste form fails the TCLP leaching test.

The remaining metals are grouped under "<u>Process metals</u>", which includes the common Group IA & IIA metals along with elements that could effect chemical processing, criticality concerns, and stabilization techniques such as grouting or vitrification. For the sludge data, all the metals are reported on a "as received" (wet weight) basis.

The section "Semi-quantitative metals by ICP-MS" includes additional metals identified in a full mass range scan by inductively-coupled plasma - mass spectrometry. This measurement helps ensure all major elements have been identified in the waste. Each element reported is not calibrated but is based upon a response factor from a curve generated **from** a few elements across the mass range. Therefore, these elemental concentrations are listed as estimates only.

The "Anions by ion chromatography" section includes the inorganic anions, several common water soluble organic acids. Two of the organic acids included, citrate and oxalate, are also classified as complexing agents.

The "Beta/gamma emitters" section summarizes the radionuclides that emit gamma-rays and beta particles. This section includes the gross beta activity, radionuclides identified by gamma spectrometry, and several "pure" beta emitters of interest. Many of the "pure" beta emitters (<sup>3</sup>H, <sup>14</sup>C, and <sup>90</sup>Sr) require radiochemical separations prior to measurement by either liquid scintillation or gasflow proportional counting. The <sup>99</sup>Tc was measured by ICP-MS without any prior chemical separation.

The "Alpha emitters" section summarizes the actinide elements in the waste. This section includes the gross alpha activity, an estimate of the activity for each alpha emitter identified in a gross alpha spectrum, and plutonium isotopes determined by alpha spectrometry after a radiochemical separation. For supernatant samples, an estimate of the <sup>232</sup>Th/<sup>239</sup>Pu mass ratio is included in this section to address criticality concerns if enough thorium is present to calculate the ratio. For the sludge samples, this mass ratio is included with the plutonium mass spectrometry data.

The remaining sections include "<u>Uranium isotopes by TIMS</u>", "<u>Plutonium isotopes by TIMS</u>", and "<u>Uranium isotopes by ICP-MS</u>". These sections summarize the uranium and plutonium data measured by thermal ionization mass spectrometry and, for more recent measurements, the uranium isotopes by ICP-MS. Also, included in these sections are the isotopic mass dilution or "denature" ratios for uranium and plutonium based on the requirements in place when the waste was generated (see section 3.5). The plutonium section for the sludge samples also includes the activity for each plutonium isotope, which was calculated **from** the mass spectrometry data.

Characteristic (Analysis)		<b>1996</b> w-24 s	2000 w-24 s	2001 W-24 S	IDL'
lPhysical propert	ies and miscellar	neous data			• 00- 01
IRequest number Sample number pH		7749c 960806-006 12.8	10224. 000509-001 9.8	, 12057 <b>010716-010</b> 9.96	
Water <sup>a</sup> TS <sup>b</sup> TSS <sup>c</sup> TDS <sup>d</sup> Bulk density TC <sup>e</sup> TIC <sup>f</sup> TOC <sup>g</sup>	(%) (mg/g) (mg/g) (mg/g) (g/mL) (mg/Kg) (mg/Kg) (mg/Kg)	51.2 488 1.37 13700 13700 < 15	64.0 $360$ $253$ $107$ $1.315$ $28800$ $8790$ $20000$	57.6 424 - 1.276 22000 8500 13400	15 15 15 15
RCRA Metals (:	±10%)				
Agh (100)' As (100) Ba (2000) Cd (20) Cr (100) Hg (4) Ni (1000) Pb (100) Se (20) Tl (18)	(mg/Kg)	< 1.9 < 5.3 75.5 13.9 61.6 38.0 45.2 303 < 5.3 < 5.3	< 0.8 86.2 21.1 236 74.0 76.3 435 < 0.8 < 0.8	< 1.7 107 37.1 276 49.9 90.0 606 < 4	0.012 0.005 0.001 0.168 0.013 0.002 0.078 0.341 0.005 0.005
Process metals (	±10%)				
Al B Be Ca co cu Cd Fe K Mg Mn Mo Na P Sb Si K Sr Th Ti U V Zn	(mg/Kg)	3330 4.35 4.45 51200 2.42 28.5 0.900 1250 13400 9280 84.7  48800 1240 < 19 3820 283 3270  6780 2.23 479	3540 9.52 7.28 42400 22.9 67.4 0.409 2990 7980 6330 911 15.1 35200 < 41 2340 208 7480 46500 < 1 658	4130 5.72 6.42 55400 42.3 70.6 < 4 3010 9 6 0 0 9270 7 6 1 56.8 40000 - < 4 3490 247 7290 30.0 33800 < 1 700	0.035 0.01 0.001 0.03 0.039 0.006 0.005 0.014 0.5 0.049 0.002 0.038 0.075 0.13 0.509 0.022 0.001 0.376 0.010 0.077 0.013 0.445

Characteristic (Analysis)		1996 w-24 s	2000 w-24 s	2001 W-24 S	IDL'
Semi-quantitative	metals by ICP-	MS (±30-50 %, * ii	ndicates data from w	vater leach)	
Au, gold	(mg/Kg)	1.5	-		
Bi, bismuth	(mg/Kg)	170	-		
Ce, cerium	(mg/Kg)	6.5	-	24.4	
Er, erbium	(mg/Kg)	0.25			
Eu, europium	(mg/Kg)	1.1			
Ga, gallium	(mg/Kg)	5.3	-		
Gd, gadolinium	(mg/Kg)	1.2	-	8.61	
Ho, holmium	(mg/Kg)	1.0	<b>-</b>		
I, iodine	(mg/Kg)	* 13			
La, lanthanum	(mg/Kg)	9.1		79.3	
Li, lithium	(mg/Kg)	* 170		111	
Mo, molybdenum	(mg/Kg)	* 2.1			
Nb, niobium	(mg/Kg)	0.93			1
Rb, rubidium	(mg/Kg)	* 1.4	-		
Sn, tin	(mg/Kg)	12	<del>.</del>		•
Ti, titanium	(mg/Kg)	21	-		
W, tungsten	(mg/Kg)	1.0	-		
Zr, zirconium	(mg/Kg)	8.4	-		_
Anions by ion chi	romatography i	n water wash of <b>slu</b> d	lge ( ±10%)		
Inorganic_					
Bromide	(mg/Kg)	< 50	< 41	149	0.05
Chloride	(mg/Kg)	2770	1560	1770	0.05
Chromate	(mg/Kg)	< 20	18.2	36.4	0.05
Fluoride	(mg/Kg)	103	468	216	0.05
Nitrate	(mg/Kg)	165000	$6\ 7\ 4\ 0\ 0$	99700	0.10
Nitrite	(mg/Kg)	2250	3920	5030	0.10
Phosphate	(mg/Kg)	< 20	35.0	< 8	0.20
Sulphate	(mg/Kg)	1370	2570	4690	0.10
Organic	(0)				***
Acetate	(mg/Kg)	242	1310	1250	0.05
Citrate	(mg/Kg)	< 20	5.12	19.3	0.50
Formate		175	110	149	0.05
Oxalate	(mg/Kg) (mg/Kg)	690	397	196	0.05
Phthalate	(mg/Kg)	< 20	27.9	< 4	0.05

Characteristic [Analysis]		1996 W-24 S	2000 <b>W-24</b> S	2001 <b>W-24 S</b>	IDL'			
Beta/gamma emit	Beta/gamma emitters (±10%)							
Gross beta	(Bq/g) (Bq/g)	4.6e+06 < 2.5e+01	4.9e+06	4.4e+06				
<sup>33</sup> Ni	(Bq/g)	3.3e+03						
юСо	(Bq/g)	2.8e+04	3.0e+04	2.5e+04				
<sup>10</sup> Sr/ <sup>90</sup> Y	(Bq/g)	1.4e+06	1.3e+06	1.2e+06				
<sup>9</sup> Tc	(Bq/g)	4.5e+02		7.9e+02				
1 <b>29</b> I	(Bq/g)			•				
<sup>134</sup> Cs	(Bq/g)	1.3e+04	-	3.3e+03	<b>-</b> .			
<sup>137</sup> Cs	(Bq/g)	5.3e+05	6.3e+05	5.5e+05	-			
<sup> 51</sup> Sm	(Bq/g)	< 6.0e+02			-			
<sup>152</sup> Eu	(Bq/g)	8.9e+04	2.4e+05	6.9e+05	-			
<sup>154</sup> Eu	(Bq/g)	3.8e+04	2.5e+05	2.2e+05				
<sup>155</sup> Eu	(Bq/g)	1.0e+04	5.1e+04	6.6e+04				
<sup>227</sup> Ac	(Bq/g)	<4.7e+03						
<sup>241</sup> Pu	(Bq/g)	1.4e+04						
Alpha emitters ( ±	<b>⊧10%</b> )							
Gross alpha	(Bq/g)	34000	200000	150000				
<sup>232</sup> Th	(Bq/g)	13	30	30				
233 U	(Bq/g)	1600	8900	640				
<sup>234</sup> U	(Bq/g)	77	710	184				
235 U	(Bq/g)	2.6	15	9.9				
238 U	(Bq/g)	84	580	419				
<sup>237</sup> Np	(Bq/g)	10						
<sup>241</sup> Am	(Bq/g)	3900	10400	15000				
<sup>244</sup> Cm	(Bq/g)	22000	155000	105000				
<sup>250</sup> Cf	(Bq/g)	< 100		•				
<sup>252</sup> Cf	(Bq/g)	< 100						
	(Bq/g)							
Total Pu alpha	(Bq/g)	6600	28000	22000				
<sup>238</sup> Pu		4000	17000	12000				
<sup>239</sup> Pu/ <sup>240</sup> Pu	(Bq/g)	2600	11000	9400				
<sup>242</sup> Pu	(Bq/g)							
TRU activity								
Pu+Am (3700)	(Bq/g)	10500	38400	37000				
Uranium isotopic	s by ICP-MS ( :	±2%)			1			
<sup>233</sup> U	(atom <b>%)</b>	0.067	0.0548	0.0546	0.001			
234 <b>[</b> ]	(atom %)	0.005	0.0067	0.0024	0.001			
235U 235U	(atom %)	0.543	0.4606	0.4157	0.001			
236U	(atom %)	0.006	0.0028	0.0032	0.001			
238 U	(atom %)	99.379	99.475 1	99.5241	0.001			
	(mg/Kg)	22.0.0	23,1.0 1	30.0211	0.001			
<sup>233</sup> U/MS	(mg/Kg)	4.45	24.9	18.1				
<sup>235</sup> U/MS	m	36.4	211	139				
<sup>238</sup> U/ <sup>235</sup> U FEM	-	159	189	206	l			

Characteristic (Analysis)		1996 W-24 S	2000 W-24 S	2001 W-24 S	IDL'		
Plutonium isotop	Plutonium isotopics by TIMS ( ±1%)						
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu "'Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %)	0.63 87.14 10.81 0.37 1.05 < 0.01	-				
Pu activity <sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	3800 1900 870 14000 1.5 < 0.1					
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu <sup>m</sup>	(ng/g)	960 3920	613	761			

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>239</sup>Pu.

Analytical Data for Sludge in Tanks W-25 Table 3 Characteristic 1996 2000 2001 W-25 S W-25 S W-25 S IDL' (Analysis) IPhysical properties and miscellaneous data Request number 7749D 10224 12057 Sample number 960822-036 000509-002 010716-011 рΗ 12.6 11.0 11.8 Watera (%) 50.9 59.5 55.4  $TS^b$ (mg/g)1 406 446 9 TSSc 282 (mg/g)  $TDS^d$ (mg/g)124 Bulk density (g/mL)1.331 1.359 1.36 TCe (mg/Kg) 20900 23300 15 15700 TIC' (mg/Kg) 15700 15 11100 11000 TOCg (mg/Kg) < 15 9800 12300 15 RCRA Metals (±10%) 0.006  $Ag^h$ (mg/Kg) < 1.8 (100)'24.8 (100)(mg/Kg) < 0.9 < 1.3 < 1.7 0.011 As(mg/Kg) (2000) 73.8 106 0.001 Ва 105 (20)(mg/Kg) 26.0 32.5 Cd11.9 0.111 (100)(mg/Kg)  $\operatorname{Cr}$ 92.1 190 197 0.008 Hg (mg/Kg) 73.2 49.1 106 0.011 (4) (mg/Kg) (1000)56.8 82.3 85.8 0.065 Ni Pb (100)(mg/Kg) 454 442 679 0.341 (mg/Kg) (20)<1.3 < 0.9 < 0.8 Se0.005 Tl (mg/Kg) < 4 (18)< 1.3 < 0.9 0.005 Process metals (±10%) (mg/Kg) Al 581Ø 2610 658Ø 0.048 (mg/Kg) В 3.76 7.34 3.65 0.024 (mg/Kg) 6.40 7.83 0.001 Be6.91 (mg/Kg) Ca 50800 56100 60000 0.017 (mg/Kg) 5.86 28.1 42.5 0.060 c o (mg/Kg) c u 37.Ø 51.4 61.6 0.006 (mg/Kg) Csi < 0.09 < 4 Ø.857 0.005 (mg/Kg) 2160 Fe 1810 253Ø Ø.233 K (mg/Kg) 8850 9820 9230 Ø.178 (mg/Kg) 765Ø 11500 10600 Mg 0.053 (mg/Kg) Mn 140 616 469 0.001 (mg/Kg) Mo 37.0 91.6 0.030 (mg/Kg) 52100 42000 46400 0.065 Na P (mg/Kg) 1850 0.13 (mg/Kg) < 50 Sb 114 < 4 0.509 (mg/Kg) Si k 8890 267Ø 563Ø 0.022 (mg/Kg)  $\operatorname{Sr}$ 325 251 234 0.001 Th (mg/Kg) 9250 896Ø 10100 0.181Ti (mg/Kg) 32.5 0.010 U (mg/Kg) 7660 30600 18000 0.105V (mg/Kg) < 1.2 3.85 < 1 0.013 Zn (mg/Kg) 285 623 458 0.390

331

0.008

Zr

(mg/Kg)

Characteristic (Analysis)		1996 W-25 S	2000 <b>W-25</b> S	2001 <b>W-25 S</b>	IDL'		
Semi-quantitative	Semi-quantitative metals by ICP-MS (±30-50 %, * indicates data from water leach)						
Au, gold	(mg/Kg)	0.28					
Bi, bismuth	(mg/Kg)	250					
Ce, cerium	(mg/Kg)	9.4		15.5	•		
Er, erbium	(mg/Kg)	0.02					
Eu, europium	(mg/Kg)	2.1					
Ga, gallium	(mg/Kg)	8.1					
Gd, gadolinium	(mg/Kg)	1.7		5.52			
Ho, holmium	(mg/Kg)	2.0					
I, iodine	(mg/Kg)	* 12					
La. lanthanum	(mg/Kg)	18		38.3			
Li, lithium	(mg/Kg)	* 33		98.6			
Mo,molybdenum	(mg/Kg)	* 2.0					
Nb, niobium	(mg/Kg)	0.72					
Rb, rubidium	(mg/Kg)	* 1.0					
Sn, tin	(mg/Kg)	18					
Ti, titanium	(mg/Kg)	47					
W, tungsten	(mg/Kg)	0.61					
Zr, zirconium	(mg/Kg)	16					
Anions by ion chi	romatography in	water wash of slud	ge ( ±10%)				
Inorganic							
Bromide	(mg/Kg)	< 50	<b>&lt;</b> 50	128	0.05		
Chloride	(mg/Kg)	2110	1630	1720	0.05		
Chromate	(mg/Kg)	95.5	29.5	89.1	0.05		
Fluoride	(mg/Kg)	118	251	155	0.05		
Nitrate	(mg/Kg)	162000	76600	120000	0.10		
Nitrite	(mg/Kg)	4967	3190	5440	0.10		
Phosphate	(mg/Kg)	< 20	< 10	< 10	0.20		
Sulphate	(mg/Kg)	1750	2590	2810	0.10		
Organic Organic	( <i>ab</i> )	1.00					
Acetate	(mg/Kg)	318	919	641	0.05		
Citrate	(mg/Kg)	< 20	25.5	52.7	0.50		
Formate	(mg/Kg)	247	130	174	0.05		
Oxalate	(mg/Kg)	521	643	238	0.05		
Phthalate	(mg/Kg)	< 20	18.4	< 5	0.05		

Characteristic		1996	2000 W 25 G	2001 W 25 G	IDI 1
(Analysis)		W-25 S	W-25 S	W-25 S	IDL'
Beta/gamma emitt	ers ( ±10%)				
Gross beta	@q/g)	8.3e+06	4.3e+06	5.9e+06	
<sup>59</sup> Ni	(Bq/g)	< 2.5e+01			
<sup>53</sup> Ni	(Bq/g)	3.4e+03			
<sup>50</sup> Co	(Bq/g)	2.5e+04	3.5e+04	2.2e+04	
90Sr/90Y	(Bq/g)	3.2e+06	1.3e+06	2.3e+06	-
99Tc	(Bq/g)	1.0e+02	1.50 / 00	9.1e+02	
129 I	(Bq/g) (Bq/g)	1.00 102		7.10102	
134Cs		6.0e+03		< 1.8e+03	
<sup>137</sup> Cs	(Bq/g)		40.105		
	(Bq/g)	4.7e+05	4.0e+05	3.5e+05	
<sup>151</sup> Sm	(Bq/g)	< 5.5e+02			
<sup>152</sup> Eu	(Bq/g)	7.1e+04	6.0e+05	3.0e+05	
<sup>154</sup> Eu	(Bq/g)	3.7e+04	< 7.1e+04	1.0e+05	
<sup>155</sup> Eu	(Bq/g)	8.4e+03	2.8e+04	2.2e+04	
<sup>227</sup> Ac	(Bq/g)	< 5.3e+03			
<sup>241</sup> Pu	(Bq/g)	2.6e+04			
Alpha emitters (±	:10%)				
Gross alpha	(Bq/g)	83000	130000	110000	
<sup>232</sup> Th	(Bq/g)	38	36	41	
	(Bq/g) (Bq/g)			3900	
<sup>233</sup> U		2800	7200		
<sup>234</sup> U	(Bq/g)	100	28	98	
<sup>235</sup> U	(Bq/g)	3.2	7.9	5.9	
238 U	(Bq/g)	95	380	230	
<sup>237</sup> Np	(Bq/g)	10			
"'Am	(Bq/g)	9300	10300	8600	
<sup>244</sup> Cm		58000	90500	75000	
<sup>250</sup> Cf	(Bq/g)	< 100			
<sup>252</sup> Cf	(Bq/g)	< 100			
	(Bq/g)				
Total Pu alpha	(Bq/g)	13000	23000	22000	
<sup>238</sup> Pu	(Bq/g)	7700	14000	13000	
<sup>239</sup> Pu/ <sup>240</sup> Pu	(Bq/g) (Bq/g)	4900	9500	9000	
<sup>242</sup> Pu	(Dd/R)	4000	9900 -	<i>9</i> 000	
- w			_		
TRU activity					
Pu+Am (3700)	(Bq/g)	22300	33300	30600	
Uranium isotopics			33333	3330	
_		1			
$^{233}U$	(atom %)	0.103	0.0674	0.0614	0.001
<sup>234</sup> U	(atom %)	0.006	0.0004	0.0024	0.001
235 Ŭ	(atom %)	0.597	0.3670	0.4147	0.001
236[ ]	(atom %)	0.006	0.0048	0.0058	0.001
238U	(atom %)	99.289	99.5604	99.5157	0.001
	,				
<sup>233</sup> U/MS	(mg/Kg)	7.72	20.2	10.8	
<sup>235</sup> U/MS	(mg/Kg)	45.2	111	73.7	
$^{238}U/^{235}U$ FEM	\ U -0)	137	221	203	
J. J. 11111		1			1

Characteristic (Analysis)		1996 W-25 S	2000 W-25 S	2001 W-25 S	IDL'
Plutonium <b>isoto</b> j	pics by TIMS ( ● 1	%)			
<sup>138</sup> Pu <sup>139</sup> Pu <sup>140</sup> Pu <sup>141</sup> Pu <sup>142</sup> Pu <sup>144</sup> Pu	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %)	0.72 84.95 12.42 0.40 1.51 < 0.01	-		
Pu activity <sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	7800 3400 1800 26000 3.8 < 0.1	-		
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu	(ng/g)	1700 6320	-	1054	

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>259</sup>Pu.

Table 4 Analytical Data for Sludge in Tanks W-26

able 4 Analytical Data for Sludge in Tanks W-26						
Characteristic	1996	2000 W 26 S	2001 <b>W-26 S</b>	IDL,		
(Analysis)	W-26 S	W-26 S	W-20 S	IDL		
Physical properties and miscellaneous data						
Request number	7749E	10224	12057			
Sample number	960830-044	000509-003	010716-012	:		
pH	9.7	9.1	9.4			
Water 4 (0/)	<b>7</b> 0.0	* *	<b>*</b> 0.0			
Water <sup>a</sup> (%) TS <sup>b</sup> (mg/g)	50.9 491	56.1 439	52.9 $471$			
TSb (mg/g) TSSc (mg/g)	491	$\frac{439}{285}$	471			
$TDS^d$ (mg/g)		154				
Bulk density (g/mL)	1.38	1.363	1.675	-		
TC" (mg/Kg)	13500	10300	11300	15		
TIC' (mg/Kg)	11600 .	3790	3000	15		
TOC <sup>g</sup> (mg/Kg)	1900	6510	8300	15		
RCRA Metals (±10%)	<u></u>	<u> </u>	-			
<u> </u>	1	<u> </u>				
Agh (100)' (mg/Kg)	< 1.9		-10	0.006		
As (100) (mg/Kg)	< 1.4	< 0.9	< 1.8	0.011		
Ba (2000) (mg/Kg) Cd (20) (mg/Kg)	63.1 19.8	77.9 21.9	$134 \\ 48.2$	0.001 0.111		
Cr (100) (mg/Kg)	74.4	153	212	0.008		
Hg $(4)$ $(mg/Kg)$	12.7	58.3	89.5	0.011		
Ni (1000) (mg/Kg)	42.8	74.5	94.3	0.065		
Pb (100) (mg/Kg)	212	331	635	0.341		
Se (20) (mg/Kg)	<b>&lt;</b> 1.4	< 0.9	< 0.8	0.005		
Tl (18) (mg/Kg)	< 1.4	< 0.9	< 4	0.005		
Process metals (±10%)	Process metals (±10%)					
Al (mg/Kg)	1980	7130	7590	0.048		
B (mg/Kg)	11.3	10.6	6.87	0.024		
Be (mg/Kg)	1.85	5.63	5.33	0.001		
Ca (mg/Kg)	45900	43200	61700	0.017		
c o (mg/Kg)	2.69	21.1	41.5	0.060		
c u (mg/Kg) Cs <sup>j</sup> (mg/Kg)	29.0 1.53	$70.6 \\ 0.729$	63.7 <b>&lt; 4</b>	$0.006 \\ 0.005$		
Fe (mg/Kg)	1010	2380	2740	0.003 $0.233$		
K (mg/Kg)	25300	18400	17400	0.178		
Mg (mg/Kg)	14700	12800	14500	0.053		
Mn (mg/Kg)	102	180	278	0.001		
Mo (mg/Kg)	See	46.4	72.4	0.030		
N a (mg/Kg)	48900	40500	39700	0.065		
P (mg/Kg)	1070			0.13		
Sb (mg/Kg)	52.8	< 50	< 4	0.509		
Si $^{k}$ (mg/Kg)	2100	7470	8330	0.022		
Sr (mg/Kg) Th (mg/Kg)	$\begin{array}{c} 254 \\ 3280 \end{array}$	195	216	0.001 0.181		
Th (mg/Kg) Ti (mg/Kg)	3400	4330	$5880 \\ 30.2$	0.181		
U $(mg/Kg)$	19400	36900	30200	0.105		
V (mg/Kg)	2.32	< 1.2	< 1	0.103		
Zn (mg/Kg)	405	360	440	0.390		
Zr (mg/Kg)	1		200	0.008		

Characteristic (Analysis)		1996 W-26 S	2000 W-26 S	2001 W-26 S	IDL'	
Semi-quantitative	Semi-quantitative metals by ICP-MS ( ±30-50 %, * indicates data from water leach)					
Au, gold	(mg/Kg)	0.92				
Bi, bismuth	(mg/Kg)	78		22.1		
Ce, cerium	(mg/Kg)	5.5		22.1		
Er, erbium	(mg/Kg)	$0.24 \\ 2.3$				
Eu, europium Ga, gallium	(mg/Kg) (mg/Kg)	$\frac{2.3}{4.0}$				
Gd, gadolinium	(mg/Kg)	6.4		8.56		
Ho, hohnium	(mg/Kg)	1.0	_	0.50		
I, iodine	(mg/Kg)	* 12	_			
La, lanthanum	(mg/Kg)	4.8	-	88.8		
Li, lithium	(mg/Kg)	* 76		116		
Mo,molybdenum	(mg/Kg)	* 2.2		110		
Nb, niobium	(mg/Kg)	0.22		-		
Rb, rubidium	(mg/Kg)	<b>*</b> 2.5				
Sn, tin	(mg/Kg)	7.3				
Ti, titanium	(mg/Kg)	3.2				
W, tungsten	(mg/Kg)	1.5				
Zr, zirconium	(mg/Kg)	5.4				
Anions by ion chr	omatography i	n water wash of slud	ge (±10%)			
Inorganic						
Bromide	(mg/Kg)	<b>&lt;</b> 50	< 42	186	0.05	
Chloride	(mg/Kg)	3070	1800	1840	0.05	
Chromate	(mg/Kg)	< 20	4.94	12.5	0.05	
Fluoride	(mg/Kg)	<b>&lt;</b> 50	119	64.3	0.05	
Nitrate	(mg/Kg)	214000	115000	144000	0.10	
Nitrite	(mg/Kg)	1652	2210	3150	0.10	
Phosphate	(mg/Kg)	< 20	< 8	< 10	0.20	
Sulphate	(mg/Kg)	2120	1520	1890	0.10	
<u>Organic</u>	( /7/Z)	000	99.6	100	0.05	
Acetate	(mg/Kg)	336	336 <b>&lt; 4</b>	198 <b>&lt; 5</b>	0.05	
. Citrate Formate	(mg/Kg)	< 20 243	<b>&lt; 4</b> 155	120	$0.50 \\ 0.05$	
Oxalate	(mg/Kg) (mg/Kg)	$\frac{243}{44.2}$	32.1	45.3	$0.05 \\ 0.05$	
	(mg/Kg)					
Phthalate	(mg/Kg)	< 20	< 4	< 5	0.05	

Characteristic Analysis)		1996 W-26 S	2000 W-26 S	2001 W-26 S	IDL <sup>1</sup>	
Beta/gamma emitters (±10%)						
Iross beta Ni <sup>3</sup> Ni <sup>0</sup> Co	(Bq/g) (Bq/g) (Bq/g)	3.5e+06 <3.0e+01 4.0e+03 5.8e+04	4.0e+06 - - 3.1e+04	4.4e+06 - 3.3e+04	-	
<sup>0</sup> Sr/ <sup>90</sup> Y <sup>9</sup> Tc <sup>29</sup> I <sup>34</sup> Cs	(Bq/g) (Bq/g) (Bq/g)	7.1e+05 1.2e+03 - 1.2e+04	1.0e+06	1.3e+06 1.1e+03 <2.8e+03	-	
<sup>37</sup> Cs <sup>51</sup> Sm <sup>52</sup> Eu <sup>54</sup> Eu	(Bq/g) (Bq/g) (Bq/g) (Bq/g)	8.9e+05 < 5.8e+02 6.4e+05 2.9e+05	8.1e+05 3.9e+05 1.7e+05	6.3e+05 4.6e+05 1.9e+05		
<sup>55</sup> Eu <sup>27</sup> Ac <sup>141</sup> Pu	(Bq/g) (Bq/g) (Bq/g)	6.3e+04 < 9.3e+03 1.5e+04	< 1.8e+04	5.6e+04		
Alpha emitters (±	10%)			ı		
Gross alpha  132 Th 133 U 134 U 135 U 138 U 137 Np 141 Am 144 Cm 150 Cf 152 Cf  Total Pu alpha 1238 Pu 1239 Pu/240 Pu 142 Pu 142 Pu 144 Am (3700)	(Bq/g)	52000 13 10000 180 4.0 240 2 3900 28000 < 100 < 100 7600 5300 2300	48000 18 10500 640 12 460 - 4880 28000 - 9100 5200 3900	68000 18 8200 210 8.8 370 - 5000 40000 - 13000 8300 5100		
Uranium isotopics by ICP-MS ( ±2%)						
233U 234U 235U 236U 238U 233U/MS 235U/MS 238U/235U FEM	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %) (mg/Kg) (mg/Kg)	0.152 0.004 0.296 0.006 99.543 28.9 56.7 202	$0.0814 \\ 0.0076 \\ 0.4800 \\ < 0.0001 \\ 99.4310$ $29.4 \\ 175 \\ 171$	0.0773 $0.003$ 1 $0.4157$ $0.0040$ $99.4999$ $22.9$ $124$ $194$	0.001 0.001 0.001 0.001 0.001	

Characteristic (Analysis)		1996 W-26 S	2000 W-26 S	2001 W-26 S	IDL'		
Plutonium isoto	Plutonium isotopics by TIMS (±1%)						
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %)	1.23 82.27 15.11 0.57 0.81 < 0.01			-		
Pu activity <sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	5400 1300 890 15000 0.8 < 0.1	-				
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu	(ng/g)	700 5730	-	765			

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>29</sup>Pu.

characteristic (Analysis)		1996 W-27 S	<b>2000</b> W-27 S	<b>2001</b> W-27 S	<b>2001</b> W-27 H	IDL'
Physical propertie	es and mi	scellaneous data				2012 XI.0014 1002
Request number Sample number oH		<b>7749F</b> 960904-248 12.3	10224 000509-004 10.0	12057 010716-013 8.8	12057 010716-014 11.7	
TSb ( TSSc ( TDSd (	(mg/g) (mg/g) (mg/g) (mg/g) (g/mL)	54.9 451	68.9 311 270 41.4	72.7 273	58.5 415	
TC°	(g/HL) (mg/Kg) (mg/Kg) m	1.44 12400 10000 2400	1.246 10100 < 1000 10100	1.169 6200 3900 2300	1.340 6800 2700 4000	15 15 15
RCRA Metals (±1	10%)					
As (100)  Ba (2000)  Cd (20)  Cr (100)  Hg (4)  Ni (1000)  Pb (100)  Se (20)	(mg/Kg)	<ul> <li>1.8</li> <li>1.4</li> <li>41.8</li> <li>14.8</li> <li>55.3</li> <li>29.0</li> <li>48.9</li> <li>157</li> <li>1.4</li> <li>1.4</li> </ul>	<1 64.2 <16 132 196 84.0 317 <1 <1	<1.6 94.6 19.3 267 118 73.9 804 <0.7 <4	<1.4 58.1 24.4 83.0 36.2 37.4 200 <0.7 <3	0.006 0.011 0.001 0.111 0.008 0.011 0.065 0.341 0.005
Process metals (±	:10%)					
B Be Ca co cu Cs' Fe K Mg Mn Mo Na P Sb Si k Sr Th Ti U	(mg/Kg)	2250 5.98 1.10 43700 2.57 14.2 0.892 935 6970 7820 65.4 58200 1000 37.4 3860 107 1296 11700 3.31	7640 11.9 17.7 26700 16.8 66.3 1.81 3780 4880 4800 387 98.5 30900 <50 6270 163' 17400 29500 <1.3	8580 4.70 8.67 23400 26.2 71.1 < 4 3670 4560 3670 314 70.8 23700 < 4 7750 87.8 6400 53.1 34300 < 1	5380 3.07 2.32 54100 28.5 22.0 <3 1770 7640 5500 65.7 35.0 52300 <3 13900 122 2250 209 3000 <3.6	0.048 0.024 0.001 0.017 0.060 0.006 0.005 0.233 0.178 0.053 0.001 0.030 0.065 0.13 0.509 0.022 0.001 0.181 0.010 0.105 0.105

Characteristic (Analysis)		1996 W-27 S	2000 W-27 S	2001 W-27 S	2001 W-27 H	IDL'		
Semi-quantitativ	Semi-quantitative metals by ICP-MS (±30-50 %, * indicates data from water leach)							
Au, gold Bi, bismuth Ce, cerium Er, erbium Eu, europium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	0.62 130 7.2 0.12 0.80		34.5	22.1			
Ga, gallium Gd, gadolinium Ho, holmium	(mg/Kg) (mg/Kg) (mg/Kg)	4.2 1.9 1.6		4 4	8.56			
I, iodine La, lanthanum Li, lithium <b>Mo,molybdenum</b>	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	* 6.8 7.3 * 53 * 2.0		218 227	88.8 116			
Nb, niobium Rb, rubidium Sn, tin Ti, titanium W, tungsten Zr, zirconium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	0.56 * 1.2 4.0 99 1.3 4.0			223			
,			l nsh of sludge ( ±1	10%)				
Inorganic Bromide Chloride Chromate Fluoride Nitrate Nitrite Phosphate. Sulphate Organic	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) m	<50 2280 <20 <50 210000 2283 <20 549	<ul> <li>&lt;41</li> <li>1190</li> <li>16.2</li> <li>246</li> <li>62500</li> <li>2600</li> <li>12.1</li> <li>943</li> </ul>	140 2450 18.8 20.7 233000 4380 < 10 530	37.8 688 37.5 235 45400 2460 < 10 1040	0.05 0.05 0.05 0.05 0.10 0.10 0.20 0.10		
Acetate Citrate Fonnate Oxalate Phthalate	(mg/Kg) (mg/Kg) (mg/Kg)	196 < 20 200 16.0 < 20	272 i4.1 77.8 228 < 4.1	448 < 5 174 21.2 < 5	42.3 <b>&lt; 4</b> 21 42.3 <b>&lt; 4</b>	0.05 0.50 0.05 0.05 0.05		

characteristic (Analysis)		1996 W-27 S	2000 W-27 S	2001 W-27 S	2001 W-27 H	IDL'			
Beta/gamma ei	Beta/gamma emitters (±10%)								
Gross beta  Ni	(Bq/g) (Bq/g)	1.6e+06 < 2.0e+01	1 <b>.0e+07</b>	5.3e+06	2.0e+06				
<sup>33</sup> Ni <sup>30</sup> Co <sup>30</sup> Sr/ <sup>30</sup> Y <sup>39</sup> Tc	(Bq/g) (Bq/g) (Bq/g)	1.7e+03 1.2e+04 4.5e+05 8.7e+01	9.9e+03 4.0e+06	7.1e+03 2.1e+06 <2.3e+02	4.93+03 6.5e+05 4.3e+02				
129 <sub>T</sub> 134 <sub>Cs</sub> 137 <sub>Cs</sub>	(Bq/g) (Bq/g) (Bq/g) (Bq/g)	< 8.1e+02 3.9e+05	6.0e+05	<1.4e+03 6.6e+05	<1.1e+03 4.2e+05				
151Sm 152Eu 154Eu	(Bq/g) (Bq/g) (Bq/g)	< 5.7e+02 4.1e+04 1.7e+04	2.2e+05 <4.2e+04	9.9e+04 2.9e+04	2.3e+04 7.8e+03				
<sup>155</sup> Eu <sup>227</sup> Ac <sup>241</sup> Pu	(Bq/g) (Bq/g) (Bq/g)	< 2.7e+03 < 6.2e+03 6.5e+03	< 1.8e+04	5.1e+03	< 4.0e+03				
Alpha emitters	(±10%)								
Gross alpha  232Th 233U 234U 235U 238U 237Np 241Am 244Cm 250Cf 252Cf  Total Pu alpha 238Pu 239Pu/240Pu 242Pu  TRU activity Pu+Am (3700)	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	26000 5.2 1000 53 2.5 145 12 2800 17000 < 100 < 100 3400 2200 1200	110000 71 6200 380 11 360 11600 77000 16000 10000 5800	62000 26 1360 343 17.2 424 6600 42000 11000 6000 4900	24000 9.1 396 < 1 0.8 37.2 4000 14500 4300 2500 1800				
Uranium isoto	pics by ICP-	MS ( ±2%)	T						
233 U 234 U 235 U 236 U 238 U	(atom %) (atom %) (atom %) (atom %) (atom %)	0.025 0.002 0.308 0.006 99.660	0.0598 0.0057 0.5068 < 0.0001 99.4276	0.0113 0.0044 0.6360 0.0001 99.3481	0.0378 0.0000 0.3454 0.0000 99.6168	0.001 0.001 0.001 0.001 0.001			
<sup>233</sup> U/MS <sup>235</sup> U/MS <sup>238</sup> U/ <sup>235</sup> U FEM	(mg/Kg) (mg/Kg)	2.86 35.6 296	17.3 148 1 7 2	3.79 215 155	1.11 $10.2$ $255$				

at					postuter A	Application of the second of t
Characteristic (Analysis)	:	1996 W-27 S	2000 W-27 S	2001 W-27 S	2001 W-27 H	IDL'
Plutonium isot	topics by TIM	IS ( ±1 %)	• rum • rol	Upport THE NOTE	nor 8888 norma superformational se	gride • • • where we
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom%) (atom%) (atom%) (atom%) (atom%) (atom%)	1.08 84.88 12.64 0.49 0.91 <0.01				
Pu activity 238Pu 239Pu 240Pu 241Pu 242Pu 244Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	2400 670 370 6500 0.5 co.1				
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu	(ng/g)	350 4390		1336	1201	

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>239</sup>Pu.

Table 6 Analytical Data for Sludge in Tanks W-28

Table 6 Aı	able 6 Analytical Data for Sludge in Tanks W-28								
Characteristic (Analysis)		<b>1996</b> W-28 s	2000 W-28 s	2001 W-28 S	IDL'				
Physical properties	s and miscellar	neous data		, , , , , , , , , , , , , , , , , , , ,					
Request number Sample number pH		7749B 960724-060 12.3	10224 000509-005 11.0	12057 010716-015 8.7					
Water <sup>a</sup> TS <sup>b</sup> TSS <sup>c</sup> TDS <sup>d</sup> Bulk density TC <sup>c</sup> TIC'	(%) (mg/g) (mg/g) (mg/g) (g/mL) (mg/Kg) (mg/Kg)	47.3 $527$ $1.37$ $12800$ $10200$	61.7 383 266 117 1.306 8000 < 1000	65.4 346 1.190 8900 5300	15 15				
TOC <sup>g</sup>	(mg/Kg)	2600	8000	3600	15				
RCRA Metals (±1	.0%)								
Agh (100) <sup>1</sup> As (100) Ba (2000) Cd (20) Cr (100) Hg (4) Ni (1000) Pb (100) Se (20) Tl (18)	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	< 1.8 < 5.0 43.3 <b>24.9</b> 54.8 6.55 53.6 195 < 5.0 5.97	<1 64.3 17.7 170 50.6 66.5 396 <1 <1	< 2.5 76.9 21.9 277 81.7 77.5 758 < 1 < 6	0.006 0.011 0.001 0.111 0.008 0.011 0.065 0.341 0.005 0.005				
Process metals (±	10%)								
Al B Be Ca c o Cu Cs <sup>j</sup> Fe K Mg Mn Mo Na P Sb Si k Sr Th Ti U V Zn	(mg/Kg)	571 7.33 1.36 45800 3.53 28.0 0.480 599 14600 14500 91.0 61000 907 < 18 1080 151 1360 18500 1.54 278	3860 7.14 7.62 44600 23.9 57.0 0.401 2180 8860 8760 339 28.0 37200 <50 3000 175 4710 31200 <1.3 368	6980 < 2.7 7.54 37600 36.3 70.6 < 6 3270 7900 7380 379 54.0 32000 < 6 5620 139 4820 30.0 41900 < 1.5 292	0.048 0.024 0.001 0.017 0.060 0.006 0.005 0.233 0.178 0.053 0.001 0.030 0.065 0.13 0.509 0.022 0.001 0.181 0.010 0.105 0.105 0.1010				

Characteristic (Analysis)		1996 W-28 s	2000 W-28 S	2001 W-28 S	IDL'
Semi-quantitative	metals by ICP-	MS (±30-50 %, * ii	ndicates data from w	vater leach)	
Au, gold Bi, bismuth Ce, cerium Er, erbium Eu, europium Ga, gallium Gd, gadolinium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	1.9 $12$ $7.9$ $0.07$ $1.5$ $3.1$ $6.0$	- - - - -	3.42 <b>&lt; 6</b>	-
Ho, holmium I, iodine La, lanthanum Li, lithium Mo,molybdenum Nb, niobium Rb, rubidium Sn, tin Ti, titanium W, tungsten Zr, zirconium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	* 170 * 2.3 0.30 * 1.9 5.9 4.1 1.4 1.8	- - - - - -	212 172 207	
Anions by ion chr	omatography i	n water wash of slud	lge ( ±10%)		
Inorganic Bromide Chloride Chromate Fluoride Nitrate Nitrite Phosphate	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	< 50 3460 < 20 < 50 248000 1120 < 20	<44 1580 10.7 70.6 98800 1750 < 8.8	57.3 1210 42.5 277 73500 2400 < 10	0.05 0.05 0.05 0.05 0.10 0.10 0.20
Sulphate Organic Acetate Citrate Formate Oxalate Phthalate	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)	$   \begin{array}{r}     325 \\     < 20 \\     271 \\     19.1 \\     < 20   \end{array} $	973 349 < 4.4 90.1 11.8 < 4.4	988 70.1 < 5 5.6 119 < 5	0.10 0.05 0.50 0.05 0.05 0.05

Characteristic (Analysis)		1996 W-28 S	2000 <b>W-28 S</b>	2001 <b>W-28 S</b>	IDL'
Beta/gamma emit	ters ( <b>±10%)</b>		*/	<i>μ</i>	
Gross beta <sup>59</sup> Ni <sup>53</sup> Ni <sup>50</sup> Co <sup>50</sup> Sr/ <sup>50</sup> Y <sup>57</sup> Tc <sup>129</sup> I <sup>134</sup> Cs <sup>137</sup> Cs <sup>151</sup> Sm <sup>152</sup> Eu <sup>155</sup> Eu <sup>155</sup> Eu	(Bq/g)	3.1e+06 <2.5e+01 3.3e+03 4.2e+04 7.0e+05 1.2e+02 4.1e-02 <1.2e+03 3.1e+05 <5.6e+02 8.0e+05 2.7e+05 7.0e+04 <6.7e+03	5.0e+06  1.5e+04 1.7e+06  - 4.8e+05 5.2e+05 1.4e+05 < 2.0e+04	4.8e+06  4.8e+03 1.8e+06 <3.6e+02 <9.3e+02 2.7e+05  1.2e+05 3.2e+04 7.8e+03	-
<sup>241</sup> Pu	(Bq/g)	1.2e+04			
Alpha emitters (:	±10%) —(Bq/g)——				
Gross alpha 232Th 233U 234U 235U 238II 237Np "'A m 244Cm 250Cf 252Cf  Total Pu alpha 238Pu 239Pu/240Pu 242Pu  TRU activity Pu+Am (3700)	(Bq/g)	44000 5.5 5200 130 3.8 230 16 4600 25000 < 100 < 100  4400 2700 1700  9000	66000 19 4800 360 11 390 7600 43000 11000 6400 4600	58000 20 1680 338 20.3 518 6200 35000 13000 7400 6100	-
Uranium isotopic	s by ICP-MS (:	±2%)			
<sup>233</sup> U <sup>234</sup> U <sup>235</sup> U <sup>236</sup> U <sup>238</sup> U/MS <sup>235</sup> U/MS <sup>238</sup> U/ <sup>235</sup> U FEM	(atom %) (atom %) (atom %) (atom %) (atom %) (mg/Kg) (mg/Kg)	0.08 1 0.003 0.296 0.007 99.613 14.7 54.1 249	0.0441 0.005 1 0.4827 0.0033 99.4647 13.5 149 186	0.0115 $0.0036$ $0.6141$ $0.0001$ $99.3707$ $4.72$ $254$ $160$	0.001 0.001 0.001 0.001 0.001

Characteristic (Analysis)		1996 W-28 S	2000 W-28 S	2001 W-28 S	IDL'
Plutonium isote	ppics by TIMS (±	1%)			
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %)	< 1.06 <b>81.54</b> 15.93 0.70 0.76 0.01			
Pu activity <sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	3000 830 600 12000 0.5 < 0.1			
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu	(ng/g)	440 3750		851	

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>259</sup>Pu.

Table 7 Analytical Data for Sludge in Tanks W-29

Table 7 A Characteristic (Analysis)		Data for Sludge 1996 W-29 S	2000 W-29 S	2001 W-29 S	IDL¹
Physical propertie	es and miscell	aneous data		regulation of Egypt States and the support of the support	
Request number Sample number pH		No Data	No Data	12057 <b>010716-016</b> 12.0	-
Water <sup>a</sup> TS <sup>b</sup> TSS <sup>c</sup> TDS <sup>d</sup>	(%) (mg/g) (mg/g)			56.5 435	<b>-</b>
Bulk density TC" TICf TOCg	(mg/g) (g/mL) (mg/Kg) (mg/Kg) (mg/Kg)			$ \begin{array}{c} 1.336 \\ 12000 \\ 4100 \\ 7700 \end{array} $	15 15 15
RCRA Metals (±					
Agh (100) <sup>i</sup> As (100) Ba (2000) Cd (20) Cr (100) Hg (4) Ni (1000) Pb (100) Se (20) Tl (18)	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			<1.6 86.0 30.7 149 40.0 71.8 434 <0.7 <4	0.006 0.011 0.001 0.111 0.008 0.011 0.065 0.341 0.005 0.005
Process metals (:	±10%)	· 17 19 18 0 间 到10 19 20 14 14 15 15 14 15 15 16 16 16 16 16 16 16 16 16 16 16 16 16			
Al B Be Ca c o c u Cs <sup>i</sup> Fe K Mg Mn	(mg/Kg)			4060 2.06 5.08 73700 40.2 48.4 1740 8040 10800 247	0.048 0.024 0.001 0.017 0.060 0.006 0.005 0.233 0.178 0.053 0.001
Mo Na P Sb Si k Sr Th Ti U V	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			50.4 37900 < 4 2940 287 4860 36.1 24100 < 1	0.030 0.065 0.13 0.509 0.022 0.001 0.181 0.010 0.105 0.013
Zn Zr	(mg/Kg) (mg/Kg)			485 187	0.390 0.008

Characteristic (Analysis)		1982 W-29 S	2005 2005 2005	2001 w-29 s	IDL'
Semi-quantitative	metals by ICP-	-MS (±30-50 %, * i	ndicates data from w	vater leach)	
Au, gold Bi, bismuth Ce, cerium Er, erbium Eu, europium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			1.76 <del>-</del>	
Ga, gallium Gd, gadolinium Ho, holmium I, iodine	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			4.34	
La, lanthanum Li, lithium Mo,molybdenum Nb, niobium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			60.7 117	
Rb, rubidium Sn, tin Ti, titanium W, tungsten Zr, zirconium	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			< 4	
Anions by ion chr	omatography i	in water wash of sluc	lge ( ±10%)		
Inorganic Bromide Chloride Chromate Fluoride Nitrate Nitrite Phosphate Sulphate Organic	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			$ \begin{array}{c} 120 \\ 1380 \\ 26 \\ 28.9 \\ 106000 \\ 1920 \\ < 10 \\ 918 \end{array} $	0.05 0.05 0.05 0.05 0.10 0.10 0.20 0.10
Acetate Citrate Formate O x a l a t e Phthalate	(mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg) (mg/Kg)			267 < 4 99.1 40.6 < 4	0.05 0.50 0.05 0.05 0.05

Characteristic (Analysis)		1996 W-29 S	2000 W-29 S	2001 w-29 s	IDL'			
Beta/gamma emitters (±10%)								
Gross beta  59Ni  63Ni	(Bq/g) (Bq/g)			4.1e+06				
%CO %Sr/%Y %Tc	(Bq/g) (Bq/g) (Bq/g) (Bq/g)			2.5e+04 1.4e+06 9.3e+02				
<sup>129</sup> I <sup>134</sup> Cs <sup>137</sup> Cs	(Bq/g) (Bq/g) (Bq/g)	And the second s		3.3e+03 5.8e+05				
151Sm 152Eu 154Eu 155Eu <sup>24</sup> AC <sup>241</sup> Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	Service Commission (Commission Commission Co		1.5e+05 4.8e+04 1.8e+04				
Alpha emitters (±		Leganitinal cultimate session archetest singlich beiteit.		•				
Gross alpha  132Th 133U 134U 135U 138U 137Np 141Am 144Cm 150Cf 152Cf  Total Pu alpha 238Pu 239Pu/240Pu 242Pu  TPLL activity	(Bq/g)			64000 20 2310 32.5 9.7 298 7500 42000 - - 11000 6600 4500				
TRU activity Pu+Am (3700)	(Bq/g)	10年 20年 20年 20年 20年 20年 20年 20年 20年 20年 2	Gu	18500				
Uranium isotopics	s by ICP-MS (			Т				
<sup>233</sup> U <sup>234</sup> U <sup>235</sup> U <sup>236</sup> U <sup>238</sup> U	(atom %) (atom %) (atom %) (atom %) (atom %)			$\begin{array}{c} 0.0274 \\ 0.0006 \\ 0.5093 \\ 0.0007 \\ 99.4620 \end{array}$	0.001 0.001 0.001 0.001 0.001			
<sup>233</sup> U/MS <sup>235</sup> U/MS <sup>238</sup> U/ <sup>235</sup> U FEM	(mg/Kg) (mg/Kg)		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	6.47 121 185				

Characteristic (Analysis)		1996 W-29 S	2000 W-29 S	2001 W-29 S	IDL <sup>1</sup>
Plutonium isotoj	oics by TIMS (	±1%)			
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %)			-	- - - - - - -
Pu activity 238Pu 239Pu 240Pu 241Pu 242Pu 244Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)			- - - - -	- - - - -
( <sup>239</sup> Pu) <sup>232</sup> T'n <sup>/239</sup> Pu	(ng/g)			- 1014	- :

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>25</sup>Pu.

Table 8 Analytical Data for Sludge in Tanks W-30

Characteristic (Analysis)	anaiyucai L	Data for Sludge 1996 W-30 S	2000 W-30 S	2001 w-30 s	IDL'
Physical proper	ies and miscella	neous data			e deeme 🗡
Request number Sample number pH		No Data	No Data	12057 010716-017 10.7	
Water <sup>a</sup> TS <sup>b</sup> TSS <sup>c</sup>	(%) (mg/g) (mg/g)			55.0 450	
TDS <sup>d</sup> Bulk density	(mg/g) (g/mL)			1.411	<u>.</u> .
TC <sup>e</sup>	(mg/Kg)			13600	15
TIC' T <b>OC</b> <sup>g</sup>	(mg/Kg) (mg/Kg)			$5200 \\ 8500$	15 15
RCRA Metals (	±10%)	· · · · · · · · · · · · · · · · · · ·	的用 <b>以相互分别性。对</b> 最初 <b>的时候</b>		duy .
<b>Ag</b> <sup>h</sup> (100)'	(mg/Kg)			-	0.006
As (100)	(mg/Kg)			< 2	0.011
Ba (2000) Cd <b>(20)</b>	(mg/Kg) (mg/Kg)			$\frac{119}{15.2}$	0.001 0.111
Cr (100)	(mg/Kg)			189	0.008
Hg (4)	(mg/Kg)			67.3	0.011
Ni (1000)	(mg/Kg)			66.5	0.065
Pb (100)	(mg/Kg)			915	0.341
Se (20)	(mg/Kg)			. < 0.9	0.005
Tl (18)	(mg/Kg)			< 5	0.005
Process metals (	······································	The state of the s			
Al	(mg/Kg)			5900	0.048
B Be	(mg/Kg)			$\frac{4.24}{7.00}$	0.024
Са	(mg/Kg) (mg/Kg)			43800	$0.001 \\ 0.017$
c o	(mg/Kg)			43600 33.6	0.060
cu	(mg/Kg)			52.9	0.006
Cs <sup>i</sup>	(mg/Kg)			< 5	0.005
Fe	(mg/Kg)			2740	0.233
K	(mg/Kg)			10400	0.178
Mg	(mg/Kg)			6670	0.053
Mn Mo	(mg/Kg) (mg/Kg)			277	0.001 0.030
Na	(mg/Kg)			$63.9 \\ 48000$	0.030 0.065
P	(mg/Kg)			40000	0.003
Sb	(mg/Kg)			< 5	0.509
Si <sup>k</sup>	(mg/Kg)			10000	0.022
$\operatorname{Sr}$	(mg/Kg)			189	0.001
Th	(mg/Kg)			5740	0.181
	(mg/Kg)			528	0.010
Ti	imar(Ka)	<ul> <li>Care that anniverse in the colors</li> </ul>	<ul> <li>Description of the NACE AND Margin Edition</li> </ul>	26300	0.105
U	(mg/Kg)				
	(mg/Kg) (mg/Kg) (mg/Kg)			< 1.2 262	$0.013 \\ 0.390$

Characteristic (Analysis)		1996 W-30 S	2000 W-30 S	2001 W-30 S	IDL'				
Semi-quantitative metals by ICP-MS (±30-50 %, * indicates data from water leach)									
Au, gold	(mg/Kg)			-					
Bi, bismuth Ce, cerium	(mg/Kg) (mg/Kg)			2.60					
Er, erbium	(mg/Kg)	i de la companya de		2.00					
Eu, europium	(mg/Kg)				-				
Ga, gallium	(mg/Kg)								
Gd, gadolinium	(mg/Kg)			< 5					
Ho, holmium	(mg/Kg)								
[, iodine	(mg/Kg)								
La, lanthanum	(mg/Kg)			155	•				
Li, lithium	(mg/Kg)			148					
Mo,molybdenum Nb, niobium	(mg/Kg) (mg/Kg)								
Rb, rubidium	(mg/Kg)			775					
Sn, tin	(mg/Kg)			. , ,					
Γi, titanium	(mg/Kg)	LONG BUTTON STATE OF	50-8 PM						
W, tungsten	(mg/Kg)								
Zr, zirconium	(mg/Kg)								
Anions by ion chr	omatography	in water wash of slu	dge ( ±10%)						
Inorganic					•				
Bromide	(mg/Kg)	i de la compania de La compania de la co		135	0.05				
Chloride	(mg/Kg)			2350	0.05				
Chromate	(mg/Kg)			21 ~	0.05				
Fluoride	(mg/Kg)			199	<b>0.05</b> 0.10				
Nitrate	(mg/Kg)			159000 <b>5620</b>	0.10				
Nitrite Phosphate	(mg/Kg) (mg/Kg)			< 5	0.10				
Sulphate	(mg/Kg)			2210	0.10				
Organic	(6								
Acetate	(mg/Kg)			531	0.05				
Citrate	(mg/Kg)			190	0.50				
Formate	(mg/Kg (mg/Kg <sub>3</sub>	* 100 miles	16	118	0.05				
Oxalate	(mg/Kg) (mg/Kg)	\$6, 6950; 56,545 (6,556) 5, 56,57,57,57,57,57,57,57,57,57,57,57,57,57,		973	0.05				
Phthalate	(mg/r/g)	The first of meaning two marines were		< 3	0.05				

Characteristic Analysis)		1996 W-30 S	2000 W-30 S	2001 W-30 S	IDL¹
Beta/gamma emitte	rs (±10%)				
Gross beta <sup>59</sup> Ni <sup>53</sup> Ni	(Bq/g) (Bq/g) (Bq/g)			6.5e+06 -	•
<sup>10</sup> Co <sup>10</sup> Sr/ <sup>90</sup> Y <sup>19</sup> Tc	(Bq/g) (Bq/g) (Bq/g)			1.6e+04 1.9e+06 5.8e+02	
<sup>129</sup> I <sup>134</sup> Cs <sup>137</sup> Cs <sup>151</sup> Sm	(Bq/g) (Bq/g) (Bq/g)			- 1.9e+04 1.9e+06	-
152Eu 154Eu 155Eu 227Ac 241Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)			7.4e+04 2.5e+04 <3.8e+03	-
Alpha emitters (±1					
Gross alpha  232Th 233U 234U 235U 238U 237Np 241Am 244Cm 250Cf 252Cf  Total Pu alpha  238Pu 239Pu/240Pu 242Pu	(Bq/g)			55000 23 1940 301 10.9 325 - 5500 34000 - 12000 '6200 6100	-
TRU activity Pu+Am (3700)	(Bq/g)			17500	
Uranium isotopics	by ICP-MS (	±2%)			
238U 238U 238U 238U	(atom %) (atom %) (atom %) (atom %) (atom %)			0.0211 <b>0.005</b> 1 0.5239 <b>0.0018</b> 99.4482	0.001 0.001 0.001 0.001 0.00 <b>1</b>
<sup>233</sup> U/MS <sup>235</sup> U/MS <sup>238</sup> U/ <sup>235</sup> U FEM	(mg/Kg) (mg/Kg)			5.44 136 182	

Characteristic (Analysis)		1996 W-30 S	2000 W-30 S	2001 W-30 S	IDL <sup>1</sup>
Plutonium isotop	ics by TIMS (±	:1%)			
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom %) (atom %) (atom %) (atom %) (atom %) (atom %)				
Pu activity  238Pu 239Pu 240Pu 241Pu 242Pu 244Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)		Part Performance   1 25 pm	- - - - -	
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu	(ng/g)			1098	

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids, (d) Total dissolved solids, (e) Total carbon, (f) Total inorganic carbon, (g) Total organic carbon, (h) nitric-hydrochloric acid prep., (i) RCRA regulatory limits, (j) measured by ICP-MS or GFAA, (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 assumes all alpha activity is from <sup>25</sup>Pu.

Table 9	Analytica	I-Data for S	<u>ludge in Tan</u>	ks W-31
Characteris	stic	1996	2000	2001
(Analysis)		w-31 s	w-31 s	w-31 s

Characteristic (Analysis)	1996 w-31 s	2000 w-31 s	2001 w-31 s	2001 W-31 H	IDL'				
Physical properties and miscellaneous data									
Request number Sample number pH	7749A 960717-023 9.9	10224 000509-006 9.7	12057 010716-018 9.3	12057 010716-019 9.3					
Water <sup>a</sup> (%) TS <sup>b</sup> (mg/g) TSS <sup>c</sup> (mg/g) TDS <sup>d</sup> (mg/g)	51.4 486	72.5 $275$ $196$ $79.2$	58.5 415	64.3 357					
Bulk density (g/mL) TC' (mg/Kg) TIC' (mg/Kg) TOC <sup>g</sup> (mg/Kg)	* 1.44 10200 5300 4900	1.210 5690 < 1000 5690	1.331 14000 9200 4800	1.283 11000 8700 2200	15 15 15				
RCRA Metals (±10%)									
Agh         (100)i         (mg/Kg)           As         (100)         (mg/Kg)           Ba         (2000)         (mg/Kg)           Cd         (20)         (mg/Kg)           Cr         (100)         (mg/Kg)           Hg         (4)         (mg/Kg)           Ni         (1000)         (mg/Kg)           Pb         (100)         (mg/Kg)           Se         (20)         (mg/Kg)           T1         (18)         (mg/Kg)	< 1.9 < 5.0 124 9.03 130 70.7 104 764 < 5.0 < 5.0	<0.9 108 <15 237 65.4 70.3 717 <0.9 <0.9	<1.3 100 148 182 70.2 89.1 933 < 0.6 < 3	<2.3 119 <12 325 76.0 74.0 1450 <1 <5	0.006 0.011 0.001 0.111 0.008 0.011 0.065 0.341 0.005 0.005				
Process metals (±10%)	· ,		<u>'</u>						
Al (mg/Kg) B (mg/Kg) Be (mg/Kg) Ca (mg/Kg) co (mg/Kg) cu (mg/Kg) Fe (mg/Kg) K (mg/Kg) Mg (mg/Kg) Mn (mg/Kg) Mo (mg/Kg) P (mg/Kg) Sb (mg/Kg) Si k (mg/Kg) Th (mg/Kg) U (mg/Kg) U (mg/Kg) Zr (mg/Kg)	$\begin{array}{c} 12700 \\ 11.6 \\ 21.0 \\ 24100 \\ 4.76 \\ 80.2 \\ 0.543 \\ 2820 \\ 8320 \\ 2170 \\ 247 \\ \\ 60600 \\ 4240 \\ < 19 \\ 10200 \\ 174 \\ 20700 \\ \\ 19800 \\ 7.18 \\ 125 \\ \end{array}$	7920 7.25 8.91 24700 14.1 64.3 0.395 3420 4570 2090 268 46.7 33700 <45 7930 88.3 8350 38600 <1.1 92.2	8700 5.55 15.6 43000 30.3 83.8 <3 2760 7330 4690 270 110 46100 <3 7940 198 13300 22.3 24200 <1 187 437	7830 <2.6 6.79 52100 36.1 70.7 <5 3960 5660 2330 262 5 3 . 1 34900 <5 7240 351 5470 26.0 34000 <1.4 129 202	0.048 0.024 0.001 0.017 0.060 0.006 0.005 0.233 0.178 0.053 0.001 0.030 0.065 0.13 0.509 0.022 0.001 0.181 0.010 0.105 0.013 0.390 0.098				

Characteristic (Analysis)		1996 w-31 s	2000 w-31 s	2001 w-31 s	2001 W-31 H	IDL'
Semi-quantitati	ive metals by	V ICP-MS (±30	-50 %, * indicat	es data from wa	ter leach)	
Au, gold	(mg/Kg)	2.6				
Bi, bismuth	(mg/Kg)	1200				
Ce, cerium	(mg/Kg)	20		278	40.3	
Er, erbium	(mg/Kg)	0.85				
Eu, europium	(mg/Kg)	0.54				
Ga, gallium	(mg/Kg)	12				
Gd, gadolinium	(mg/Kg)	0.75		< 4	<b>&lt;</b> 5.3	
Ho, holmium	(mg/Kg)	0.22				
I, iodine	(mg/Kg)	* 20				
La, lanthanum	(mg/Kg)	54		146	272	
Li, lithium	(mg/Kg)	* 81		411	178	
Mo,molybdenum	(mg/Kg)	* 1.4				
Nb, niobium	(mg/Kg)	2.0				
Rb, rubidium	(mg/Kg)	* 1.3				
Sn, tin	(mg/Kg)	40		< 3	233	
Ti, titanium	(mg/Kg)	34				
W, tungsten	(mg/Kg)	1.3				
Zr, zirconium	(mg/Kg)	51				
Anions by ion	chromatogra	phy in water w	ash of sludge (±	. 10%)		
Inorganic						
Bromide	(mg/Kg)	< 50	< 50	78.9	63.8	0.05
Chloride	(mg/Kg)	2570	817	1610	1220	0.05
Chromate	(mg/Kg)	51.5	21.9	46.8	85.3	0.05
Fluoride	(mg/Kg)	125	604	370	920	0.05
Nitrate	(mg/Kg)	197000	51900	148000	89800	0.10
Nitrite	(mg/Kg)	3470	2680	4710	4140	0.10
Phosphate	(mg/Kg)	< 50	169	< 10	103	0.20
Sulphate	(mg/Kg)	1090	919	1240	1140	0.10
<u>Organic</u>						
Acetate	(mg/Kg)	237	< 50	149	191	0.05
Citrate	(mg/Kg)	<b>&lt;</b> 50	< 5	15.1	< 5	0.50
Formate	(mg/Kg)	251	< 50	93.4	15.3	0.05
Oxalate	(mg/Kg)	89.8	107	261	162	0.05
Phthalate	(mg/Kg)	< 50	< 5	< 5	< 5	0.05

Characteristic (Analysis)		1996 w-31 s	2000 w-31 s	2001 w-31 s	2001 W-31 H	IDL'				
Beta/gamma emitters (±10%)										
Gross t s  59Ni 63Ni	(Bq/g) (Bq/g) (Bq/g)	2.4e+07 < 3.3e+01 4.4e+03	6.4e+06	1.4e+07	7.6e+06					
<sup>60</sup> Co <sup>90</sup> Sr/ <sup>90</sup> Y <sup>99</sup> Tc	(Bq/g) (Bq/g) (Bq/g)	2.2e+04 1.1e+07 1.4e+02	< 6.4e+03 2.6e+06	9.2e+03 6.3e+06 7.3e+02	2.3e+03 3.2e+06 <3.3e+02					
129 <sub>T</sub> 134 <sub>C</sub> s 137 <sub>C</sub> s	(Bq/g) (Bq/g) (Bq/g)	4.5e = 02 2.5e+03 4.3e+05	5.7 <b>e</b> +05	9.8 <del>c+</del> 02 4.6 <del>c+</del> 05	<1.6e+03 6.4e+05					
151Sm 152Eu 154Eu 155Eu 227Ac 241Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	<6.0e+02 3.0e+04 2.0e+04 <3.4e+03 <5.8e+03 2.4e+04	<3.0e+04 <1.9e+04 <1.1e+04	4.9e+04 1.9e+04 1.0e+04	1.0e+04 <3.6e+03 <3.0e+03	•				
Alpha emitters	1	2,,,,								
Gross alpha 232Th 233U 234U 235U 238U 237Np 241Am 244Cm 250Cf 252Cf  Total Pu alpha 238Pu 239Pu/240Pu 242Pu  TRU activity Pu+Am (3700)	(Bq/g)	160000 84 5200 310 10 244 21 14000 110000 <100 <100 19000 13000 6200	43000 34 1970 860 17 480 2900 29000 9300 5200 4100	92000 54 2100 200 12 300 7900 65000 16000 9700 5800	33000 22 12 180 17 420 3200 19000 8600 4200 4300					
Uranium isoto	pics by ICP-	MS ( ±2%)								
233U 234" <sup>235</sup> U <sup>236</sup> U <sup>238</sup> U	(atom %) (atom %) (atom %) (atom %) (atom %)	0.075 0.007 0.750 0.004 99.165	0.0146 0.0098 0.6333 <0.0001 99.3423	$\begin{array}{c} 0.0250 \\ 0.0036 \\ 0.6460 \\ 0.0000 \\ 99.3254 \end{array}$	$\begin{array}{c} 0.0001 \\ 0.0023 \\ 0.6471 \\ 0.0000 \\ 99.3504 \end{array}$	0.001 0.001 0.001 0.001 0.001				
<sup>233</sup> U/MS <sup>235</sup> U/MS <sup>238</sup> U/ <sup>235</sup> U FEM	(mg/Kg) (mg/Kg)	14.5 150 118	5.52 241 154	5.92 154 148	co.1 217 <b>155</b>					

Characteristic (Analysis)	:	1996 w-31 s	2000 w-31 s	2001 w-31 s	2001 W-31 H	$\mathbf{IDL}_1$
Plutonium isot	topics by TIN	IS (±1%)				
<sup>238</sup> Pu <sup>239</sup> Pu <sup>240</sup> Pu <sup>241</sup> Pu <sup>242</sup> Pu <sup>244</sup> Pu	(atom%) (atom%) (atom%) (atom%) (atom%) (atom%)	< 1.16 81.94 14.55 0.34 1.9 0.11		,	-	•
Pu activity  238Pu 239Pu 240Pu 241Pu 244Pu 244Pu	(Bq/g) (Bq/g) (Bq/g) (Bq/g) (Bq/g)	13000 3400 2200 24000 5.1 < <b>0.1</b>		-		
( <sup>239</sup> Pu) <sup>232</sup> Th/ <sup>239</sup> Pu	(ng/g)	1820 13800	~	1908	1460	

<sup>(</sup>a) Free water content of sludge, (b) Total solids, (c) Total suspended solids (d) Total dissolved, solids, (e) Total carbon, (f) Total inorganic carbon, Total or anic carbon, (h) mitric, hydrololoric act p rep. (i) RCRA regulatory limits, (j) measured by I#-MS or &AA (k) nitric-hydrofluoric acid prep., (l) Instrument detection limits, and (m) the ratio reported for year 2000 ass&es all alpha activity is from <sup>239</sup>Pu.

## 5.2 <u>Discussion of MVST Sludge Characteristics</u>

Determination of the mass and charge balance for a sludge sample typically has a larger error bar than what is observed with a supematant sample. The assumptions required about the chemical form and the oxidation state of the species present in the sludge are not well known, and many of the compounds in the sludge are mixed oxides which are not directly measured. The sludge is actually a slurry with a high water content. The interstitial liquid is in close contact with the sludge, and there are many ionic solubility equilibriums. The anion data for the sludge samples discussed in this report are based on the water soluble anions that would be available to a water wash. The water wash does not account for the insoluble hydroxides, carbonates, and mixed oxides present in a sludge sample. The insoluble species do not contribute to the charge balance, and the cation charge is not used in the calculation, as indicated in Table 10. Most of the nitrate reported for the sludge is due to the interstitial liquid. Considering these limitations, the compounds listed in Table 10 were used to estimate the mass and charge balance.

Table 10 Assumption Used for Major Compounds in MVST Sludge

Cation	Chemical Form	Cation Charge Used	Gravimetric Factors
Al <sup>3+</sup>	$Al_2O_3$	0	1.890
Ca <sup>2+</sup>	CaCO <sub>3</sub>	0	2.497
Fe <sup>3+</sup>	$Fe_2O_3$	0	1.430
K <sup>+</sup>	$K^+NO_3$	+1	2.586
$\mathrm{Mg}^2$	$Mg(OH)_2$	0	2.399
Mg <sup>2</sup> Mn <sup>2+</sup>	$Mn(OH)_2$	0	1.619
Na <sup>+</sup>	Na <sup>+</sup> NO <sub>3</sub> -	+1	3.697
Th⁴+	Th(OH) <sub>4</sub>	0	1.293
$UO_2^{2+}$	UO <sub>2</sub> ((OH) <sub>2</sub> -H <sub>2</sub> O	0	1.353

Table 11 summarizes the mass and charge balance for the MVST tank sludge samples. Considering the limitations of these calculations, the mass balance is within the analytical error (±20%) for these sludge samples. The charge balance is more influenced by the chemical form assumptions, and the results have a larger corresponding error range.

Table 11 Summary of Quality Checks for MVST Sludge Data

Tank	Mass Balance (TS <sub>calc.</sub> /TS <sub>meas.</sub> )	Charge Balance (M <sup>+</sup> /A <sup>-</sup> )	pН	<sup>134</sup> Cs+ <sup>137</sup> Cs (%)	<sup>90</sup> Sr/ <sup>90</sup> Y (%)	Beta Recovery (%)
W-24 S	0.97	0.92	10.0	18.9	64.0	95.6
W-25 S	0.98	0.89	11.8	7.5	84.4	94.3
W-26 S	0.98	0.83	9.4	17.9	63.3	94.9
W-27 S	0.95	0.28	8.8	15.1	82.2	97.6
W-27 H	0.99	2.66	11.7	26.8	71.1	92.6
W-28 S	0.99	1.07	8.7	7.7	88.2	86.0
w-29 s	0.98	0.97	12.0	18.1	75.0	92.5
w-30 s	0.91	0.79	10.7	36.0	61.8	95.5
w-31 s	0.95	0.76	9.3	4.1	95.2	95.2
W-31 H	1.03	0.86	9.3	10.4	89.4	94.6

The beta recovery results are listed in Table 11. As discussed before, the variability for the beta recovery is probably due to the analytical error on the  ${}^{90}$ Sr measurement. Any measurement error for the  ${}^{90}$ Sr activity would be doubled when considering the beta recovery calculation.

The distribution, by weight percent, of the major compounds **from** Table 8 are illustrated in Fig. 1 for each MVST sludge sample. The distribution of the total uranium and thorium concentrations in the MVST sludge samples are shown in Fig. 2, and the change in uranium concentration between sludge sample collected in 1996 and 2000 is shown in Fig. 3.

Figure 1 Distribution of Major Compounds in the MVST Sludge

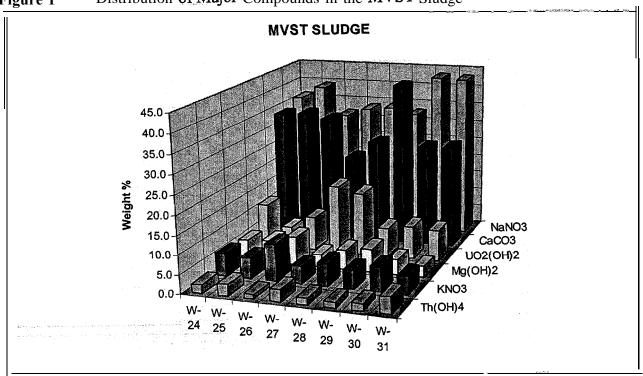


Figure 2 Distribution of Uranium and Thorium in the MVST Sludge

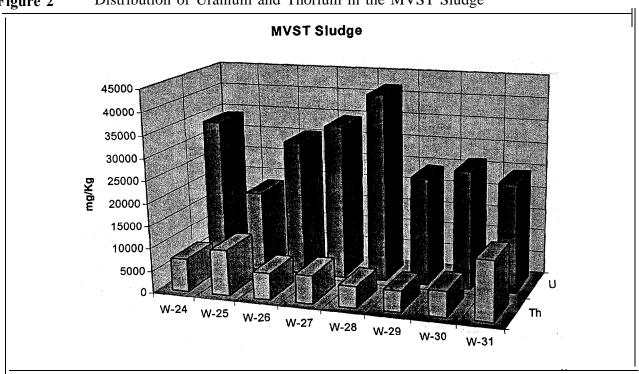
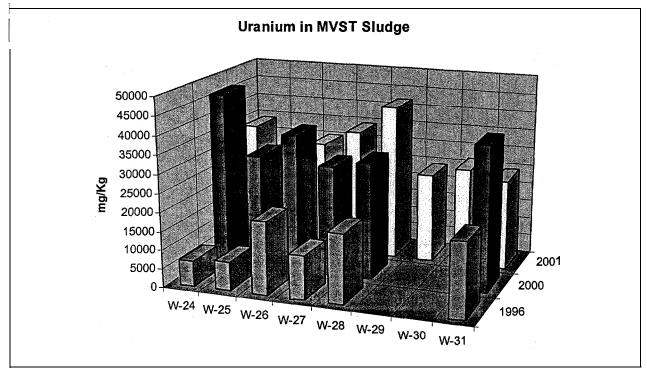


Figure 3 Distribution of Uranium in the MVST Sludge in 1996, 2000, and 2001



The distribution of the major beta emitters found in the MVST sludge samples are summarized in Table 12. The distributions of the beta activity are shown to be dependent upon the radionuclides present, which is a function of the age of the radioactive waste, and the pH of the supernatant found over the sludge. Under the typical basic conditions for ORNL waste tanks, the major difference in the beta distribution between the supernatant and the sludge is that the distribution of the longer lived fission products ( $^{90}$ Sr and  $^{137}$ Cs) are reversed due to the differences in solubility. The Group IA metals ( $^{134}$ Cs and  $^{137}$ Cs) and the radionuclides that form anionic species ( $^{99}$ TcO<sub>4</sub>,  $^{129}$ T, and  $^{129}$ IO<sub>3</sub>) are more soluble in the supernatant. The solubility of the Group IIA metals ( $^{90}$ Sr) in the supernatant are a function of both pH and carbonate concentration. At high pH most of the other metals, lanthanides, and actinide elements form insoluble hydroxides and mixed oxides, which are found in the sludge. The  $^{99}$ Tc activity is higher in the supernatant than the sludge. The source of most of the  $^{99}$ Tc found in the sludge samples was the interstitial liquid, and not insoluble forms of technetium. The shorter lived radionuclides observed include the europium ( $^{152}$ Eu,  $^{154}$ Eu, and  $^{155}$ Eu) isotopes and to some extent  $^{134}$ Cs.

Table 12 Distribution of Beta Activity in the MVST Sludge

		Percent of Total Beta Activity						
Tank 	pН	<sup>90</sup> Sr/ <sup>90</sup> Y (%)	<sup>134</sup> Cs+ <sup>137</sup> Cs (%)	<sup>66</sup> Co (%)	<sup>99</sup> Tc (%)	<sup>152,154,155</sup> Eu (%) .		
W-24 s	10.0	64.0	18.9	0.6	0.02	"15.6		
w-25 s	11.8	84.4	7.5	0.4	0.02	7.6		
W-26 S	9.4	63.3	17.9	0.8	0.03	17.9		
W-27 S	8.8	82.2	15.1	0.1	0.00	2.5		
W-27 H	11.7	71.1	26.8	0.3	0.02	1.6		
W-28 S	8.7	88.2	7.7	0.1	0.00	3.7		
w-29 s	12.0	75.0	18.1	0.7	0.02	5.7		
w-30 s	10.7	61.8	36.0	0.3	0.01	1.5		
w-31 s	9.3	95.2	4.1	0.1	0.01	0.6		
W-31 H	9.3	89.4	10.4	0.0	0.00	0.1		

Table 13 Summary of Actinide Elements in MVST Sludge

	W-24S	W-25S	W-26S	W-27S	W-27H	W-28S	W-29S	w-30s	w-31s	W-31H
Actinide	_(% α activity)									
<sup>232</sup> Th	0.02	0.04	0.03	0.04	0.04	0.03	0.03	0.04	0.06	0.07
$^{233}{ m U}$	4.34	3.55	12.14	2.20	1.67	2.94	3.64	3.56	2.32	0.04
<sup>234</sup> U	0.12	0.09	0.32	0.55	0.00	0.60	0.05	0.56	0.22	0.56
<sup>235</sup> U	0.01	0.00	0.01	0.02	0.00	0.03	0.01	0.02	0.01	0.05
$^{238}{ m U}$	0.28	0.20	0.56	0.69	0.16	0.90	0.47	0.60	0.33	1.34
<sup>237</sup> Np										
<sup>238</sup> Pu	8.08	11.95	12.36	9.73	10.53	12.92	10.43	11.40	10.65	13.40
<sup>239+240</sup> Pu	6.33	8.27	7.59	7.95	7.58	10.65	7.11	11.21	6.37	13.72
<sup>241</sup> Am <sup>a</sup>	10.10	7.90	7.44	10.70	16.85	10.82	11.85	10.11	8.67	10.21
<sup>244</sup> Cm	70.71	68.00	59.55	68.11	63.18	61.10	66.39	62.50	71.37	60.61
Gα	150000	110000	68000	62000	24000	58000	64000	55000	92000	33000

<sup>&</sup>lt;sup>a</sup> The "Am data is based on subtracting the <sup>238</sup>Pu by ICP-MS from the alpha peak measured at  $5.15\,MeV$  (<sup>238</sup>Pu + <sup>241</sup>Am) in the alpha spectrum.

The distribution of the alpha activity is summarized in Table 13, which includes the percent alpha for each MVST sludge sample. In general, the alpha activity in the MVST system is strongly weighted by the <sup>244</sup>Cm, which has a high specific activity. The list of actinides in Table 13 required several radiochemical and inorganic analytical measurements to generate the best estimates for each of the alpha activities. The <sup>232</sup>Th activity is calculated from the total thorium measured by ICP-AES. The other thorium isotopes (228Th, 229Th, and 230Th) are present in the ORNL sludge waste at such low mass, their presence would not effect the ICP-AES measurement. The uranium isotopes are measured by ICP-MS. The atom % results are converted to weight %, which is used to calculate the concentration of each uranium isotope from the total uranium results obtained by ICP-AES. The activity for each uranium radionuclide is then calculated **from** the specific activity for each isotope. The plutonium isotopes are first measured by ICP-MS, and the total plutonium alpha activity, measured after a chemical separation, is, used to calculate the activity for each isotope. The <sup>244</sup>Cm was measured directly by alpha spectrometry without any chemical separation. The <sup>241</sup>Am activity is determined by subtracting the  $^{238}$ Pu activity from the sum of the  $^{238}$ Pu +  $^{241}$ Am measured by alpha spectrometry. Both <sup>238</sup>Pu and <sup>241</sup>Am have an alpha energy of about 5.50 MeV and can not be resolved by alpha spectrometry. There was no chemical separation of the plutonium and americium for this project because of cost concerns.

#### 5.4 RCRA Characteristics for the MVST System

The RCRA regulatory limits are listed in Table 14, which also includes the limits for the EPA Toxicity Characteristic Leaching Protocol (TCLP) extract and the functional total metal limits for a solid or sludge waste. The total metal limits are a factor of twenty times higher than the TCLP extraction limits and are based on the 1:20 dilution used for the TCLP extraction procedure.

Table 14 Summary of RCRA Regulatory Limits

Metals		TCLP Extract and Liquids (mg/L)	Solid/Sludge Total Metal (mg/Kg)
Silver	(Ag)	5	100
Arsenic	(As)	5	100
Barium	(Ba)	100	2000
Cadmium	(Cd)	1	20
Chromium	(Cr)	5	100
Mercury	(Hg)	0.2	4
Nickel	(Ni)	50	1000
Lead	(Pb)	5	100
Selenium	(Se)	1	20
Thallium	(Tl)	0.9	18

If the RCRA metal concentrations are found to be below the total metal limits, the solid waste can not fail the TCLP leach test. If the RCRA metal concentrations exceed the total metal limits, the TCLP leach test must be done to determine if the solid waste is hazardous. For solid samples, the TCLP leach test is only valid for the final waste form ready for disposal. The total metal concentration data can be used as acceptable process knowledge if the final waste form only results in a dilution of the RCRA metal concentrations. Examples of waste forms that result in a dilution of a solid waste includes grouting (2 fold dilution) and vitrification (3 fold dilution). If the total metal limit is exceeded after stabilizing the waste, the TCLP leach test would be required for only the metals that had the potential to exceed the regulatory limits.

All of the MVST tank sludge samples exceed the total metal limits for lead and mercury, and two tanks are over or near the limit for chromium. Most of the ORNL radioactive waste sludge samples, characterized to date, have exceeded the total metal limits for these three RCRA metals. Based on past experience, it is expected that solidification of the ORNL MVST sludge would fix these RCRA metals such that the final waste form would pass the TCLP leach test.

### 5.5 TRU Classifications for LLLW System

The DOE definition for Transuranic (TRU) Waste includes the following conditions,

- TRU activity  $\geq 3700 \text{ Bq/g} (100 \text{ nCi/g})$ ,
- TRU isotopes must be alpha emitting actinide with Z > 92 (uranium),
- TRU isotopes must have a half life ≥ 20 years.

This definition excludes all thorium and uranium isotopes. The short lived actinide <sup>244</sup>Cm (t<sub>1/2</sub> = 18.1 years), which is common to ORNL waste, falls outside the TRU definition. Also, the plutonium isotope, <sup>241</sup>Pu, would be excluded from calculation of the TRU activity because it is a pure beta emitter. The primary actinide elements common to ORNL waste, that are present at sufficient levels to meet the TRU definition, include <sup>238</sup>Pu, <sup>239</sup>Pu, <sup>240</sup>Pu, and <sup>241</sup>Am. There is some current work at the Radiochemical Engineering Development Center (Mark-42 fuel assembly processing) that could generate enough <sup>243</sup>Am to make a significant contribution to TRU alpha content of the waste. The remaining actinide elements present in ORNL waste are generally not available at high enough activity, and/or do not have a long enough half-life to meet the TRU definition.

None of the MVST supematant samples which were analyzed in previous sampling campaigns had enough alpha activity to be considered as TRU waste. All of the MVST sludge that has been characterized to date has been classified as TRU waste based on only the plutonium and americium activity. The alpha activity reported is based on wet weight, if adjusted for dry weight the activity would almost double. The MVST sludge samples contained enough plutonium and americium activity to easily satisfy the WIPP waste acceptance criteria'\* for transuranic waste. Based on the TRU activity, any dilution of the sludge that would result from a solidification process such as grouting or vitrification would most likely not effect the TRU classification.

#### 5.6 Distribution of Fissile Material in LLLW System

As discussed in section 3.5, the ORNL LLLW waste acceptance criteria (WAC) requires the **fissile** isotopes of uranium and plutonium to be diluted with <sup>238</sup>U and <sup>232</sup>Th, respectively. A summary of the dilution ratios for fissile material in the sludge samples is provided in Table 15. All the dilution ratios for the MVST sludge samples exceed the required dilution factors for the fissile isotopes of

uranium and plutonium. All the dilution ratios listed in Table 15 are based onequations discussed in section 3.5 of this report.

Table '15 Summary of Denature Ratios for MVST Sludge

Tank	$^{238}$ U/ $^{235}$ U $f_{35}$ (> 110) $^{a}$	<sup>238</sup> U/ <sup>233</sup> U (> 200) <sup>b</sup>	<sup>232</sup> Th/( <sup>239</sup> Pu+ <sup>240</sup> Pu) (> 200)
W-24	206	1017	7 6 1
w - 25	203	906	1054
W-26	194	718	765
W-27 S	155	2730	1336
^ W-27-H	255	1680	1201
W-28	160	2900	851
w-29	185	1640	1014
w-30	182	2060	1098
w-31 s	148	1190	1908
W-31 H	155	284000	1460

<sup>&</sup>lt;sup>a</sup> Calculation based on equation 1 in Section 3.5. <sup>b</sup> Calculation based' on equation 2 in Section 3.5., <sup>c</sup> Conservative estimate of <sup>239</sup>Pu dilution since <sup>240</sup>Pu is included.

The dilution ratios listed in Tables 15 are based on the ratio of weight %, not the ratio of atom % given in the data tables. There is a small difference between atom %, reported for the uranium and plutonium, and weight %, which is needed for many calculations performed with the analytical data. To convert from atom % to weight %, we used the following equation,

$$W_i = \frac{a_i M_i}{\sum_{i=1}^{n} a_i M_i} X 100\%$$

where, 
$$W_i$$
 = weight %,  $M_i$  = nuclidic mass  $a_i$  = atom %.

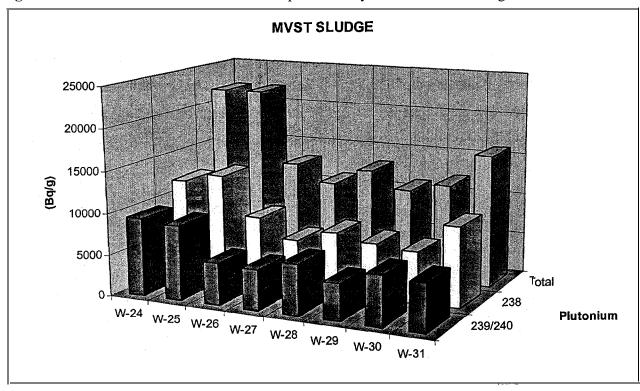
An example of this calculation is provided in Table 16, which shows there is not much difference between the atom % and the weight %.

Table 16 Example of Converting Atom % to Weight % for W-31 Sludge

Isotope	Nuclidic mass (g/mol)	atom %	$(\mathbf{a_i}  \mathbf{M_i})$	weight %
<sup>233</sup> U	233.039629	0.056	13.0502	0.0548
234U	234.040947	0.004	0.9362	0.0039
<sup>235</sup> U	235.043924	0.621	145.9623	0.6132
236 U	236.045563	0.002	0.4721	0.0020
$^{238}\mathrm{U}$	238.050785	99.316	23642.2518	99.3260
Total		99.999	23802.6726	99.9999

The distribution of plutonium isotopes by alpha activity are illustrated in Fig. 4 for each of the MVST samples. One should note that the <sup>238</sup>Pu dominates the alpha activity and the <sup>239</sup>Pu is the major isotope by weight or concentration.

Figure 4 Distribution of Plutonium Alpha Activity in the MVST Sludge

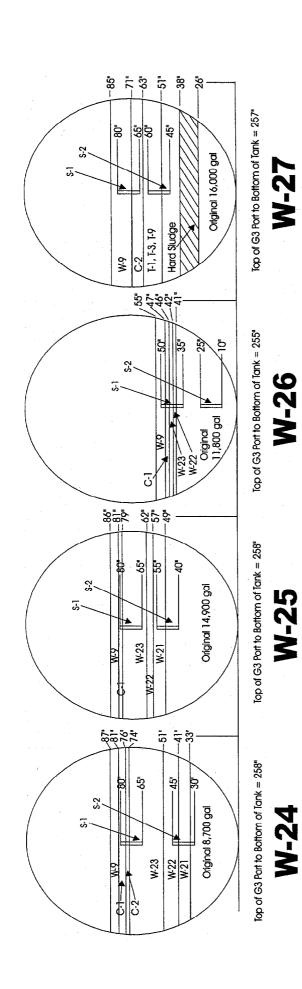


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## APPENDIX A

The following diagram graphically illustrates where sludge samples were collected from each MVST tank during the 2000 sampling campaign. Several of the other ORNL waste tanks systems were transferred to the MVST system prior to the sludge sample being collected. This diagram also provides an estimate of which MVST tanks collected these transfers. There were only minor changes in the contents of these tanks prior to this sampling campaign. The sludge samples for this campaign consisted of a full core from the top to the bottom of the sludge layers.



W-26

-73\* 55 Hard Sludge 6₹ Original 11,020 gal Original 12,070 gal -95 ---44" ---35" -21 Original 4,500 gal 1-1, 1-2, 1-4 6-∕\

Top of G3 Port to Bottom of Tank = 256"

W-28

W-29

0**%-**%

W-31

Top of G3 Port to Bottom of Tank = 248"

# APPENDIX B

# Radioactive Materials Analytical Laboratory QC Acceptance Criteria for Radioactive Liquid/Solid Waste Samples

Analysis	Method (s) CASD-AM-	Quality Control Check	SW-846 Acceptance	RMAL Acceptance
		(per batch)	Criteria	Criteria
		<b>(</b> • • • • • • • • • • • • • • • • • • •	(%D, %R, RPD)	(%D, %R, RPD)
Metals by ICP-AES (inductively coupled plasma atomic emission spectroscopy)	SW846-6010A	high standard calibration verifications (ICV & CCV) <sup>a</sup> calibration blank & checks (ICB & CCB) <sup>b</sup> method blank (sample prep) matrix spike matrix spike duplicate or sample duplicate laboratory control sample (sample prep) serial dilution (if interference suspected) post digestion spike <sup>d</sup>	±5%D ±10%D <3 x IDL <3 x IDL ±20%D ±20 RPD none specified ±10%R ±20%D	±5%D ±10%D <3 x IDL <3 x IDL ±25%D (liq.), ±30%D (solid) ±20 RPD (liq.), ±30 RPD (solid) ±20%D ±10%R ±25%D (liq.), ±30%D (solid)
Metals by ICP-MS (inductively coupled plasma- mass spectrometry)	SW846-6020	calibration verifications (ICV & CCV) calibration blank & blank checks (CCB) <sup>b</sup> method blank (sample prep) <sup>c</sup> matrix spike matrix spike duplicate or sample duplicate laboratory control sample (sample prep) <sup>c</sup> internal standard post digestion spike <sup>d</sup>	±10%D <3 x IDL none specified none specified ±20 RPD none specified 30-120% R ±10%D	±10%D <3 x IDL <10 x IDL  ±25%D (liq.), ±30%D (solid)  ±20 RPD (liq.), ±30 RPD (solid)  ±20%D  ±30%D  ±20%D
Metals by GFAA (graphite furnace atomic absorption)	SW846-7000A	high standard calibration verifications (ICV & CCV) method blank (sample prep)* matrix spike matrix spike duplicate laboratory control sample (sample prep)* serial dilution (if interference suspected) post digestion spike4	not required  ±10%D (ICV), ±20%D (CCV)  none specified  none specified  none specified  none specified  ±10%R  ±15%D	±5%D ±10%D (ICV), ±20%D (CCV) <3 x IDL ±25%D (liq.), ±30%D (solid) ±20 RPD (liq.), ±30 RPD (solid) ±25%D ±10%R ±25%D (liq.), ±30%D (solid)
Mercury by CVAA (cold vapor atomic absorption)	SW846-7471A SW846-7470	instrument blank calibration verification (ICV & CCV)* method blank (sample prep)° laboratory control sample (sample prep)° matrix spike matrix spike duplicate or sample duplicate post digestion spike*	none specified	<pre>&lt;5 x IDL</pre>
Carbon (total organic carbon, total carbon, total inorganic carbon)	SW846-9060	instrument blank calibration verification (ICV & CCV) matrix spike matrix spike duplicate	none specified none specified none specified none specified	<3 x IDL ±10%D (ICV.), ±20%D (CCV) ±25%D (liq.), ±30%D (solid) ±20 RPD (liq.), ±30 RPD (solid)
Anions by Ion Chromatography (IC)	SW846-9056	calibration verification (ICV & CCV) matrix spike sample duplicate	±10%D (ICV), ±5%D (CCV) none specified none specified	±10%D (ICV), ±15%D (CCV) ±25%D ±20 RPD
pH measurement	SW846-9040A SW846-9045B	check standard sample duplicate	none specified none specified	±10%D ±20%D

Analysis	Method (s) CASD-AM-	Quality Control Check (per batch)	SW-846 Acceptance Criteria (%D, %R, RPD)	RMAL Acceptance Criteria (%D, %R, RPD)
Fotal and dissolved solids (TS & TDS)	EPA600-160.2 EPA600-160.3	sample duplicate check standard	none specified none specified	±10 mg/ 10mL sample ±10%D
Carbonate and bicarbonate titration	AC-MM-1 003105	sample duplicate check standard	none specified none specified	±20 RPD ±20%D
Gross alpha/beta	EPA-900.0 RML-RA02 RML-RA12	background check calibration verification method blank (optional) <sup>f</sup> sample duplicate matrix spike	none specified none specified none specified none specified none specified	<3sigma daily change ±10%D evaluated for contamination ±25 RPD (liq.), ±30 RPD (solid) ±25%D (liq.) & ±30%D (solid)
Nuclides by gamma spectrometry	EPA-901.1	background check calibration verification sample duplicate	none specified none specified none specified	< 3 sigma daily change ± 10%D ±25%D (liq.) & ±30%D (solid)
3r-90 determination	RML-RA13 EPA-905.0	method blank (optional) <sup>f</sup> laboratory control sample matrix spike matrix spike duplicate or sample duplicate associated instrument QC	none specified none specified none specified none specified none specified	evaluated for contamination <sup>6</sup> 20%D ±25%D(liq.)&● 30%D(solid) ±25 RPD (liq.), ±30 RPD (solid) see gross alpha/beta criteria
Γc-99 letermination	DOE Compendium RP550 RML-RA05	method blank (optional) <sup>f</sup> laboratory control sample matrix spike matrix spike or sample duplicate associated instrument QC	none specified none specified none specified none specified none specified	<3 x IDL 20%D ±25%D (liq.) & ±30%D (solid) • 25RPD(liq.),±30RPD(solid) see ICP-MS criteria
I-3 determination	EPA-906.0	method blank (optional) <sup>f</sup> laboratory control sample matrix spike matrix spike duplicate or sample duplicate associated instrument QC	none specified none specified none specified none specified none specified	evaluated for contamination <sup>a</sup> 20%D ±25%D (liq.) & ±30%D (solid) ±25 RPD (liq.), ±30 RPD (solid) see gross alpha/beta criteria
Th Determination	EPA-901.1 RML-RA09	method blank (optional) <sup>f</sup> laboratory control sample matrix spike matrix spike duplicate or sample duplicate associated instrument QC	none specified none specified none specified none specified none specified	evaluated for contamination <sup>6</sup> 20%D ±25%D (liq.) & ±30%D (solid) • 25RPD(liq.),±30RPD(solid) see gamma spectrometry criteria
PCBs (polychlorinated- biphenyls)	SW846-8080	calibration verification (ICV & CCV) method blank (sample prep) surrogate standard matrix spike matrix spike duplicate sample duplicate laboratory control sample (sample prep)	refer to method 8080 none specified	to be specified <sup>h</sup> < regulatory limit (2ppm) ± 50-150%R ± 50-150%R ± 50-150%R to be specified <sup>h</sup> to be specified <sup>h</sup>

- a Initial calibration verification (ICV) is typically performed at the beginning of a run to check the calibration and must be independent of the calibration standards. The continuing calibration verification (CCV) must also be independent of the calibration standards, but may be the same standard as the ICV. The CCV is typically analyzed every 10 samples and at the end of the run for metals analysis or every 12 samples for organic analysis.
- b The calibration blank is an instrument blank used in the calibration to initially determine the blank value and therefore used as blank subtraction. The continuing calibration blank (CCB) is also an instrument blank which is analyzed every 10 samples and at the end of the run, but is not used in blank subtraction, but only to monitor instrument contamination.

- Method blanks and laboratory control samples are only required if a sample preparation is performed before analysis. Sample preparation does not include dilutions or transfers to containers.
- d Post digestion spikes are not necessary if the pre-digestion spike is in control. If this control does not meet the QC acceptance criteria, the post digestion spike should be performed.
- e Acceptance criteria:

%D = % deviation from true value

%R = % recovery of true value

RPD = relative percent difference between two compared values

- f Method blanks for radiochemical analysis are used to monitor cross contamination. However, due to the levels of radioactivity present in samples at the **RMAL**, the effect of contamination may be insignificant in most cases. Therefore, the requirement to analyze a method blank for radiochemical analysis is **optional** (i.e. at the discretion of the chemist or supervisor).
- g Acceptance criteria for the method blanks performed for radiochemical analysis varies based upon the level of activity in the samples and the amount of background activity. A qualified chemist reviews the data from method blanks to determine if significant contamination is present.
- h The acceptance criteria for PCB analyses which are not identified in this table, shall be specified at a later date. Currently, the Analytical Methods Group group leader specifies the QC criteria if different **from** SW846 and if not specified by the sample generator.

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