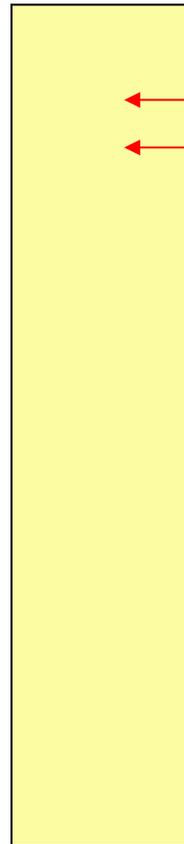


Post-Combustion CO₂ Capture: Beyond 2020, a BES Perspective

- **Post-combustion Capture: Background and Challenges**
- **BES Current Portfolio relevant to CO₂ Capture**
- **Going Forward**

Post-Combustion Capture

Scrubbing



10 – 15 % CO₂
Predominately N₂
1 atm.
H₂O
Trace or more SO_x, NO_x

Flue gas

5% of efficiency loss

Materials Challenges

Absorb/adsorb fast
High selectivity vs. N₂
Desorb fast
Chemically stable
Physical properties stable
(Viscosity, mechanical integrity)
Non-corrosive
CHEAP

Science

Build knowledge base that will
provide guidance for rationale
design of next generation sorbents

Stripping



CO₂ + Steam

33%
loss

CO₂

H₂O

ΔT (Steam)

60% of efficiency loss

Key Characteristics for CO₂ Capture (Post-Combustion)

- Perform in presence of water (steam) and trace contaminants including SO_x and NO_x, etc.
- Temperature stability over a wide temperature range
- If Solid sorbent – mechanical stability, If liquid – no viscosity change
- Sustainable
- Cheap (scale)
- High selectivity for CO₂ over N₂
- Work in dilute streams (10% CO₂)
- Rapid kinetics for capture (scrubbing) and regeneration (stripping)
- Low energy regeneration (stripping)
- Non-corrosive
- High cycle life

Challenge - Scale



- 30 Giga tons of Anthropogenic CO₂ in 2004
- Take 13.5 million Houston Astrodomes to hold that at atmospheric pressure*
- CO₂ from US electricity 2006 – only 1.05 million Houston Astrodomes

*From: George C. Marshall Institute Policy Outlook, June 2008.

Challenge - Science

Conundrum

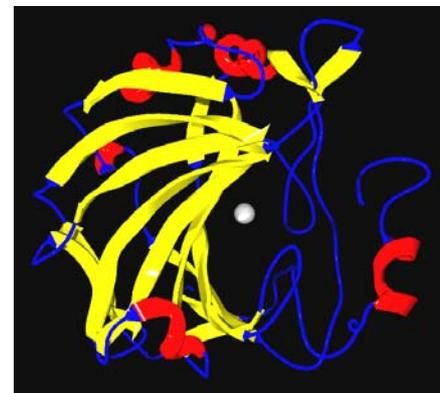
- To remove CO₂ from flue gas selectively and rapidly → strong bond
- To strip the CO₂ from scrubber with little energy need → weak bond



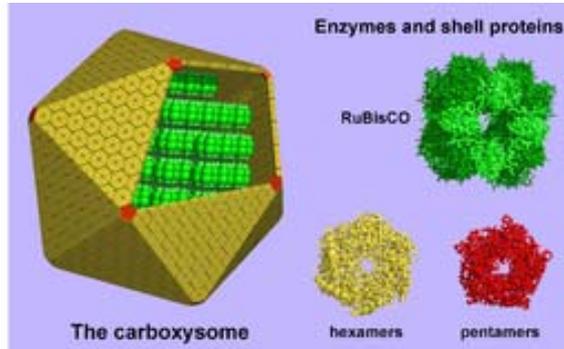
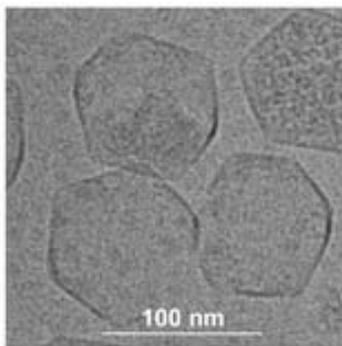
- Analogy – Superglue
- Challenge – Find the acetone – lower energy trigger



- Nature and CO₂
- Plants
 - Deal with dilute CO₂ (not efficiently)
 - Carbonic Anhydrase - in plants – increases CO₂ concentration
 - RuBisCO – plant enzyme converts CO₂ to sugar – 3 CO₂/sec
- Bacteria
 - Carboxysome - concentrates CO₂ levels to increase chemical transformation
 - Strategy: Concentrate, isolate and transform



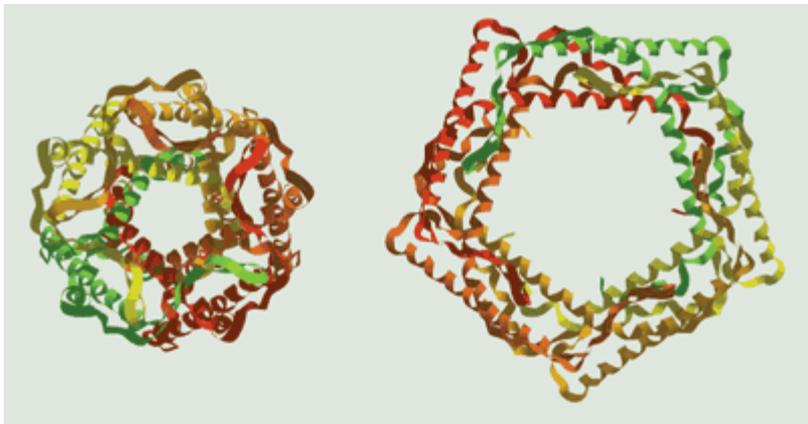
Ribbon diagram of human carbonic anhydrase with Zn atom in center



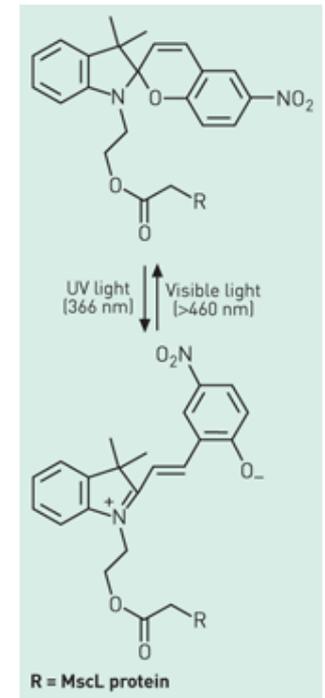
Carboxysomes – protein bacterial organelles on left is electron microscope image, on right is model of structure

Nature's Triggers

- While nature does not deal with CO₂ ideally for post-combustion capture – can provide hints for non-thermal triggers
 - Ecoli have a channel protein that opens and shuts with pressure –
 - Has been modified to open and shut with light¹



Channel proteins modified with spiropyran opens or closes channel with light.



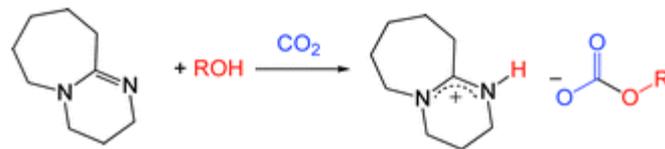
Reversible spiropyran photo-switch.

- Synapses electrical field triggers release of neurotransmitter

¹Source – *Science*, **2005**, 309,755, Copied from C&E News report
<http://pubs.acs.org/cen/news/83/i31/8331notw1.html>

Alternate science approach may offer a better pathway in the future Heldebrant, PNNL

- Eliminate water and H bonding – reduces energy of stripping
 - Lower specific heat of organic liquids
 - Same strip temperature, considerable energy savings
- Organic amidine or biological guanidine base plus alcohol
 - Binds CO₂ – reversed with heat
 - Higher binding capacity for CO₂ than commercial MEA (monoethanolamine)
 - Design base to minimize hydrogen bonding – more energy savings



- Demonstrated uptake in 50% CO₂/ 50% N₂, strip 90 to 130 C
- Issue - water form bicarbonate salt of base – strips MEA temperature

- CO₂ sorbent that can be stripped with minimal energy
 - Look to Nature for ideas on triggers but unlikely to be a biological solution
 - Electrical, electrochemical, pressure, light, pH, phase change (liquid crystal)*
 - Nanoscale approaches

- Build the knowledge base to facilitate a rationale design of sorbents from first principles
 - Advanced computational modeling and theory - First-principles methods for capture and release of nonpolar molecules

 - Advanced characterization tools, including those supported by BES, to
 - Identify structure of binding sites
 - Understand kinetics and thermodynamics of sorption/desorption in realistic conditions

 - New chemistries for advanced sorbents
 - New sorbents
 - Synthesis

- **Directly related (EFRC funding) – CO₂ from N₂**
 - Zeolitic imidazolate framework (ZIFs) , metal organic frameworks (MOFs), membranes for separation
 - Synthesis, characterization, computation/modeling

- **CO₂ Binding/Separation**
 - Often from CH₄
 - MOFs, ZIFs
 - Polymer membranes
 - Computation, theory, modeling

- **Hydrogen – Binding and Separation**
 - H₂ sorbents
 - Inorganic clathrates, MOFs
 - Membranes

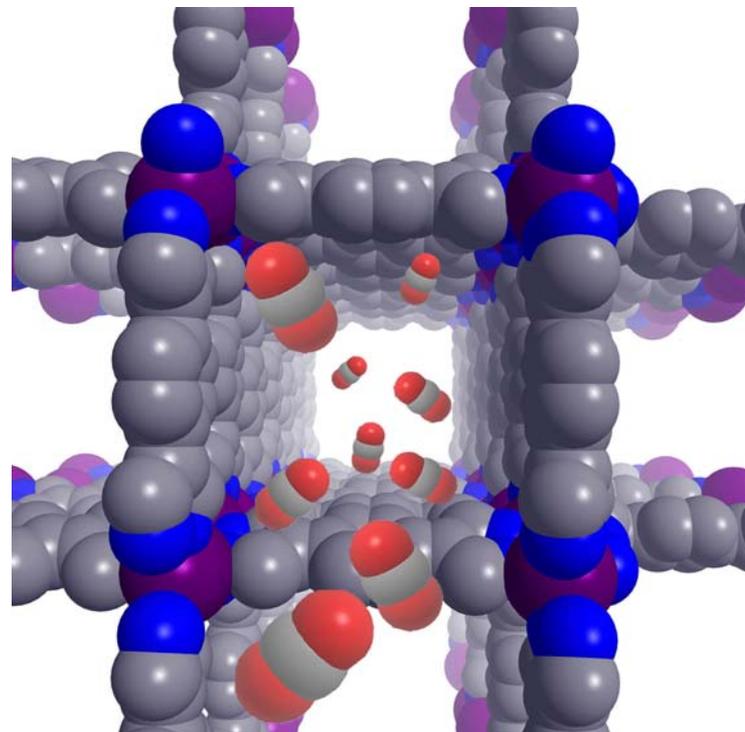
- **Separation Sciences, Materials, Theory & Computation**
 - Membrane synthesis
 - Transport
 - Theory and Computation

EFRC: Gas Separations for Clean Energy Technologies: (Smit, Berkeley)

Synthesis: Generate high surface area MOFs & self-assembled synthetic biomimetic polymer films

Characterization: Atomic-level structural characterization - before and after exposure to gas, accurate means of assessing the selectivity, kinetics, and thermodynamics of gas adsorbate binding – use to test computational models

Computational Separations: Strong computational component - understand chemical interactions at a molecular level, guide synthetic efforts



Goal: New strategies and materials for *energy efficient selective capture or separation of CO₂* from gas mixtures based on molecule-specific chemical interactions

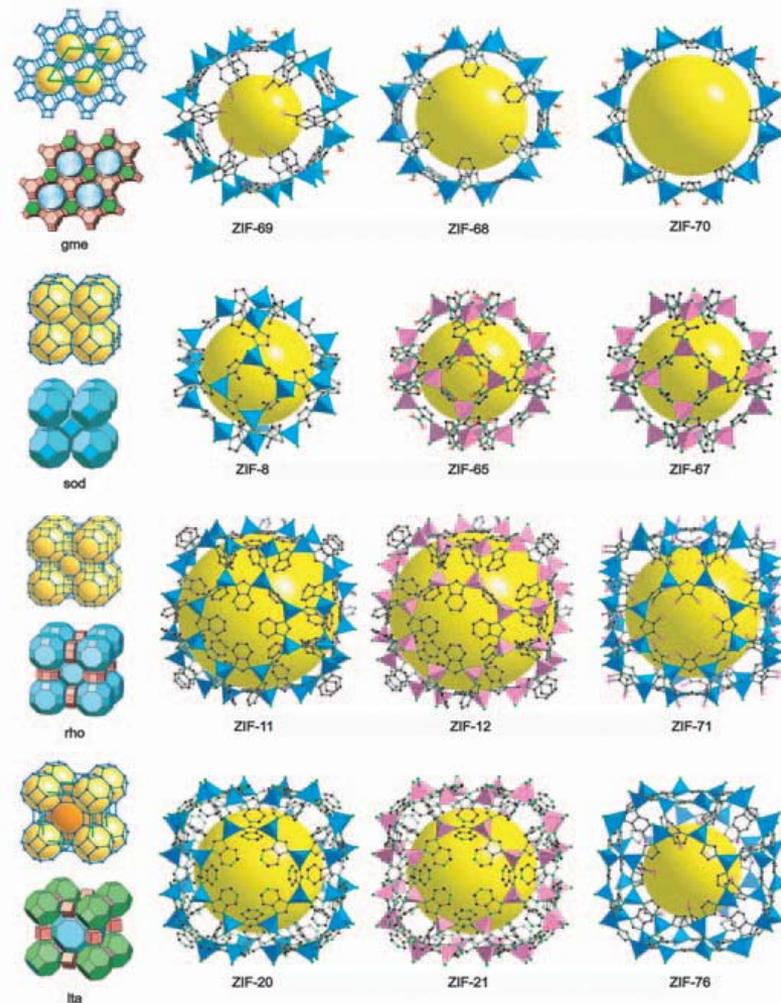
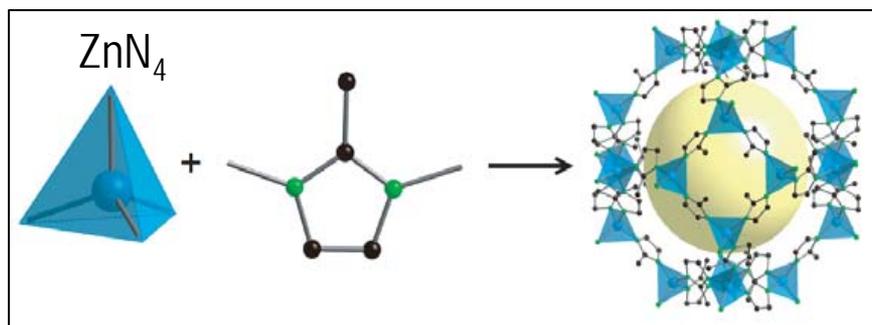
EFRC: Molecularly Engineered Energy Materials (Ozolins, UCLA)

Capture is 1/3 of this EFRC: Novel ZIFs for gas separation including CO₂ capture

Yaghi – Synthesis, characterization

Asta – Electronic structure calculations

Laird & Houndonougbo – Monte Carlo Simulations



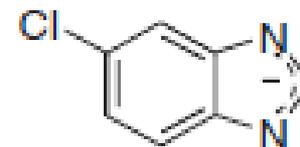
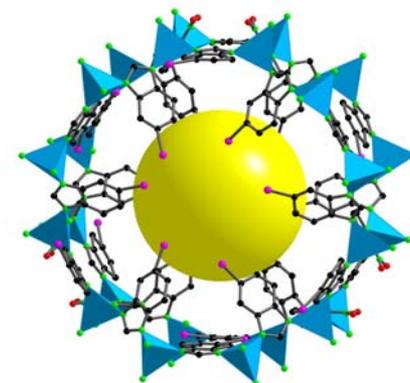
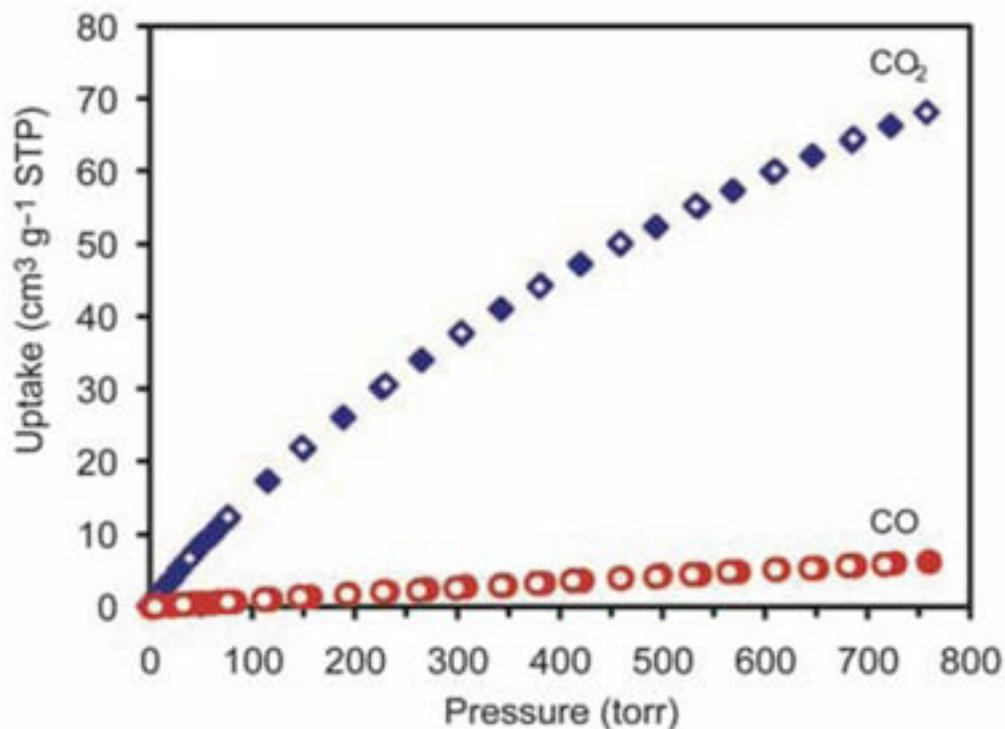


CO₂ Uptake in ZIF-69

Separate CO₂ from CO

Surface areas to 1970 m²/g

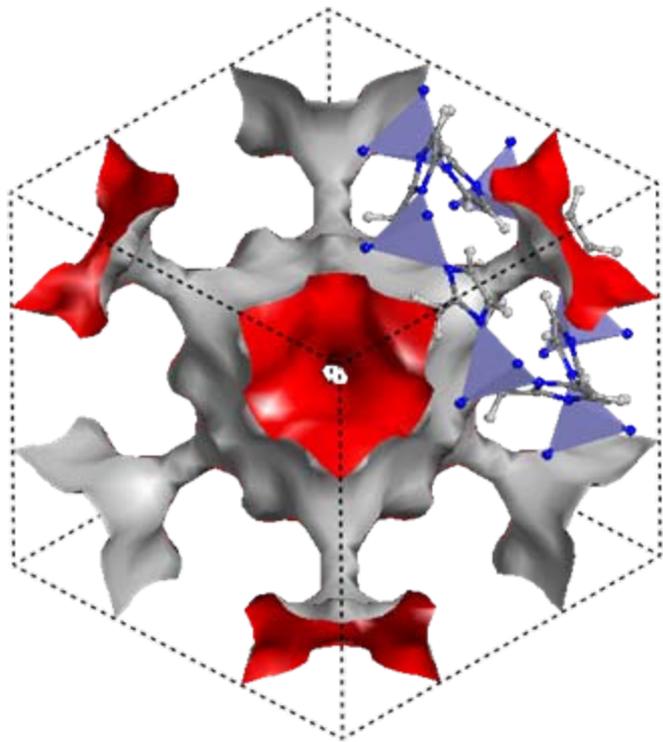
Stable to 390 °C, 1 liter ZIF-69 holds ~83 liters of CO₂ at 273 K



cbim

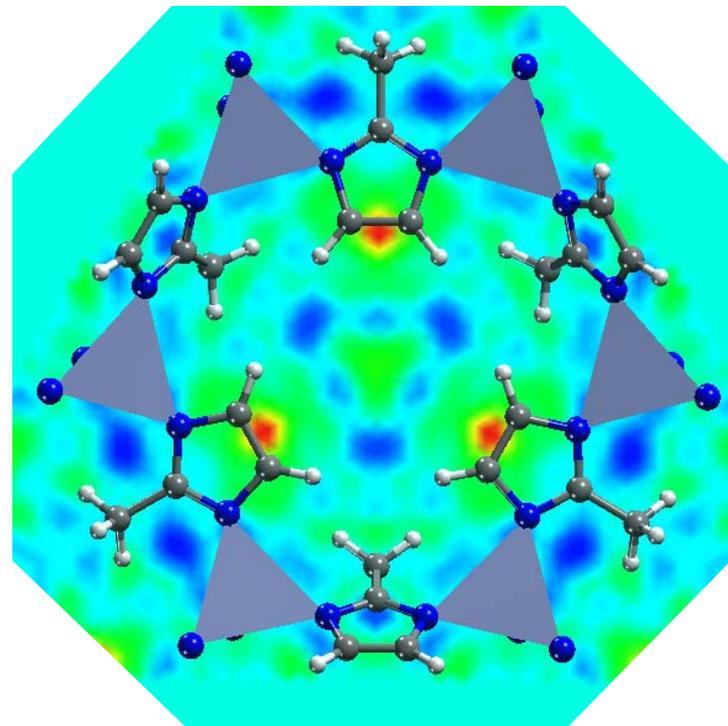
Gas adsorption isotherms of CO (red) and CO₂ (blue) into ZIF-69 at 273 K. Uptake and release are represented by solid and open symbols, respectively. ZIF-69 has been shown to have substantially greater uptake capacity for CO₂ over CO.

Where Does Hydrogen Go in ZIFs? Apply Technique to CO₂



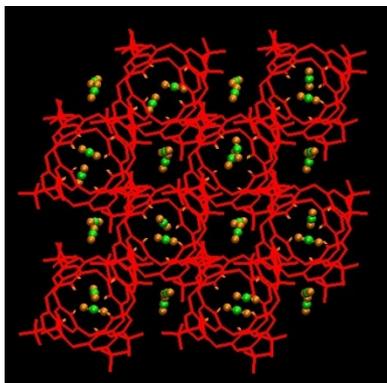
Calculated isosurface in ZIF-8

- Available nanopore volumes are connected by narrow channels



Difference Fourier analysis, showing the main-H₂ absorption sites in ZIF-8 are near the top of C-C bond of the linker and NOT the metal center

Examples of BES-Supported Projects in Theoretical and Computational Modeling



PNNL: Dang et al

One interpretation: Computer simulations show that at ambient temperatures, CO₂ molecules require ~ 4 times more energy than average to desorb. *Novel Strategies for selective heating may be necessary.*

Computational studies of load dependent guest dynamics and free energies of inclusion for CO₂ in low density *p-tert*-butylcalix[4]arene at loadings up to 2:1.

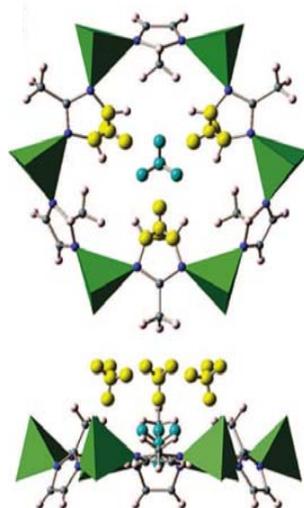
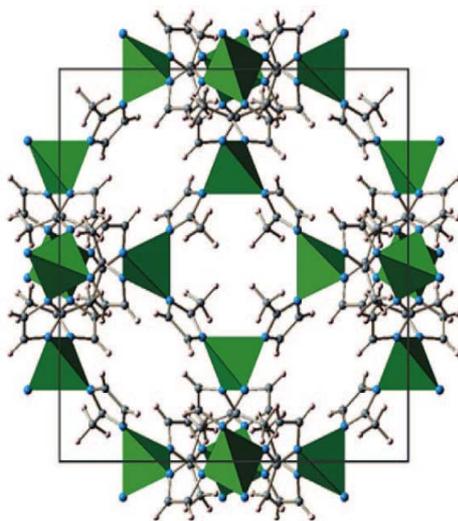
John L. Daschbach, Xiuquan Sun, Tsun-Mei Chang, Praveen. K. Thallapally, B. Peter McGrail, and Liem X. Dang. *J. Phys. Chem. A, Vol. 113, No. 14, 2009*

Theoretical and Computational Chemistry (Start FY09): Modeling CO₂ capture and separation in zeolitic imidazolate frameworks (Wisconsin, Schmidt)

- Molecular Level Mechanism of CO₂ adsorption?
- Specificity of CO₂ over N₂ ?
- Mechanism for CO₂/N₂ Selectivity in ZIFs ?
- Thermal and Solvent Stability?

Neutron Diffraction

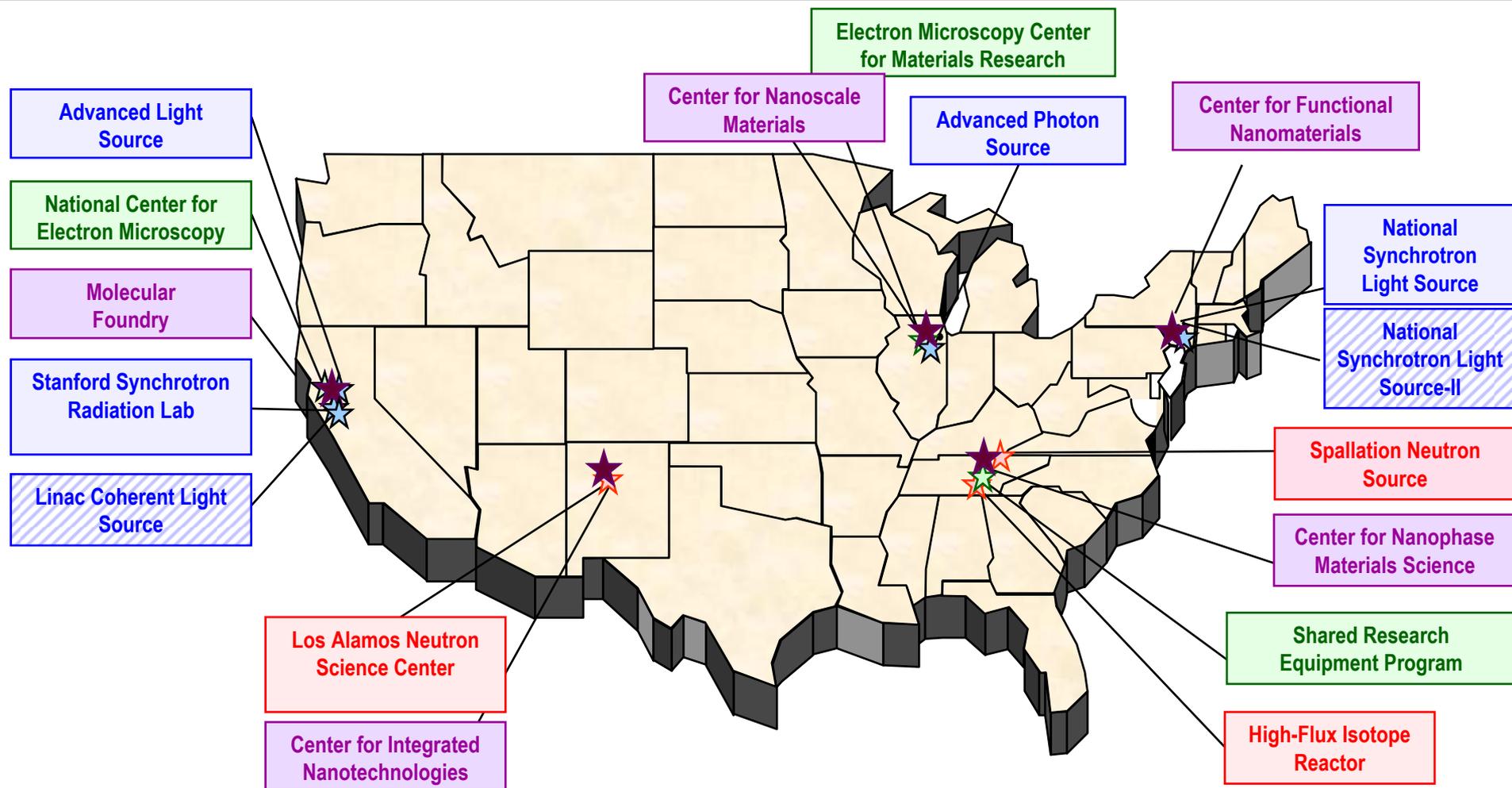
- Studied CD₄ absorption in a ZIF with a deuterated methyl imidazolate linker.



- Structure of ZIF
 - Changes with sorption
 - Changes with temperature
- Identify binding sites
 - Primary and stronger binding site - imidazolate
 - Secondary and weaker site - center

Note: This can be used with CO₂ since the organic linkers in the ZIF can be deuterated

BES Scientific User Facilities

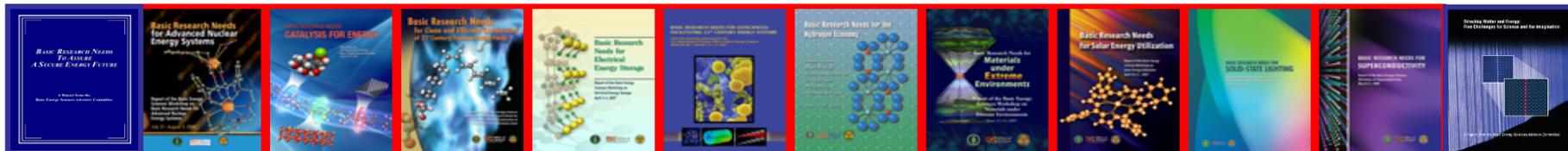


- 4 Synchrotron Radiation Light Sources
- Linac Coherent Light Source (Under construction)
- 3 Neutron Sources
- 3 Electron Beam Microcharacterization Centers
- 5 Nanoscale Science Research Centers

“Basic Research Needs” Style Workshop – Carbon Capture Beyond 2020

- Timing – after Carbon Capture 2020 Workshop
- Build on output of Carbon Capture 2020 Workshop
 - Start - define the current state for both research and industrial needs (factual document)
 - Goal - Identify priority research directions for the challenges beyond 2020
- Educate basic researchers on the technical and scientific challenges of CO₂ capture
- Approach (Basic Research Needs style and format)
 - Integration with DOE BES, DOE FE, Academia, National Labs, and Industry
- Draft Charge:

“Identify basic research needs and opportunities underlying post-combustion carbon capture technology, with a focus on new or emerging science challenges with potential for significant, long-term impact on the effective separation and capture of CO₂ in post-combustion gas streams.”



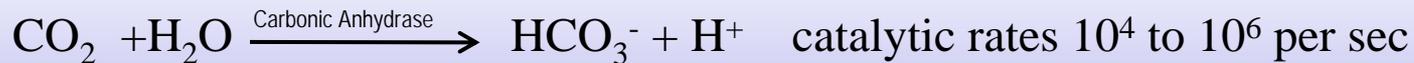
- Workshop format:
 - Plenary talks on technical and scientific challenges
 - Breakout panels focused on development of priority research directions
 - Crosscutting panel focused on identification of grand challenge science themes
 - Dedicated report writers to ensure rapid progress on written report
- Finite focus means moderate size: 50-75 participants, including agency reps
- Venue in the DC area to facilitate attendance by DOE staff
- 6 months timeline from initial planning to producing a final workshop report
 - Planning could begin in parallel with Carbon Capture 2020
 - Carbon Capture - Beyond 2020 workshop would be held after the report from Carbon Capture 2020 is available



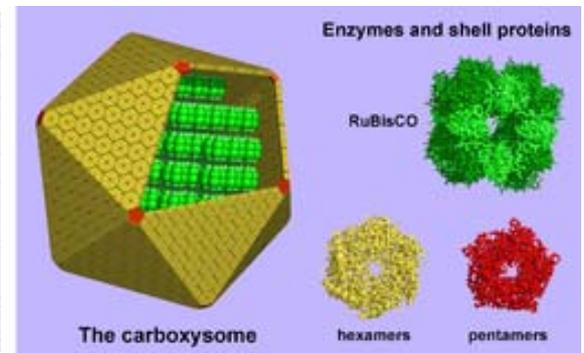
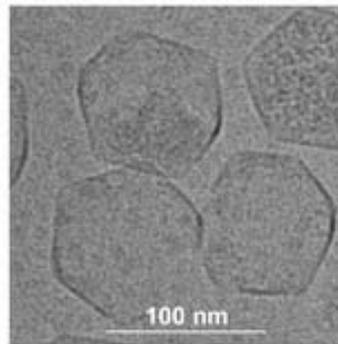
- Thank you
- Questions?

■ Nature and CO₂

- Also deals with dilute CO₂ (not efficiently)
- Carbonic Anhydrase, in animals – interconvert CO₂ and bicarbonate, in plants – increases CO₂



- RuBisCO – plant enzyme converts CO₂ to sugar– 3 CO₂/sec

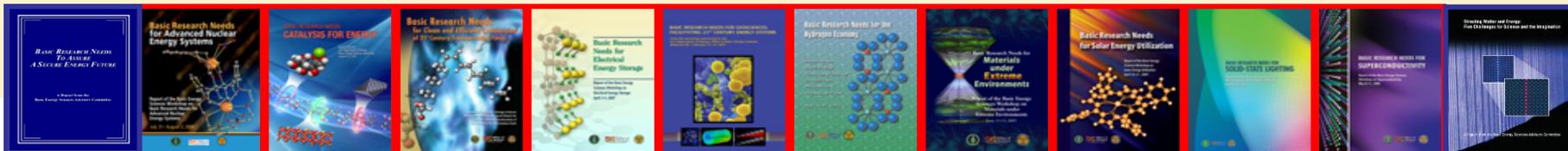
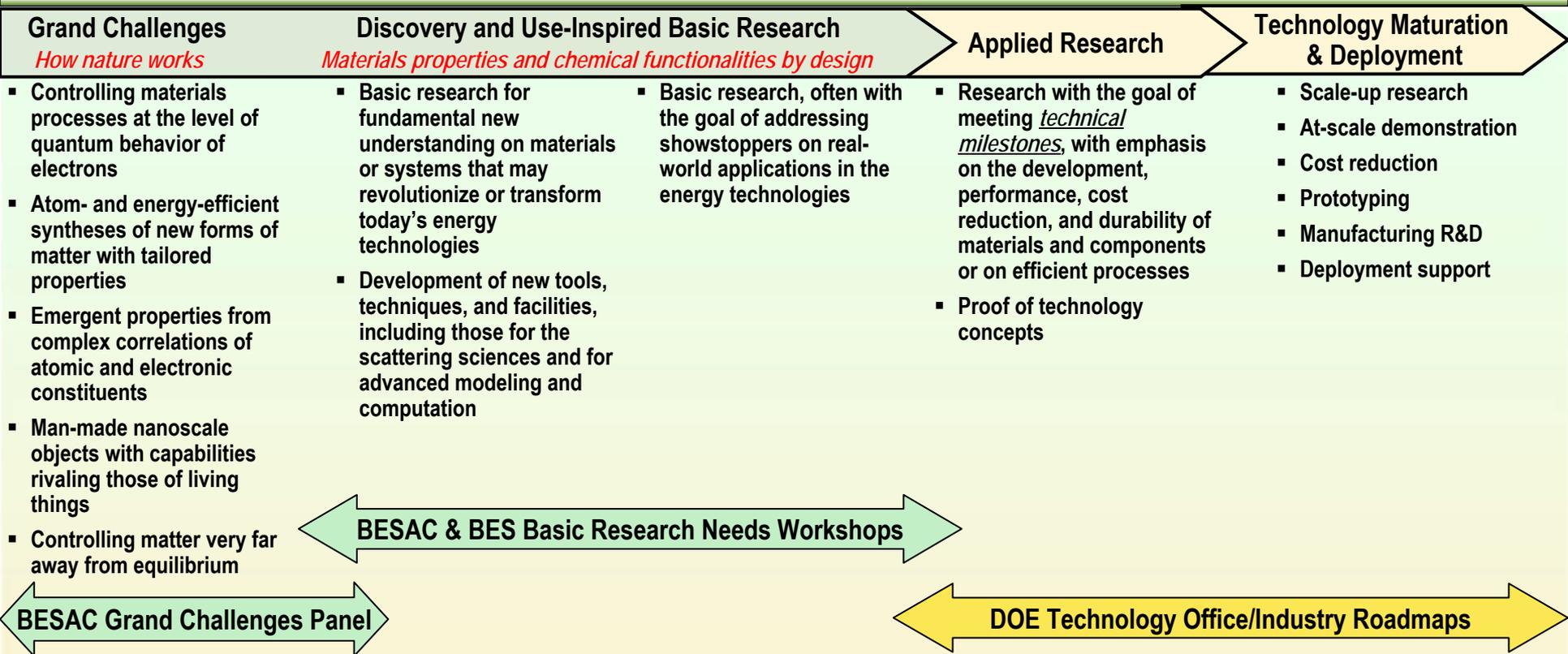


Nature - concentrators in bacteria to increase CO₂ levels, strategy concentrate, isolate and transform

Source: http://en.wikipedia.org/wiki/File:Carbonic_anhydrase.png and <http://en.wikipedia.org/wiki/Carboxysome>

Basic and Applied R&D Coordination

How Nature Works ... to ... Design and Control ... to ... Technologies for the 21st Century



- Newer Materials
 - ZIFs – zeolite imidazole frameworks
 - Imidazoles increase selectivity
 - Non silicates – resistant to degradation from steam
 - Membranes – functionalized with amines or other groups to enhance selectivity (>200), increase cycle life
 - Chemical functionalization of polymer can increase selectivity beyond what is achieved with size separation
 - Fibrous membranes coated with amines (glass or carbon fiber mesh)
 - Poly (ionic liquids)
 - Some have higher sorption than ionic liquids
 - Avoid viscosity increases seen with some ionic liquids
 - Ion Engineering – amines in ionic liquids (65% cost savings over water)

■ Amines

- Monoethanolamine (MEA) or amine blends - MEA with methyldiethanolamine (MEDA) or sterically hindered amines
- NH_3
- Can be aqueous solution in post-combustion

■ Chilled ammonia (35 F)

- ammonium carbonate to ammonium bicarbonate

■ Advantages - amines

- Selective, reversible, nonvolatile and somewhat inexpensive

■ Disadvantages - amines

- Corrosive, degrade esp. with O_2 or SO_2 present, require large thermal energy for stripping
- NH_3 – less energy to strip but toxic vapor to deal with

- Total estimated cost to consumer
 - Capture and compression (typically 100 atm or more) adds 35 – 40% to price of generating electrical power (Monoethanolamine (MEA) Based)
 - Computer calculations – with piperazine + MEA only 20% increase
- Parasitic energy load
 - Capture + compression drops plants thermal efficiency from 38% to 29%
 - 60% - Steam taken from turbines for stripper reboiler - fundamental research focus
 - 33% - Electric to drive CO₂ Compressor
 - 5% - Electricity to drive flue gas through scrubber

Source: *Advanced Post-Combustion CO₂ Capture*, prepared for Clean Air Task Force, Howard Herzog, Jerry Meldon and Alan Hatton

Today's Carbon Capture Approaches – A Brief Review

- Pre-Combustion Capture
 - Applicable to plants where coal gasification is first processing step
 - Few existing plants – will apply mostly to new plants
 - CO₂ capture should be easier

- Oxy-Combustion
 - Generally involves cryogenic oxygen separation plant
 - Novel concepts involving oxygen transport membranes and chemical looping – lab/pilot scale development

- Post-Combustion Capture
 - Existing coal fired power plants
 - Commercial amine-based processes available from Fluor, Mitsubishi Heavy Industries, & others
 - Parasitic power requirements and cost -major challenges
 - Pilot and demonstration projects - “advanced amines,” chilled ammonia, ionic liquid solvents
 - New approaches, “Breakthrough Concepts,” - metal organic frameworks (MOFs), zeolite imidazole frameworks (ZIFs), membranes, and other sorbents - smaller scales